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## TURBINE ENGINE LUBRICANT RECLAMATION

Richard J. Bruns and George L. Beemsterboer MONSANTO RESEARCH CORPORATION Dayton Laboratory Dayton, Ohio 45407

June 1983 Final Report for Period 1 September 1979 - 1 March 1983

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A distillation and adsorption treatment for reclaiming used MIL-L-7808 turbine oils was investigated. A viable additive package was tested on different MIL-L-7808 type virgin base stocks. Fifteen used oils were analyzed by acid number, high performance liquid chromatography, and gas chromatography. A distillation process utilizing caustic (sodium hydroxide) pretreatment was developed on 500-mL and 13-litre scales. Adsorption treatment of distilled oils consisted of barium hydroxide monohydrate. A total of ten 15-25 gallon batches of oil were

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reclaimed/reformulated and MIL-L-7808H tested. Test results were generally good. However, none of the 10 batches completely pas :d all of the tests.
An extensive screening procedure was found necessary to identify contaminated used oil samples.
An engineering cost study is included.
•

### FOREWORD

This final technical report was prepared by Monsanto Research Corporation (MRC), 1515 Nicholas Road, Dayton, Ohio 45407. The effort was sponsored by the Aero Propulsion Laboratory (APL), Air Force Wright Aeronautical Laboratories (AFWAL), Air Force Systems Command, Wright-Patterson AFB, Ohio, under Contract No. F33615-79-C-2052 during the period 1 September 1979 to 1 March 1983. The work herein was accomplished under Project 3048, Task 304806, Work Unit No. 30480611, "Turbine Engine Lubricant Reclamation," with Mr. G. A. Beane IV, AFWAL/POSL, as Project Engineer.

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### SUMMARY

The objective of this program is to improve upon the purification process and additive replenishment package, developed in an earlier program (Technical Report AFAPL-TR-78-50), for reclaiming used synthetic turbine engine oil meeting MIL-L-7808G specifications. The program is also to demonstrate the effectiveness of the total process by successfully reclaiming 12 batches of used synthetic oil representing a variety of MIL-L-7808 compositions.

Base stocks formulated by five different manufacturers were selected for MIL-L-7808H testing using MRC's proposed additive package.

Conditions were established for optimizing a sodium hydroxide distillation followed by barium hydroxide monohydrate treatment. The use of clay and activated charcoal was found to provide no distinct performance advantages over nonuse.

Ten-25 gallons large scale batches were reclaimed and reformulated. These oils were evaluated with MIL-L-7808H tests with good results. However, none of the ten batches completely passed all tests.

An extensive used oil screening method was found necessary to identify samples contaminated beyond reclaimability.

A cost analysis of the process was carried out.

#### INTRODUCTION

The re-refining and reuse of petroleum base oils has been a successful standard practice in railroad, automotive, and aircraft industries for many years. More recently, the reclamation of synthetic aviation oils has proven successful in those applications involving a specific and known formulation. However, this is not representative of normal U.S. military aircraft operational experience where numerous brands of specification products consisting of widely differing base stocks and additive packages are normally mixed in service. An estimated total of up to one half million quarts of used ester based oils are generated each year and could be collected for reclamation. Reclamation of these oils for reuse in military aircraft turbine engines represents a significant potential source of supply in the event of serious availability problems. Furthermore, reclamation could offer a significant cost savings if the technique developed is also economical and technically effective. Currently available information suggests that the ester base stocks are not significantly degraded during use. Therefore, there is potential for recovery of large portions of the used oils.

An earlier program, described in AFAPL-TR-78-50, demonstrated the potential feasibility of recycling used MIL-L-7808G oils to a satisfactory performance level.

The objective of this program is to improve upon the purification process and additive replenishment package described in AFAPL-TR-78-50, and to demonstrate the effectiveness of the total process by successfully reclaiming 12 batches of used synthetic oil representing a wide variety of MIL-L-7808 compositions.

### 2. SUMMARY OF THE FIRST OIL RECLAMATION PROGRAM

A process for reclaiming of a used synthetic turbine engine oil meeting MIL-L-7808G specifications was developed. Techniques for characterizing new and used 7808G oils were developed and applied. The components of a reclamation process were defined and the technical feasibility of the process was established.

Two additive packages developed for use in diester/triester base stock mixtures showed considerable promise.

A cost analysis of the process indicated that a continuous batch reclamation process at the 5000-gallon batch size would be cost effective.

Details of the first oil reclamation program can be found in Technical Report AFAPL-TR-78-50, "Reclamation of Synthetic Turbine Engine Oil Mixtures."

### 3. RESULTS AND DISCUSSION

The object of this program is to further improve the reclamation process developed earlier (summarized in Figure 1), define the variables within the process, verify the additive replenishment package, and demonstrate the effectiveness of the total process by reclaiming 12 batches of used synthetic oil to MIL-L-7808H specifications.

#### 3.1 USED OIL CHARACTERIZATION

To determine how widely properties vary between used oil lots and to decide under what conditions some lots should be rejected for reclamation, Aero Propulsion Laboratory (APL) furnished us with 15 different used oil lots. Each lot was examined and characterized (See Appendix C for analysis procedures) by gas chromatography (GC), high performance liquid chromatography (HPLC), infrared (IR) analysis, and acid number. Table 1 lists the quantities and acid numbers of the 15 used oil lots received.

### 3.1.1 <u>Infrared Spectrophotometry</u>

Infrared (IR) analysis was performed on filtered and dried samples of the used oils to see if any gross differences were observable which would interfere with reclamation.

Examination of the IR spectra of used oils 0-79-01 through 0-79-15 and a diester base stock (Figures 2 through 7) show that all are quite similar and made from mixtures of both diester and triester type base stocks. All 15 oil samples show characteristic absorption at 1350 cm<sup>-1</sup> for the diester and at 720 cm<sup>-1</sup> for the typical triester. Oils 0-79-03, -04, -05, and -09 show absorption at about 1100 cm<sup>-1</sup>. This absorption is quite strong in 0-79-05 and is present to a lesser degree in samples 0-79-03, -04, and -09. The remaining oils show absorption in this area to a still lesser degree. This absorption is no doubt associated with the C-O-C bond in the ester, and its shape is influenced by the type of acid structure in the ester.

The IR spectra for all these oils suggest they are composed of di and tri esters, the diester portion being quite similar to the di-2-ethylhexyl azelate virgin base stock shown in Figure 7.

### 3.1.2 Gas Chromatography

Gas chromatography (GC) was also used to characterize the used oils. The chromatograms of the 15 used oils are shown in Figures 8 through 15. Also shown (Figure 16) is the chromatogram for a commercially available diazelate base stock. The major component of this base stock, which elutes at ~29.22 min, and the

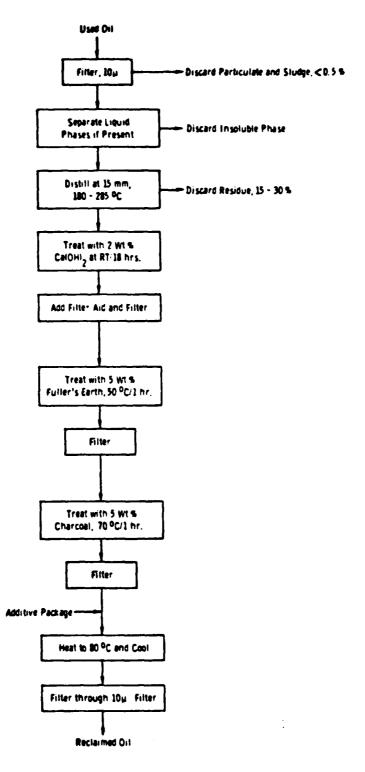


Figure 1. Reclamation process, first program.

TABLE 1. ACID NUMBERS AND QUANTITIES OF OILS RECEIVED FROM APL

Oil number	Quantity, gal	Acid number
0-79-01 0-79-02	35 55	0.42
0-79-03	<b>∿35</b>	0.43
0-79-04 0-79-05	∿35 ∿35	14.45 23.78
0-79-06 0-79-07	55 • 35	1.02
0-79-08	∿35 55	1.15
0-79-09 0-79-10 0-79-11	55 55 55	0.65 0.64 0.88
0-79-12 0-79-13 0-79-14 0-79-15	55 55 55 55	0.52 0.63 1.50 1.72
0-12-12	<b>3</b> 3	1./2

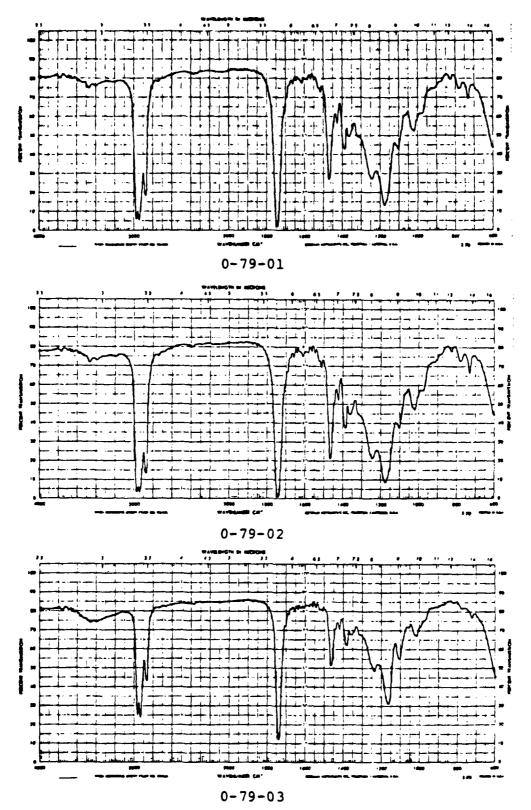


Figure 2. Infrared spectra of used oils.

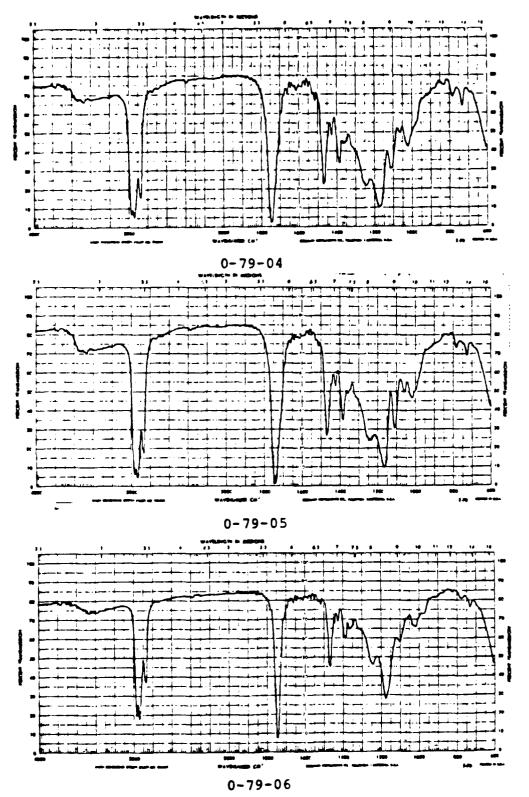


Figure 3. Infrared spectra of used oils.

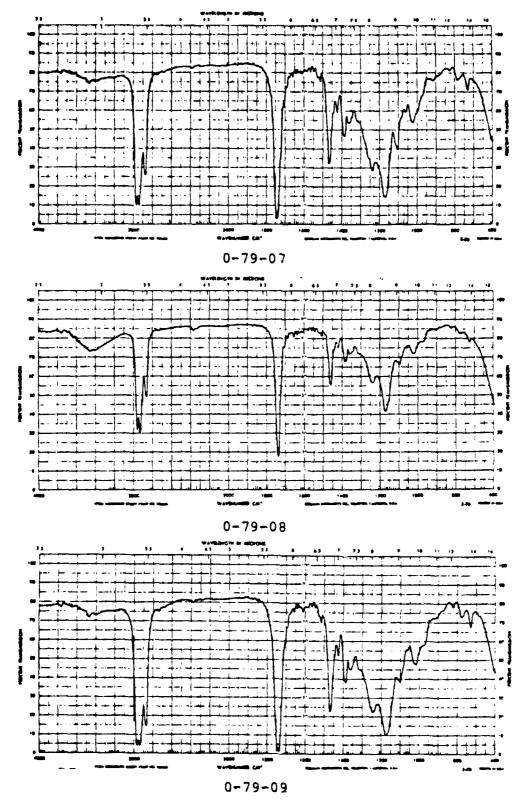


Figure 4. Infrared spectra of used oils.

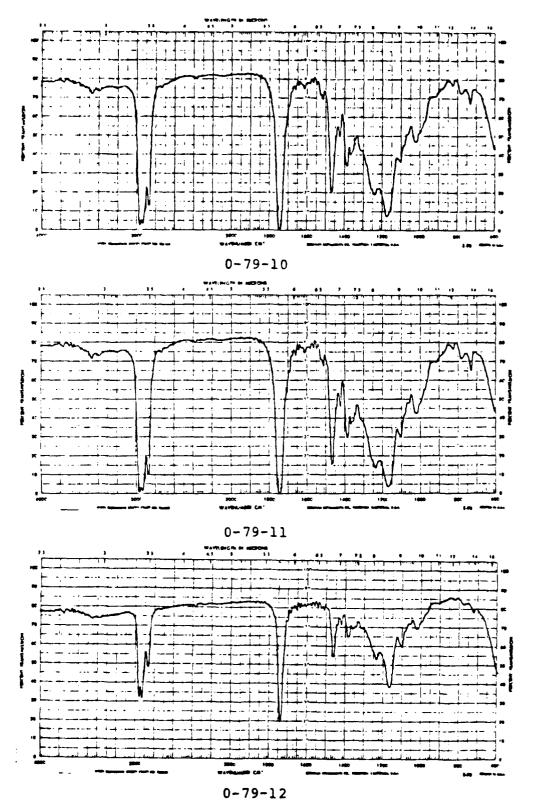


Figure 5. Infrared spectra of used oils.

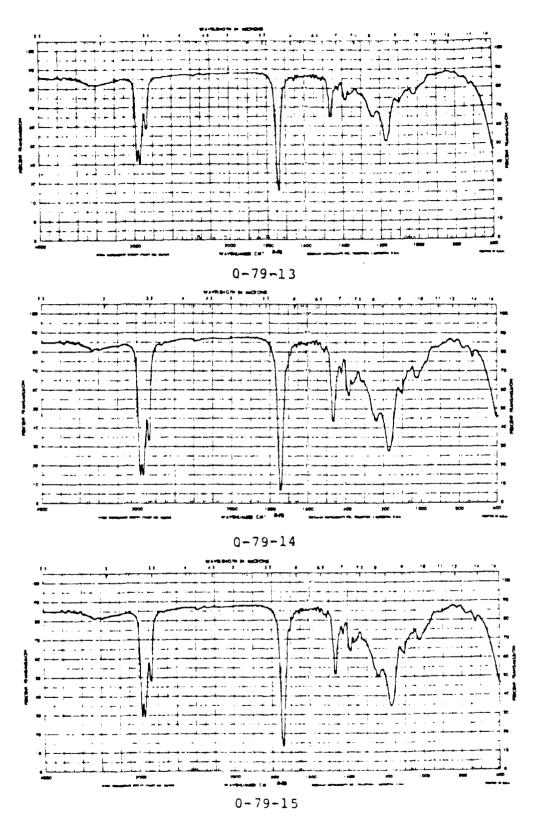


Figure 6. Infrared spectra of used oils.

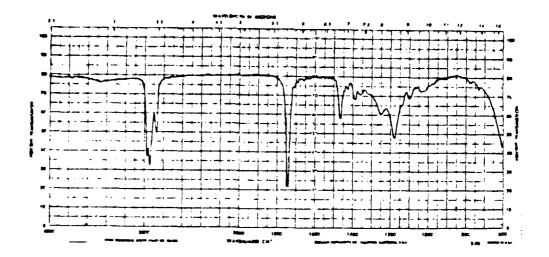


Figure 7. Infrared spectrum of commercially available di-2-ethylhexyl azelate base stock.

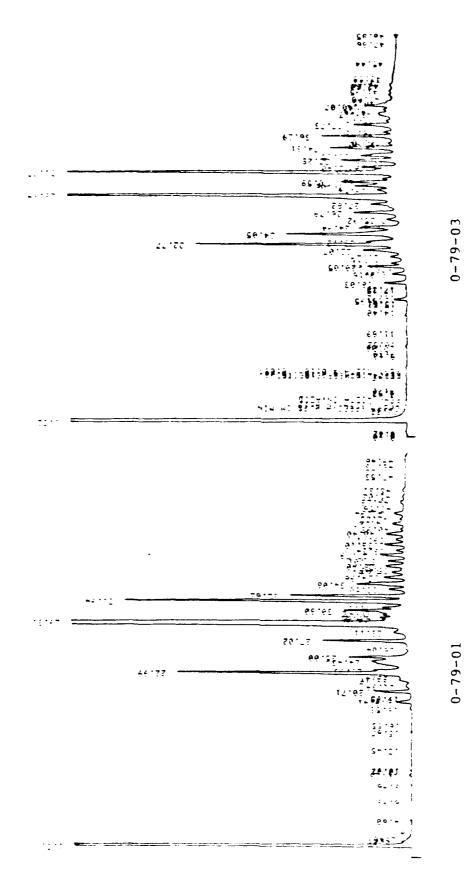
minor peaks at ~20.54, 22.76, 24.83, 26.85, 30.62, 32.4, and 34.07 min are seen in every sample of used oil that has been analyzed. All analyzed oil samples show peaks corresponding to those of the commercial diester but in varying concentrations. These oils are apparently based on mixtures of several components, some of which are quite high boiling in all the samples, but particularly in 0-79-03, -05, and -04.

Such close similarities exist between oil samples 0-79-01, 0-79-07, and 0-79-08 that they may be from the same base stock. Oil samples 0-79-09, 0-79-10, 0-79-11, 0-79-12, and 0-79-13, showing only minor differences, are very similar in composition to each other. Peak analysis of possible diester components suggests that in oils 0-79-03 and 0-79-04 approximately 48% of the composition is due to diester; in 0-79-01, 0-79-07, and 0-79-08 diester comprises approximately 70% of the composition. Of the remaining ten oils approximately 45% of the composition is due to diester.

### 3.1.3 High Performance Liquid Chromatography

Samples which were used for gas chromatographic analysis were examined by high performance liquid chromatography (HPLC) to detect aromatic additives and breakdown products and to establish patterns, if any, for the used oils.

Basic patterns emerged, showing many similarities between all oils. Samples 0-79-01, -02, -06, -07, -08, -10, -11, -12, -13, -14, and -15 are very similar. Sample 0-79-03 is unique in that its gross appearance is that of an almost unused oil. Samples 0-79-04 and 0-79-05 are similar, showing high concentrations of



Gas chromatograms of used oils. Figure 8.

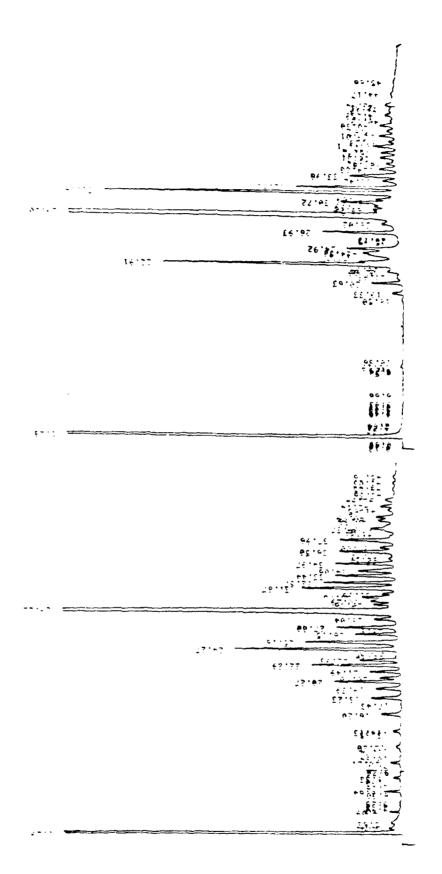


Figure 9. Gas chromatograms of used oils.

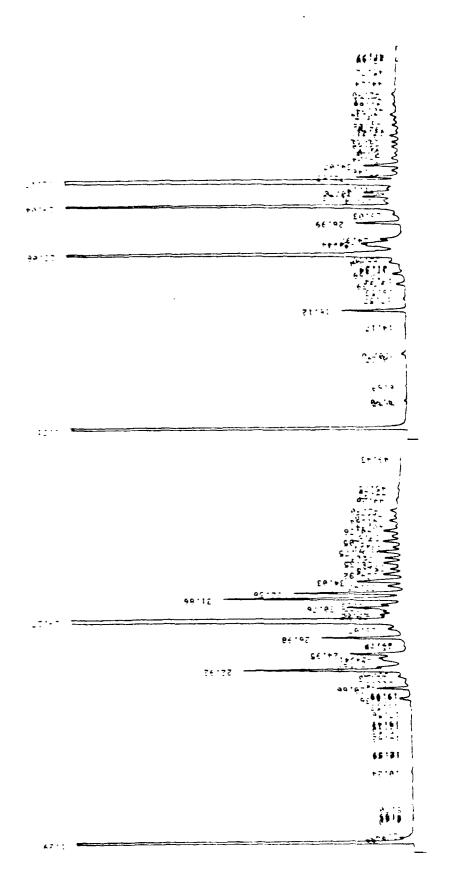


Figure 10. Gas chromatograms of used oils.

80-62-0

0-19-09

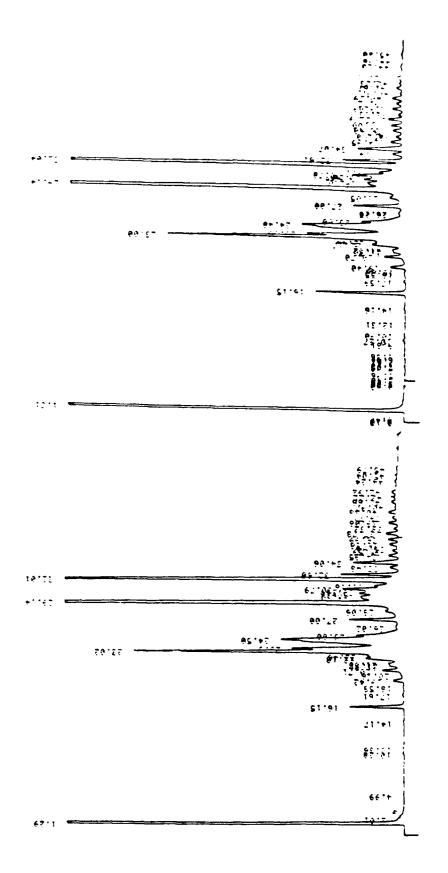


Figure 11. Gas chromatograms of used oils.

0-79-10

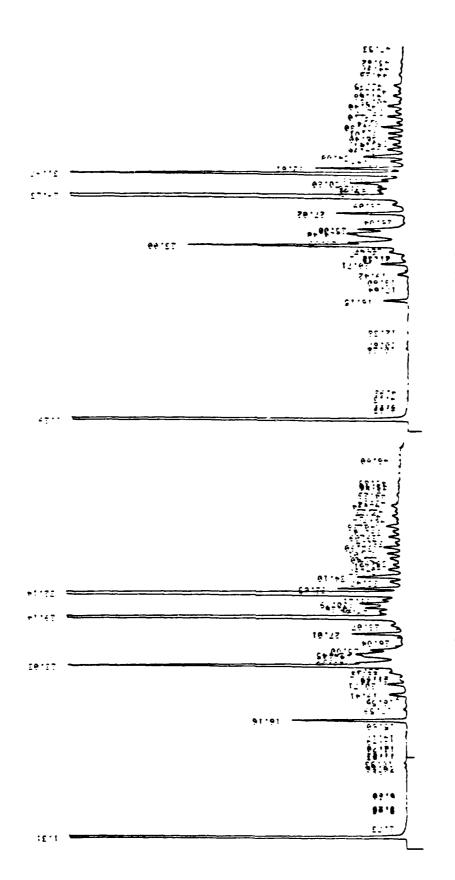


Figure 12. Gas chromatograms of used oils.

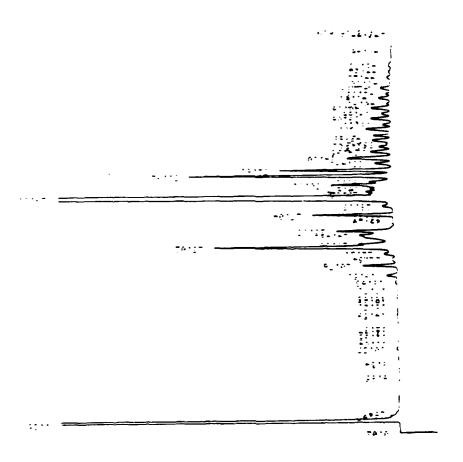


Figure 13. Gas chromatogram of used oils.

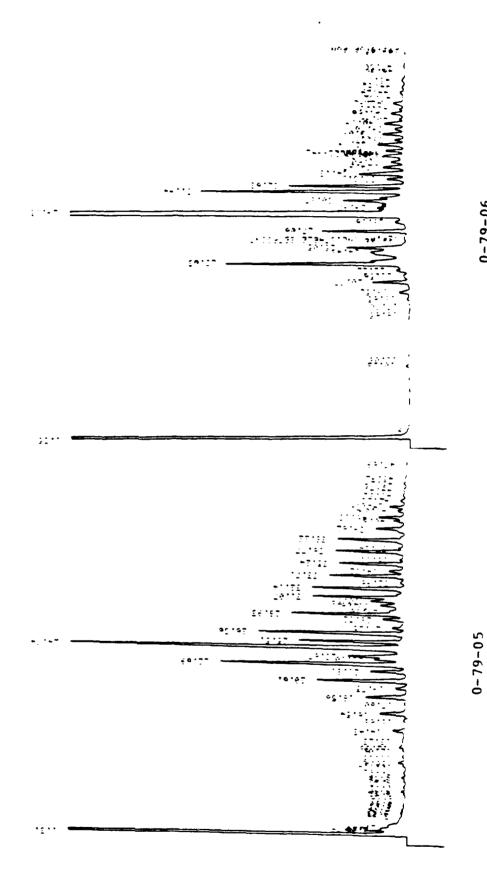


Figure 14. Gas chromatograms of used oils.

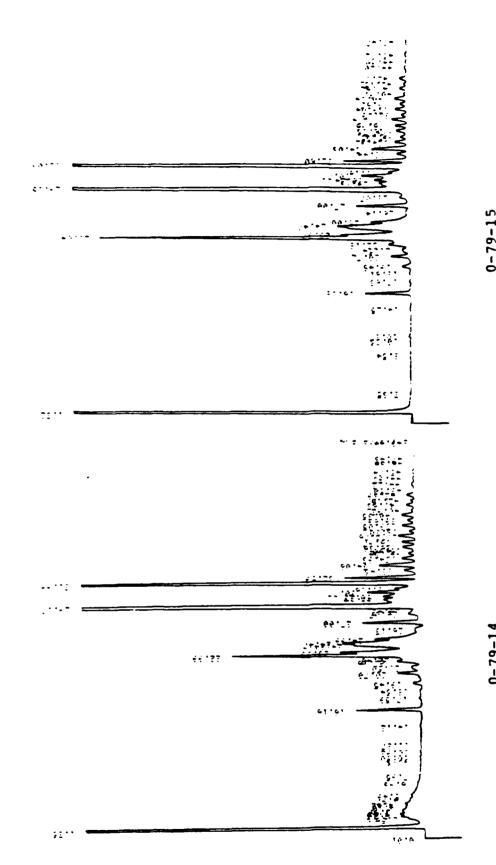


Figure 15. Gas chromatograms of used oils.

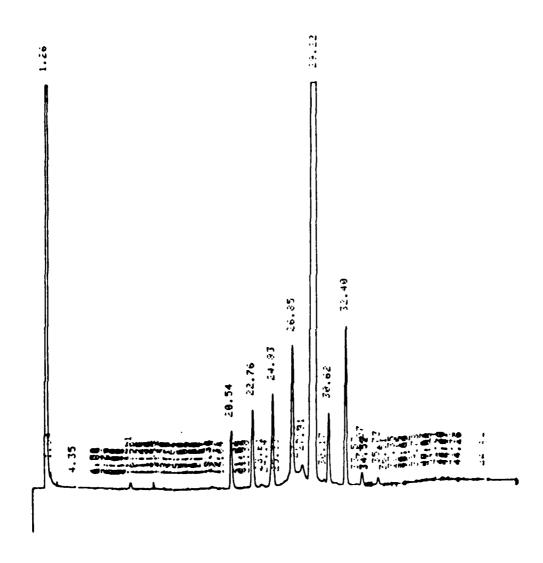


Figure 16. Gas chromatogram of commercially available di-2-ethylhexyl azelate base stock.

cresols. Note that these two oils had very high acid numbers, possibly attributable to the high cresol concentration at approximately 24+ min. Sample 0-79-09 is also unique and not readily comparable with the other 14 oils. Representative chromatograms of each group are shown in Figures 17 and 18. Chromatograms of all the used oils are provided in Appendix D.

### 3.2 OIL CHARACTERIZATION

Six new oils, ATL 9118 through 9123, supplied by AFAPL represent new MIL-L-7808H that could ultimately be found in current used oil mixtures. These new oils were analyzed by HPLC and GC. Among the six oils, only four patterns were seen.

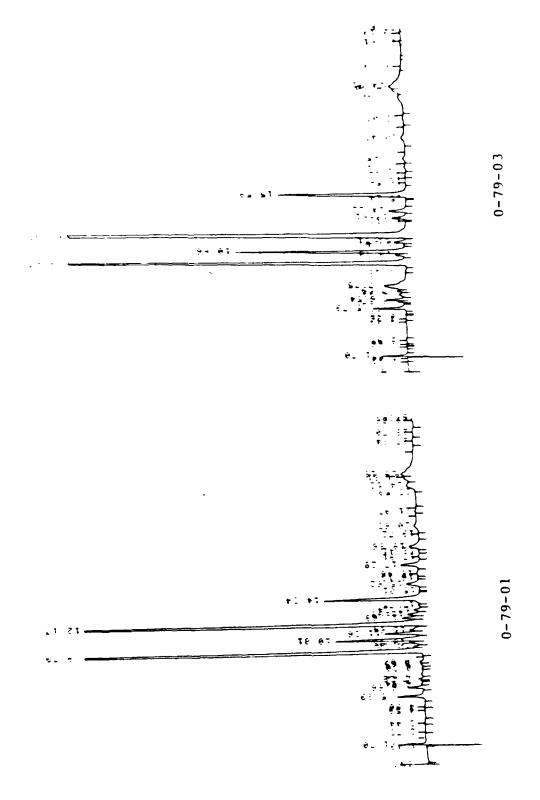
Three oils showed the same HPLC pattern: ATL 9118, 9119, and 9121 (the pattern is represented in Figure 19). They have peaks with the same retention times (rt's) as 4,4'-di-octyldiphenylamine (DODPA) or triphenyl phosphite (TPP), approximately 9.5-10 min; a peak corresponding to the rt's of 3,7-dioctylphenothiazine or N-phenyl-\alpha-naphthylamine approximately 12.0-12.5 min; and a group of peaks corresponding the rt's of tricresyl phosphate and other cresols, 24.0 to 26 min. Sample ATL 9123 (Figure 20) has an additional major peak that corresponds to the rt of phenothiazine, approximately 15.7 min.

Samples ATL 9120 and 9122 (Figure 21) have certain identifiable peaks and, as of yet, major unidentified peaks. Many of these additive peaks are seen in the used oils also, but absolute identification was not completed.

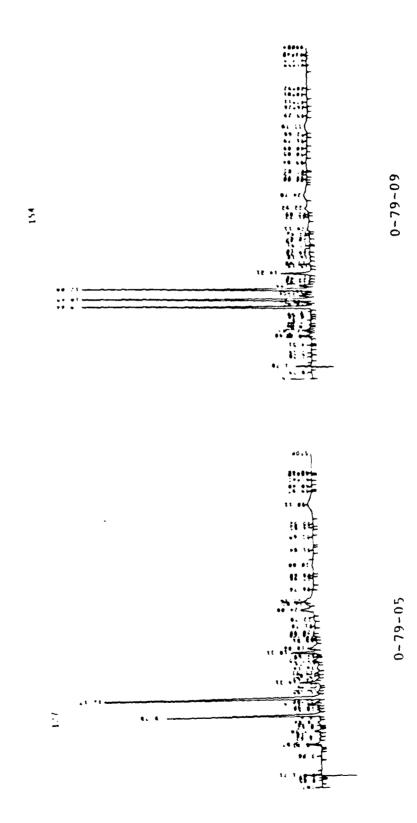
GC's of the new oils are presented in Figures 22, 23, and 24. Molecular weight distribution in ATL 9118 and 9121, also between 9120 and 9123, would suggest possibly the same base stock components.

Throughout this report, the following designations will be used for oils at various stages of reclamation to more readily indicate the state in the reclamation process for any particular sample and eliminate the necessity for lengthy identification of each sample.

Designation	Sample description
Used oil	Oil as received from AFAPL.
Reclaimed base stock	Oil that has been processed, no replenishment of additives.
Reclaimed oil	Oil that has been processed and additives replenished.



High pressure liquid chromatography of used oils. Figure 17.



High pressure liquid chromatography of used oils. Figure 18.

ATL 9121 Figure 19. HPLC representative of new oil ATL 9121.

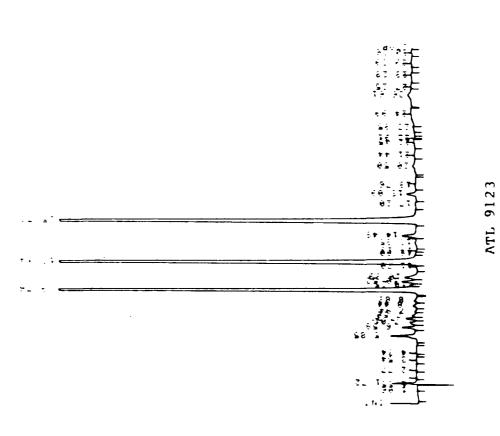
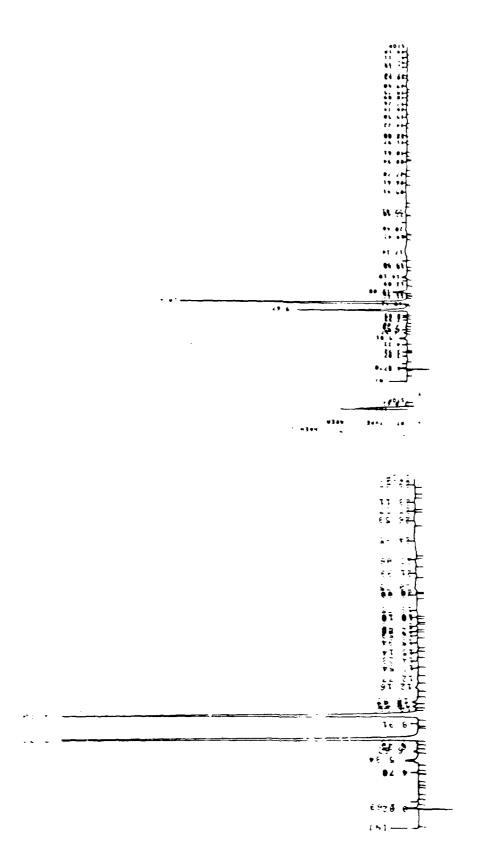


Figure 20. HPLC representative of new oil ATL 9123.



HPLC representative of new oils ATL 9120 and 9122. ATL 9122 ATL 9120 Figure 21.

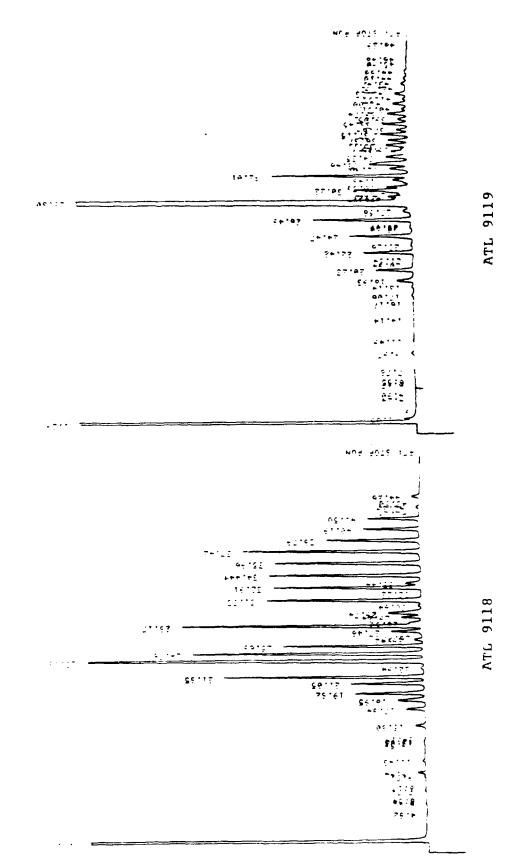


Figure 22. Gas chromatograms of new oils.

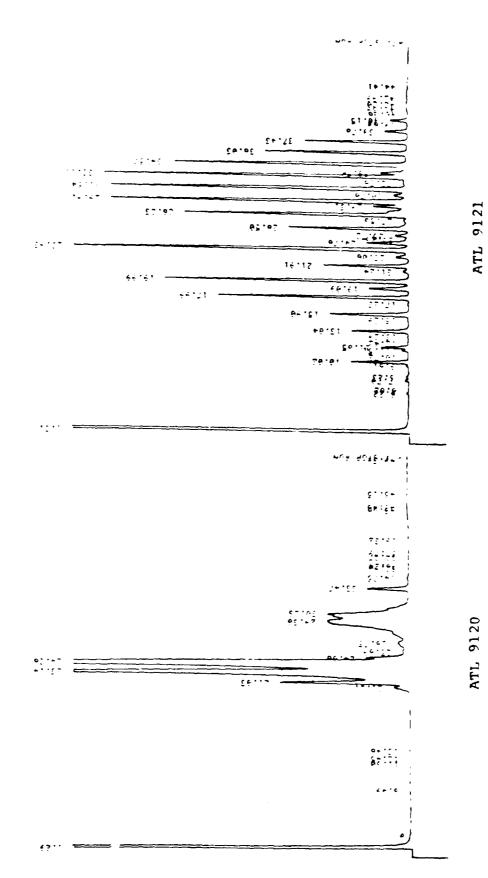
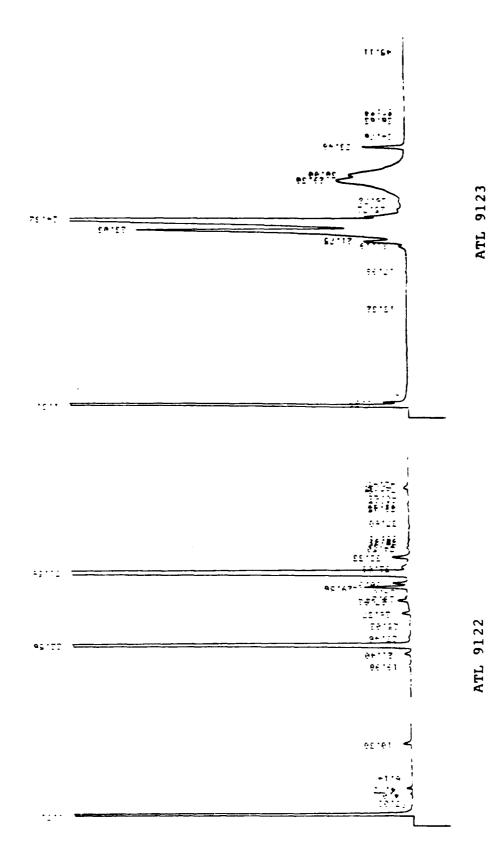


Figure 23. Gas chromatograms of new oils.



Gas chromatograms of new oils.

Figure 24.

29

## 3.3 ADDITIVE PACKAGE VALIDATION

To validate our proposed additive package, GC's were run on six new oil samples supplied by APL, representing various manufacturers. Gas chromatograms labeled ATL 9118 through 9123 are seen in Figures 22, 23, and 24. From these chromatograms, we were able to select five base stock blends (without additives), supplied by various manufacturers and APL, that closely resembled the new oils with which to validate MRC's additive package.

# 3.3.1 <u>Selection of Base Stocks for Use in Additive</u> Package Validation

The following MIL-L-7808 type base stock and base stock blends were received from manufacturers:

Stauffer Chemical Company ~10 gal 7808H Hatco 5 gal 7808H Royal Lube 10 gal 7808H ATL 9148 (from APL) 5 gal mix of polyols ATL 9149 (from APL) 5 gal mix of diesters and polyols Rohm and Haas 5 gal diisooctyl adipate 5 gal diisodecyl adipate 5 gal di-2-ethylhexyl azelate Emery Industries 5 gal trimethylol propane triester 10 gal 5-cst polyol ester Hercules 5 gal Herculube 401

Figures 25, 26, and 27 show gas chromatograms of these base stocks. Figure 28 shows a base stock formulated (1:1:1:1 Royal Lube/Emery 2958/Emery 2932/Herculube 401) and tested with the additive package from the previous program (F33615-76-C-2037), from which a gas chromatogram was not run at the time. Sample 1558834-2A (see Figure 29) represents another, similar sample which was also tested on the prior program.

Our initial tries at blending for viscosity requirements (Figures 29 and 30) show that GC patterns of blended base stocks somewhat match some of the new oils supplied by APL. By using 76.9% Plexol 244 (diisooctyl adipate, Rohm and Haas) and 23.1% Plexol 273 (diisodecyl adipate, Rohm and Haas) a pattern similar to those of oils 9120 and 9123 was produced. A combination of 85% Emery 2958 (di-2-ethylhexyl azelate "mery Lubricants) and 15% Emery 2932 (trimethylol propane triester, Emery Lubricants) gave a pattern somewhat similar to those of oils 9121 and 9118. 92.6% Emery 2958 and 7.4% Emery 2939 (a 5-cSt polyol ester), produced a chromatogram similar to that of new oil 9119.

From these data we formulated and tested five additional base stocks with our additive package: ATL 9149, ATL 9148, Stauffer base stock, Hatco base stock, and a base stock of Plexol 244 and Plexol 273 (see Figure 29). They represent a wide variety of base stock formulations and seem to represent most of the used oils we have received and many of the new oil formulations we have examined.

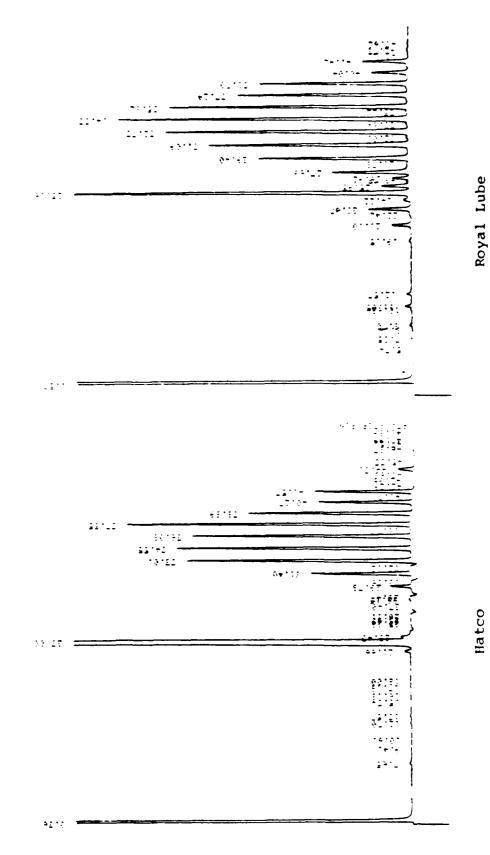


Figure 25. Gas chromatograms of formulated base stocks.

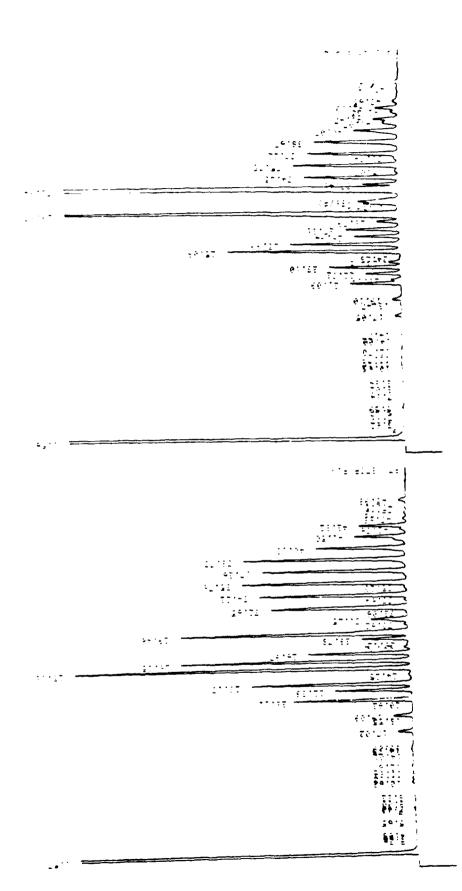
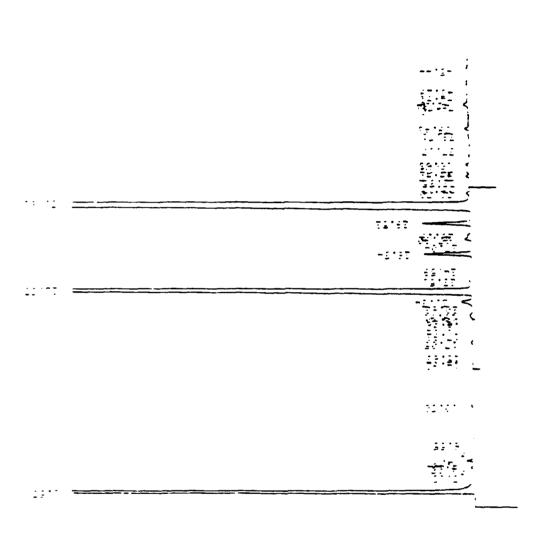
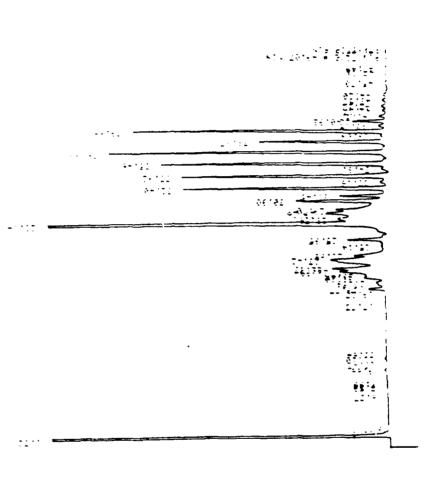


Figure 26. Gas chromatograms of formulated base stocks.

ATL 9148

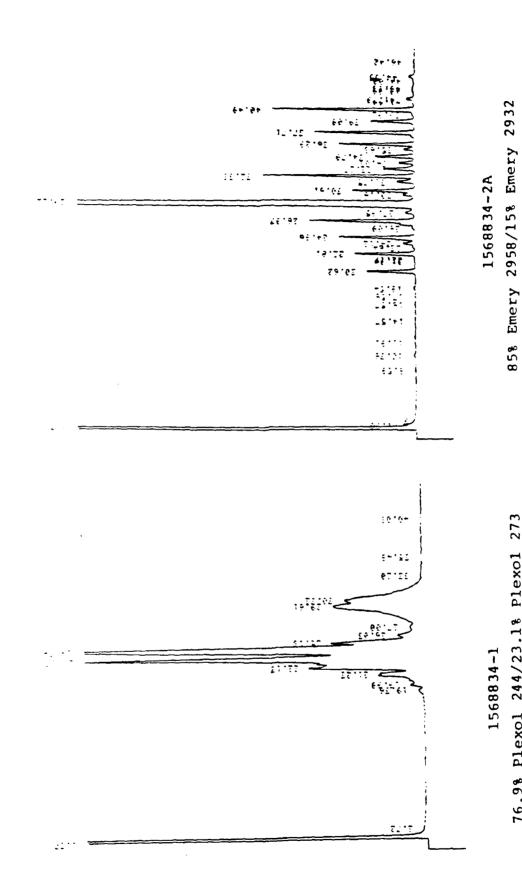


Gas chromatogram of Stauffer formulated base stock. Figure 27.



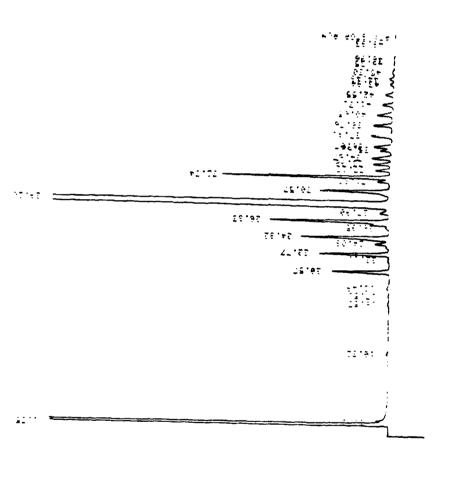
Royal Lube Diester, Emery 2958 and 2932, and Herculube 401 1:1:1:1

Gas chromatogram of base stock mixture tested earlier with a proposed additive package. Figure 28.



Gas chromatograms of blends of commercially available base stocks. Figure 29.

76.9% Plexol 244/23.1% Plexol 273



1568834-3 92.6% Emery 2958/7.4% Emery 2939

Gas chromatogram of a blend of commercially available base stocks. Figure 30.

## 3.3.2 MRC's Additive Package Added to Selected Base Stocks

The 3,7-dioctylphenothiazine (DOPTA) used in the earlier program was replaced with phenyl- $\alpha$ -naphthylamine (PANA) since DOPTA is no longer produced in the United States. We also wanted to determine if the additive had a singular effect on the foam testing of a formulated base stock. Royal Lube base stock was formlated with all of the additives minus PANA and exhibited a foam volume of 5 mL. The formulated base stock with PANA had a foam volume of 45 mL. Additional testing of this base stock with PANA gave similar results (40-45 mL foam volume).

When informed of this foaming problem, the manufacturer provided a new supply of PANA, which produced a foam volume of 15 mL. This newer, cleaner appearing PANA was used to formulate the virgin base stocks.

The five selected virgin base stocks (Hatco, Stauffer, ATL 9148, ATL 9149, and Plexol) were formulated with:

- 1.0% DODPA (dioctyldiphenylamine) Van Lube 81
- 1.0% PANA (phenyl-α-naphthylamine) Uniroyal
- 2.0% TCP (tricresyl phosphate) Kronitrex AA FMC
- 0.1% Ethyl antioxidant 703 Ethyl Corp.
- 0.1% TPP (triphenyl phosphite) Eastman
- 0.1% Benzotriazole photo grade Sherwin Williams
- 0.05% Quinizarin (GAF purified)

These oils were sent to Alcor for specific testing. They were evaluated in the following sequence to allow detection of the most common modes of failure.

- 1. Total acid number
- 2. Static foaming characteristics, test method 3213
- 3. Viscosity at -65°F, 100°F, and 210°F
- 4. Lead corrosion
- 5. Corrosion and oxidation stability, 96 hr at 392°F
- 6. Dynamic foaming characteristics
- 7. FA elastomer compatibility at 347°F
- Gear load carrying rating (one gear, two determinations)
- 9. Silver and bronze corrosion
- 10. Deposition number

The results of these tests are included in Appendix A of this report. MRC's proposed additive package in the five selected base stocks had met all test specifications except the static foam test in one of the base stocks furnished by APL. A comparison of the MRC versus Alcor static and dynamic foam test results on the five formulated base stocks (see Table 2) suggests that the static foam volume of 115 mL for oil 1732509 (ATL 9149),

TABLE 2. SELECTED DATA FROM MIL-L-7808H TESTING

Oil	MRC static foam volume, mL	Alcor static foam volume, mL	Acid b	Alcor dynamic foam volume, mL	Viscosity at 210°F, cSt	Viscosity at -65°F, cSt 35 min
Hatco 1732506	15	15	0.20	10	3.0	12,136
Stauffer 1732510	25	30	0.25	10	3.1	13, 325
Rohm and Haas 1732511	35	35	0.14	10	3.0	11,341
ATL 9148 1732508	09	100	0.22	10	3.4	14,462
ATL 9149 1732509	06	115	0.17	10	3.5	16,642
ATL 9148 <sup>d</sup>	ស					
ATL 9149 <sup>d</sup>	0					

Amaximum 100 mL.

b Maximum allowable acid number is 0.30.

 $^{
m C}_{
m Foam}$  volumes same at both 176 $^{
m G}_{
m F}$  and 230 $^{
m G}_{
m F}$ , 1000 cc air.

d Base stock with no additives; only static foam volume tested.

is only a marginal failure. Another oil (1732511, Rohm and Haas), though not out of specifications as far as corrosion and oxidation stability (96 hr at 392°F), did exhibit high magnesium corrosion and oxidation values.

Effects on acid numbers, viscosity, and foam volume of the same additive package in various MIL-L-7808 fluids, produced by unique manufacturing processes, are seen in Table 2. The base stocks received from AFAPL seem to be much more affected by the formulation process than the commercially available virgin base stocks. The cause of this effect on the APL base stocks is unknown.

The interaction on PANA with the APL-supplied base stocks may contribute to the higher than normal foam volumes. After the virgin base stocks had already been formulated, PANA from another supplier arrived for testing. Table 3 compares test results for PANA from Union Carbide and from Uniroyal; both produced equal response. The data suggest base stock interaction with the additive package, specifically with PANA.

TABLE 3. COMPARING FOAM TEST RESULTS PANA FROM TWO SUPPLIERS

MRC's	Foam test volume, mL		
additive	Royal lube	ATL 9149	
package with	base stock	base stock	
Uniroyal PANA	15	115	
Union Carbide PANA	25	120	

## 3.3.3 Effect of Additive Package on Viscosity

From the previous contract, an increase in viscosity was expected from the addition of our proposed additive package to base stock oils. Results were much more sporadic than expected. These random viscosity increases (see Table 4) fall into distinct patterns, making future viscosity increase predictions much more difficult.

TABLE 4. VISCOSITY INCREASES DUE TO ADDITIVE ADDITION

		iscosity meas -20°F	ured in o	
ATL 9149 ATL 9148	16.9 17.0	Pattern 1	3.0 3.6	Pattern 1
Rohm and Haas Stauffer	27.6 22.9	Pattern 2	10.0	Pattern 2
Hatco	7.6	Pattern 3	0.4	Pattern 3

### 3.3.4 Chromatograms of Additives in Base Stock

Chromatograms of MRC's additive package in Stauffers base stock are shown in Figure 31. All the high performance liquid chromatograms of the formulated fluids tested by Alcor are the same. The gas chromatograms of the formulated fluids are duplications of those for the base stocks without additives, although PANA does appear at 18.24 min in each chromatogram.

## 3.3.5 Estimated Additive Shelf Life

According to their manufacturers, the additives contained in MRC's additive package have different shelf lives. DODPA and PANA should be tested yearly; quinizarin and ethyl antioxidant 703 should not be used if stored over 2 years; and TCP, benzotriazole (BT), and TPP are unaffected by storage.

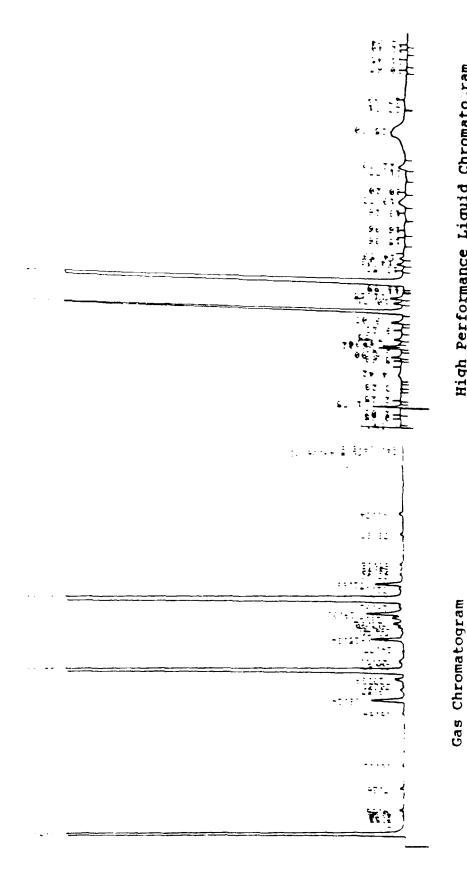
#### 3.4 DISTILLATION STUDY

Distillation ideally should be included in the early part of any reclamation process to remove any low boiling material (water, toluene, etc.) and major contamination such as degraded additives and polymeric esters. Removal of the major contaminants insures lower effective absorbent treatment levels in subsequent process steps.

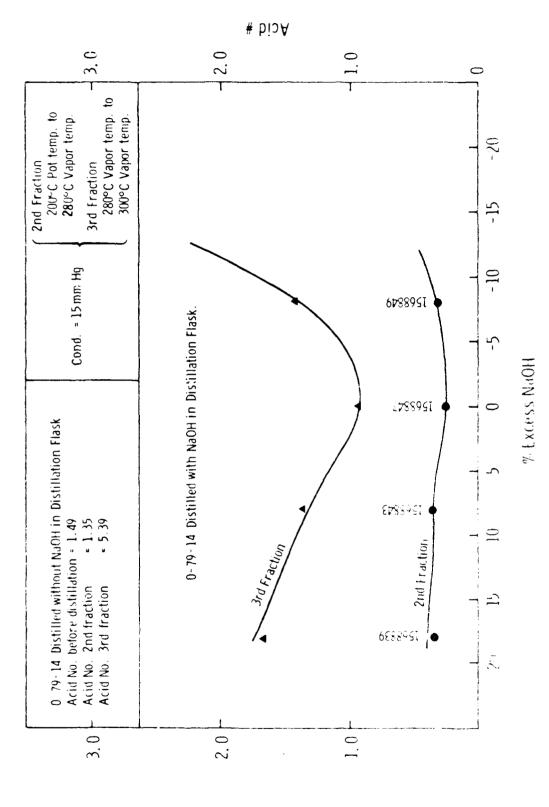
## 3.4.1 Distillation of Neutralized Used Oils

In the earlier program, a straight takeover distillation was performed. In the current program, a literature search revealed a paper on commercial oil reclamation which suggests that treatment of oils with metallic sodium prior to distillation reduces the formation of nondistillable impurities [1]. The metallic sodium also reacts to form nondistillable compounds with some degradation products. Since any water of low boilers would normally be removed during the topping step on distillation, it was felt we could add aqueous or alcoholic caustic solutions to the oil prior to distillation and possibly achieve the same effects. During this study, sodium hydroxide (NaOH) at 5% volume dissolved in water/isopropyl alcohol (1/1) was added to the oil just prior to The amount of NaOH added to the oil was ba. distillation. the stoichiometric amount (or proportions thereof) necessary to neutralize the acid number. Figure 32 shows a graph comparing the effects of varying the percentage of NaOH on the acid number of the 2nd (main) and 3rd (secondary) fractions. The samples were distilled under the same conditions and cut-off points. Oil 0-79-14 was used for this experiment. Acid numbers of the same oil distilled without NaOH treatment are presented with the graph.

<sup>[1]</sup> Reynolds, J. W.; Whisman, M. L.; Brinkman, D. W.; Goetzinger, J. W.; Cotton, F. O. "From Oil:Oil," Chemtech, October 1979.



High Performance Liquid Chromatogram Figure 31. Chromatograms of Stauffer base stock with MRC additive package.



Arid numbers of distillate fractions, after varying North Poved in used oil 0-79-14. Figure 32.

Because of improved acid numbers and obvious visual improvements in color, additional studies were warranted.

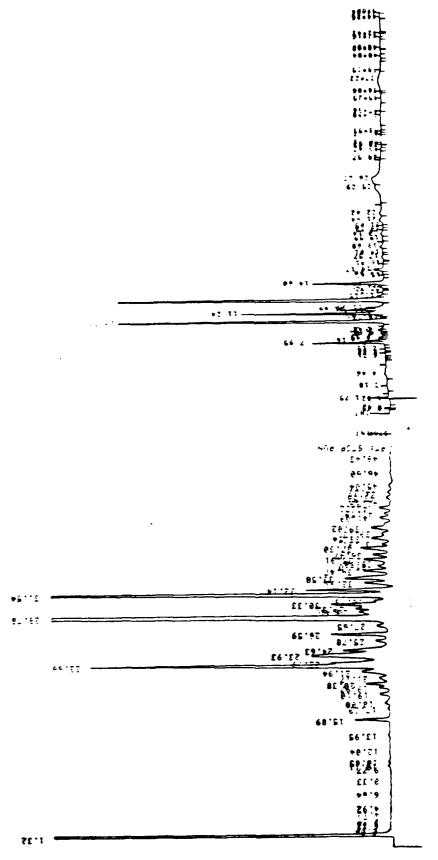
Two samples of a mixture of all 15 oils in the volume proportions received (see Figure 33 for HPLC and gas chromatograms) were distilled, one treated with sufficient NaOH to effect total neutralization and the other without treatment. Samples were taken throughout distillation for analysis. Figure 34 shows results for the mixture distilled without caustic treatment and Figure 35 gives the corresponding results for the mixture treated with NaOH. The circles represent vapor temperature vs. acid number and the squares represent cumulative percent recovery vs. vapor temperature.

There is virtually no difference between the plots of cumulative recovery vs. vapor temperature, suggesting there is no deleterious effect on recovery by NaOH treatment. The acid number data show significant differences. The total amount of distilled material having low acid numbers is much greater in the NaOH treated sample. These lower acid numbers may mean that less postdistillation treatment will be required. This plot might also be used as an accurate means of determining a proper distillation cut-off point. HPLC analyses show no differences between the treated vs. non-treated samples. The treated samples were lighter colored, having fewer visible degradation products present. Thin layer chromatography (TLC) analysis did show removal of additional material from the NaOH-treated samples.

## 3.4.2 Analysis of Distilled Neutralized Used Oils

Various analyses (GC, HPLC, acid number, and foam testing) were performed on oil samples distilled with and without NaOH treatment. A GC comparison of three equal (vapor temperatures) points on the recovery curves (Figures 36-38) showed no differences between the treated and nontreated samples. HPLC comparisons of combined distillates in the same vapor temperature ranges are shown in Figure 39. No differences are seen. These combined samples were foam tested and acid numbers were run with the following results:

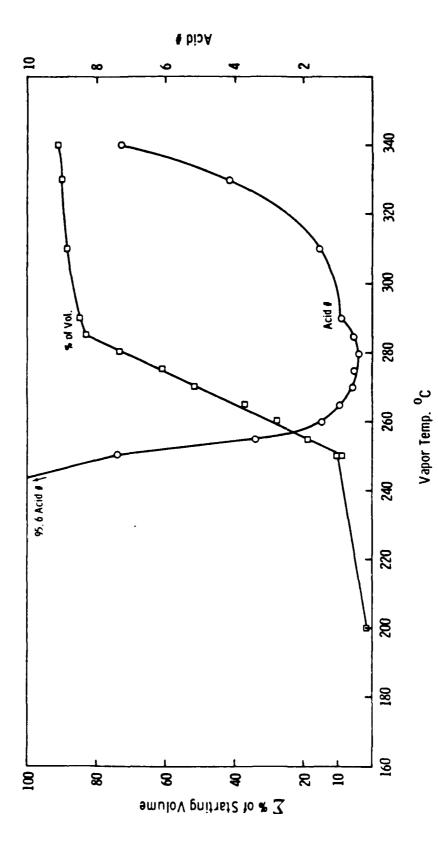
	Foam volume, mL	Acid number
1568887-A (156882 mixed fractions 3 through 11, distilled with no NaOH)	15	1.40
1568887-B (1568865 mixed fractions 4 through 11, distilled with NaOH)	10	0.35



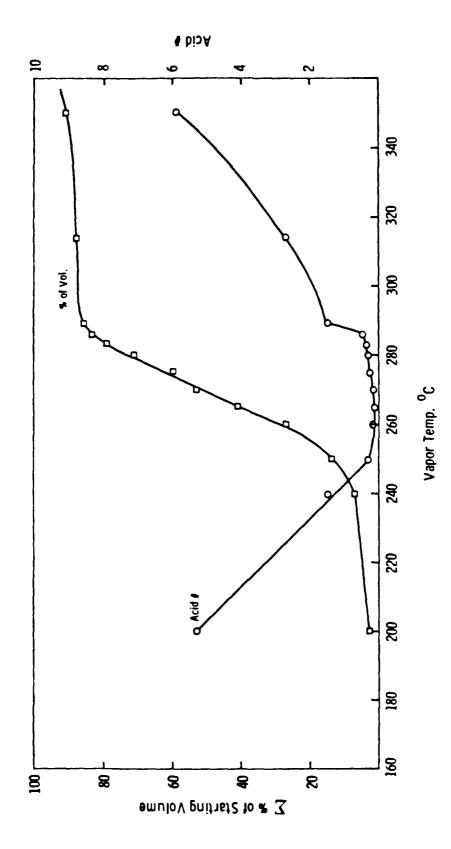
Chromatograms of the mixture of 15 used oil samples. Figure 33.

High Performance Liquid Chromatogram

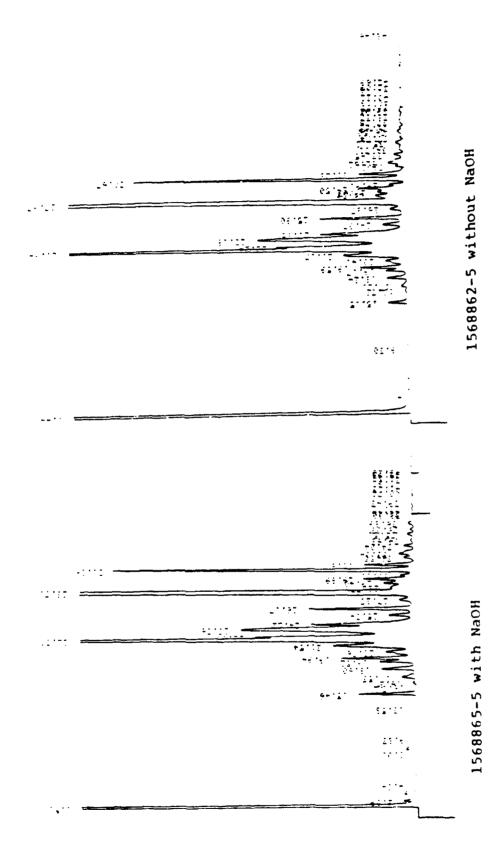
Gas Chromatogram



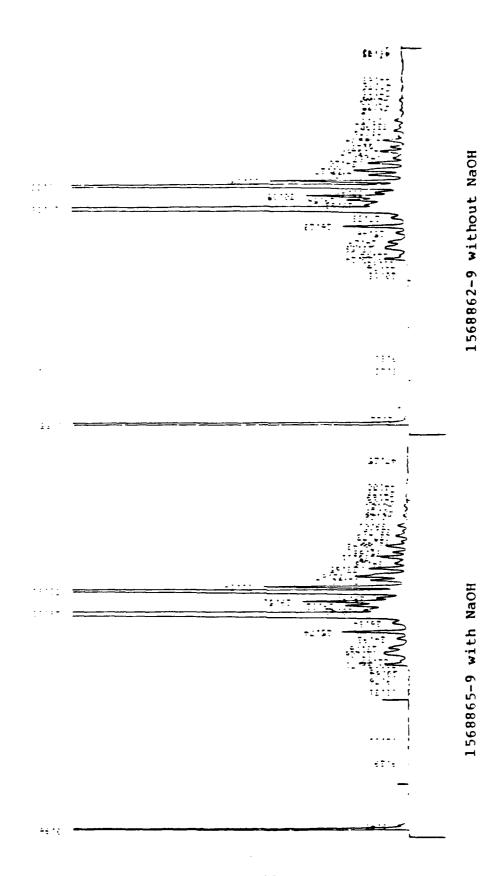
Sample 1568862, mixture of 15 used oils distilled with no NaOH (used oi starting acid number: 2.88). Figure 34.



Sample 1568864, mixture of 15 used oils distilled with sufficient NaOH (added with water/isopropyl alcohol) for neutralization (used oil starting acid number: 2.88). Figure 35.



Gas chromatograms of distillation samples at 260°C vapor temperature. Figure 36.



Gas chromatograms of distillation samples at 280°C vapor temperature. Figure 37.

Gas chromatograms of distillation samples at 310°C vapor temperature. Figure 38.

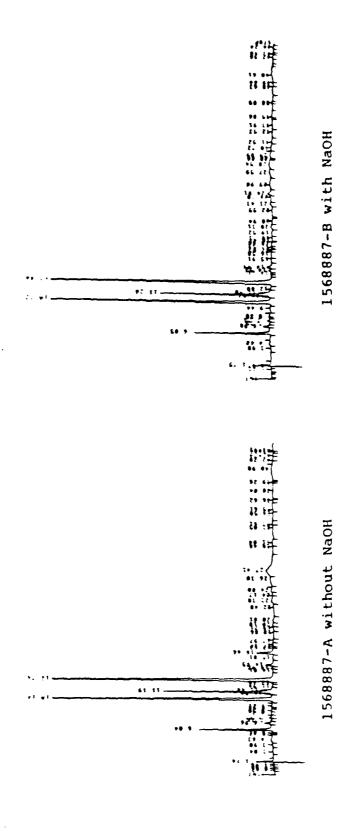


Figure 39. High performance liquid chromatogram of combined distillation samples.

Acid number was significantly improved by this treatment. Results show that NaOH (caustic) treatment does not adversely affect the combined distillates. In fact, the data suggest that such treatment improves acid number, color, and odor, and may eliminate some portions of posttreatment.

## 3.4.3 Substitution of Other Basic Material for Sodium Hydroxide

Because results of distillation with NaOH were so promising, we sought a more practical substitute. Since calcium hydroxide [Ca(OH)<sub>2</sub>] can be added as a powder, its use would save total process time and energy. When Ca(OH)<sub>2</sub> was substituted, however, a solidified rubbery gel materialized in the still at 320°C, with a residue level of 13% of the original oil volume. These disappointing results terminated the study.

Other quantities of the mixture of the 15 used oils were distilled with NaOH/methanol, sodium carbonate, magnesium oxide, and a combination of 75:25 NaOH/methanol and magnesium oxide. Comparisons were made with previous distillations of NaOH/water:isopropyl alcohol. Figures 40 through 43 contain graphs showing acid number vs. vapor temperature vs. cumulative volume percent of these distillations. HPLC's of distillates in the same vapor temperature range are shown in Figures 44 and 45. Examination of all data indicates the following decreasing order of effectiveness:

- NaOH/methanol (Figure 40)
- 2. NaOH/methanol with magnesium oxide (Figure 41)
- 3. Magnesium oxide (Figure 42)
- 4. NaOH/water:isopropyl alcohol (Figure 35)
- 5. Sodium carbonate (Figure 43)

## 3.4.4 Comparison of NaOH Distillation of Used Oil and New Oil

Virgin oil containing MRC's proposed additive package was distilled with and without NaOH treatment to examine the distillation process with completely known additives. High performance liquid chromatograms (HPLC) of two selected distillate fractions, at the same vapor temperature range, are shown in Figure 46. It can be readily observed from the chromatograms that NaOH had removed material from the distillate. Thin layer chromatography studies of the same distillates confirmed the HPLC results. A typical chromatogram (HPLC) of MRC's additive package in a virgin base stock is shown in Figure 31.

Plots of acid number vs. vapor temperature vs. cumulative percent recovery of the virgin oil discussed above are shown in Figures 47 and 48. These two figures reemphasize the decrease in acid number with no deleterious effects on recovery by NaOH distillation.

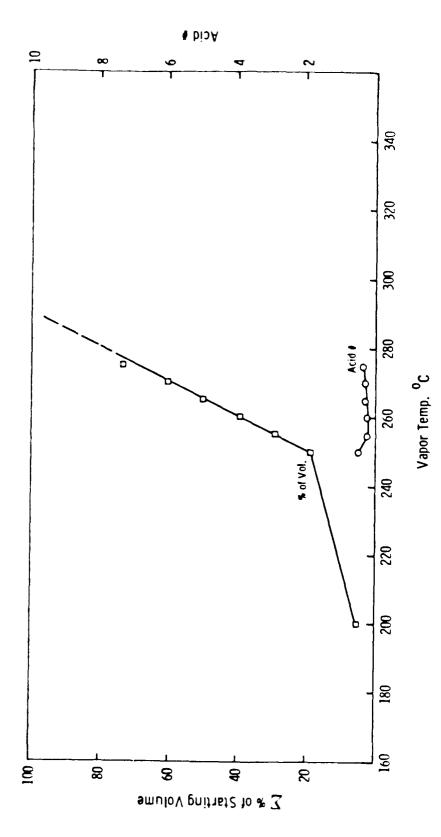
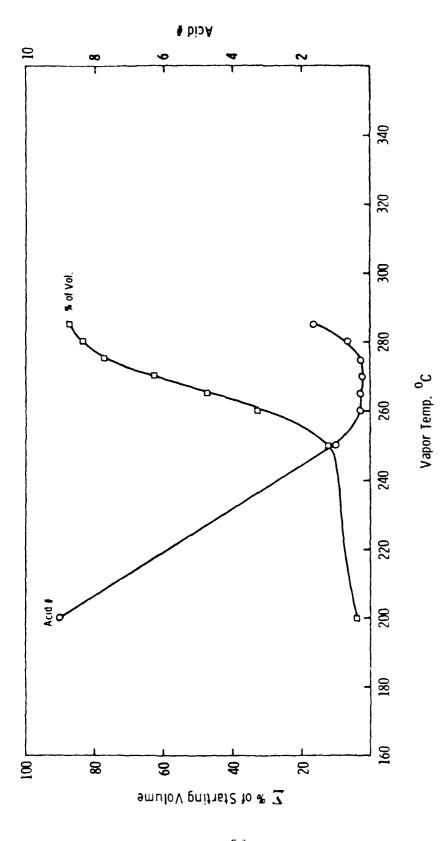
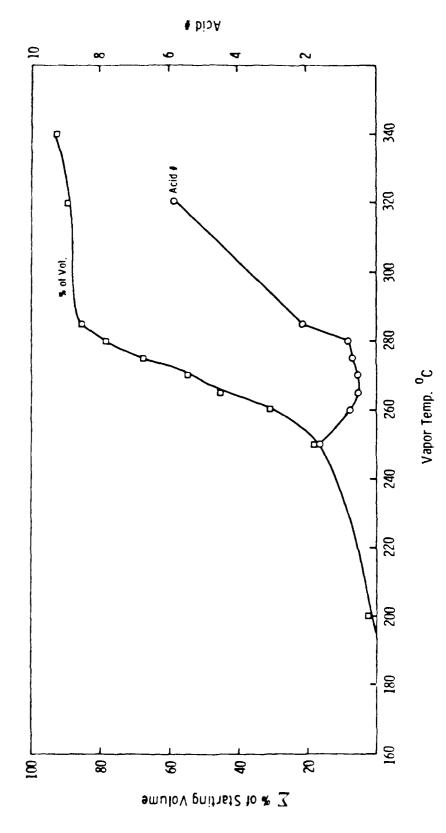


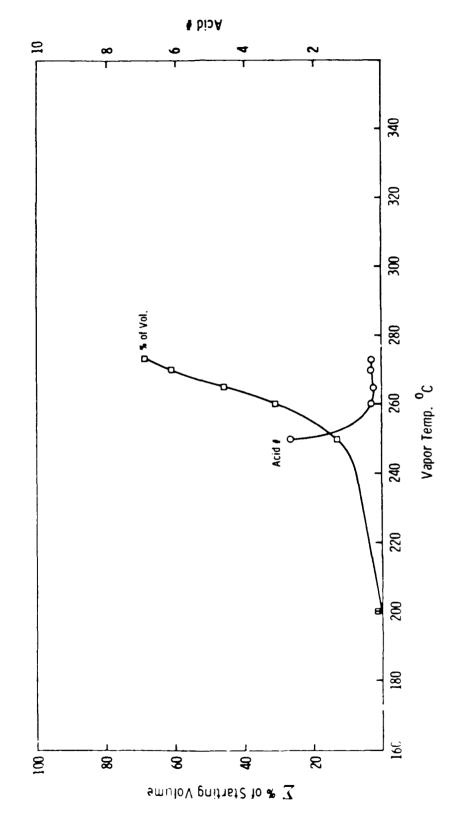
Figure 40. Sample 1732568, mixture of 15 used oils distilled with NaOH/methanol.



Sample 1732583, mixture of 15 used oils distilled with 75:25 NaOH/methanol:magnesium oxide. Figure 41.



Sample 1732570, mixture of 15 used oils distilled with magnesium oxide. Staure 42.



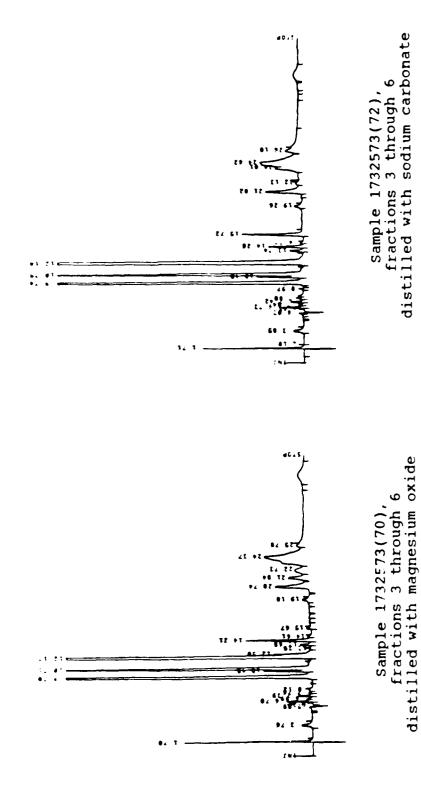
Sumple 1732572, mixture of 15 used oils distilled with sodium carbonate. Figure 43.

17

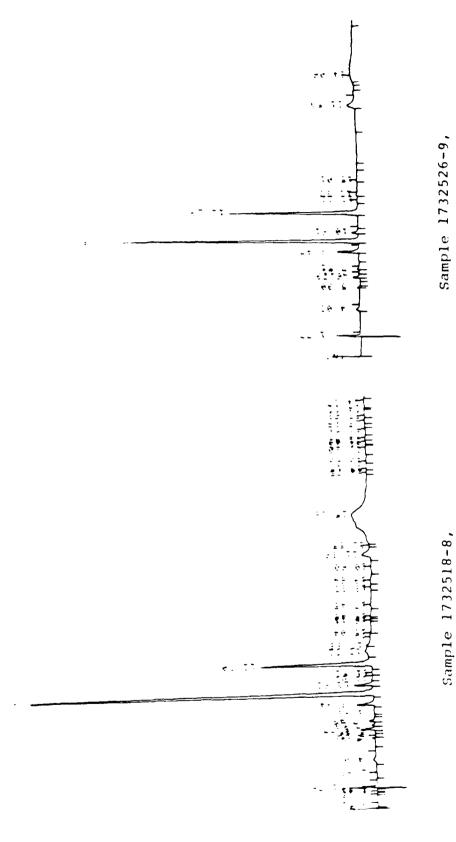
Sample 1732573(68), fractions 4 throw;h 7 distilled with NaOH/methanol

Sample 1732583, fractions 3 through 6 distilled with 75:25 combination of NaOH/methanol with magnesium exide

High performance liquid chromatograms of oil distillates in 260-275°C vapor temperature range. Figure 44.



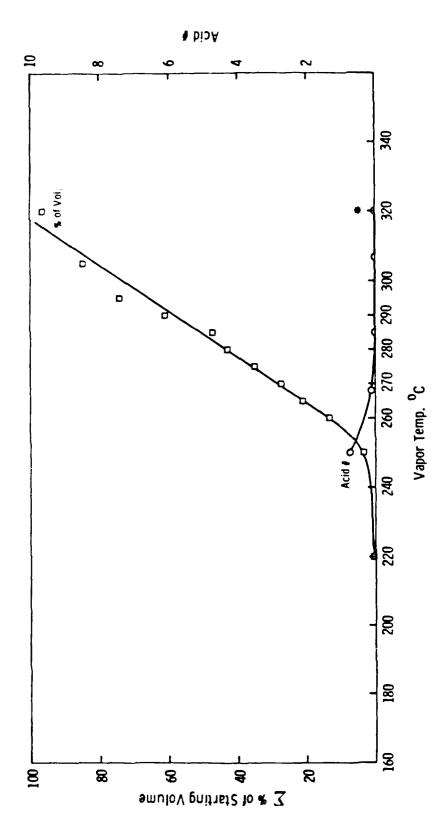
High performance liquid chromatograms of oil distillates in 260-275°C vapor temperature range. Figure 45.



HPLC's of two selected fractions of virgin oil containing MRC proposed additive package. Figure 46.

NaOH distillation

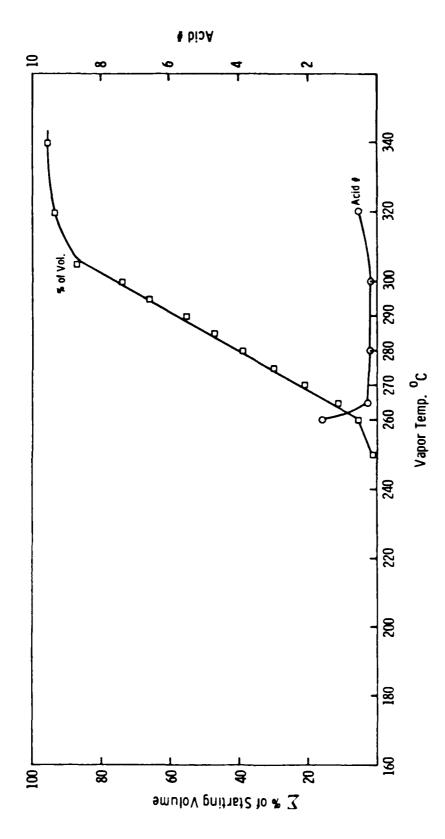
regular distillation



Sample 1732526, Royal Lube base stock with MRC proposed additive package, distilled with sufficient NaOH for neutralization. Figure 47.

Oil had Bumped over on this Fraction

from Distillation Flask



Sample 1732518, Royal Lube base stoc!: with MRC proposed additive package. Figure 48.

#### 3.4.5 Scaleup of Selected Distillations to 13-Liter Quantities

Results from previous small-scale (300-mL) distillations using various basic materials added to used oil warranted scaleup to 13-liter batch distillations with NaOH/methanol, 75/25 NaOH/ magnesium oxide in methanol, NaOH/isopropyl alcohol:water, and regular distillation. Tables 5 and 6 summarize the intended and actual distillation conditions of the four batches. distillation conditions listed in Tables 5 and 6 should be compared to prior distillation studies shown in Figures 34, 35, 40, and 41. Intended conditions were chosen by expected acid numbers and quantity recovered. The differences between the two sets of conditions were partially brought about by unfamiliarity with the larger scale equipment, and by the extension of the distillation time (2 hr to 7-9 hr) caused by distilling larger oil quantities. Even with scaleup, and startup difficulties, NaOH distillation still improves acid numbers.

From these experiments, we produced sufficient quantities of distilled oil for adsorbent and other studies.

#### 3.5 USE OF CALCIUM HYDROXIDE TO LOWER ACID NUMBERS

Our reclamation process will have to lower acid numbers to a level in which attapulgus clay is most effective in meeting MIL-L-7808H specifications. Data from earlier work in this program, described in Section 3.6, suggested the use of a slurry with sufficient Ca(OH)<sub>2</sub> to neutralize the acid number, but data from distillations with acid numbers around 0.50 or less did not support this idea. Results of a study initiated to determine the necessary level of Ca(OH)<sub>2</sub> are presented in Table 7. The data generally indicate 0.2% by weight Ca(OH)2 as the minimum amount, unless a high acid number requires a larger amount for neutralization. In some instances treated samples of the same distillate do not have comparable acid numbers, but the numbers are still sufficiently lowered for reformulation. For our process, we will use a minimum of 0.3% by weight of  $Ca(OH)_2$ ; The maximum will be the amount necessary to neutralize larger acid numbers. The time required to slurry the Ca(OH)<sub>2</sub> will be determined individually for each batch reclaimed in the pilot plant, and the resultant data will enable us to determine a minimum slurry time.

#### 3.6 ADSORPTION TREATMENT STUDY

Another part of our reclamation process will be an adsorption treatment to remove surfactants, trace metals, and other material that distilled over and was not removed during Ca(OH)<sub>2</sub> treatment.

LARGE-SCALE (13-LITER) DISTILLATION CONDITIONS TABLE 5.

	Intended distillation conditions <sup>a</sup>	stillat	ion condi	tions	Actual dis	Actual distillation conditions <sup>a</sup>	nditions <sup>a</sup>
	Distillation temperature,	n Pe	Percent original	Estimated acid	Distillation temperature,	n Percent , original	Acid L
Batch identify	၁့	>	volume	number	၁့	volume	number
			17325	1732588 with NaOH/methanol	H/methanol		
lst Fraction Precollection	200 PT <sup>C</sup> -240 VT <sup>d</sup>		16.0	1.5+	200 PT-240 VT	T 5.1	3.73
2nd Fraction Collected for intended refor- mulation	240 VT-285 VT		73.0	0.25	240 VT-285 VT	T 70.5	0.45
3rd Fraction Postcollection	285+ VT		0.9	1.5+	285 VT-305 VT	1 9.0	1.58
	1	732591	with 75/2	5 NaOH/magm	1732591 with 75/25 NaOH/magnesium oxide in methanol	n methanol	
lst Fraction Precollection	200 PT-250 VT		11.0	1.5+	200 PT-250 VT	T 7.0	2.57
2nd Fraction Collected for intended refor- mulation	250 VT-285 VT		76.0	0.40	250 VT-282 VT	T 65.5	0.32
3rd Fraction Postcollection	285+ VT		8.0	1.5+	282 VT-314 VT	т 20.0	1.31

al4-15 mm pressure.

 $^{\rm b}$ Original acid number for oil before distillation = 2.45.

 $^{\text{C}}_{\text{PT}} = \text{pot temperature.}$ 

 $d_{\rm VT}$  = vapor temperature.

LARGE-SCALE (13-LITER) DISTILLATION CONDITIONS TABLE 6.

	Intended distillation conditions <sup>a</sup>	llation con	ditionsa	Actual distillation conditions <sup>a</sup>	lation condi	itionsa
	Distillation temperature,	Percent original	Estimated acid	Distillation temperature,	Percent original	Acid
Batch identity	၁့	volume	number	ی د	volume	number
		NaO	H/isopropyl	NaOH/isopropyl alcohol/water		
lst Fraction Precollection	200 PT <sup>C</sup> -240 VT <sup>d</sup>	15	1.5+	200 PT-240 VT	1.7	8.95
2nd Fraction Collected for intended refor- mulation	240-290 VT	71	0.3	240-290 VT	65.5	2.10
3rd Fraction Postcollection	290+ VT	ស	1.5+	290-335 VT	27.2	1.70
		Regular di	stillation	Regular distillation (without treatment)	t)	
lst Fraction Precollection	200 PT-255 VT	19	3.4+	200 PT-260 VT	9.5	23.88
2nd Fraction Collected for intended refor- mulation	255-320 VT	71	1.0	260-310 VT	82	16.3
3rd Fraction Postcollection	320+ VT	7	2.7+	310-330 VT	5.2	7.3

al4-15 mm pressure.

 $^{\rm b}$ Original acid number for oil before distillation = 2.69.

 $^{C}PT = pot temperature.$ 

dyr = vapor temperature.

TABLE 7. RESULTS OF VARYING Ca(OH)2 TREATMENT LEVEL

Reference number	Treatment level of Ca(OH),	Acid	Percent by weight Ca(OH) <sub>2</sub> to oil
1830378	Amount to neutralize acid number in 1732591-2ª	0.39	0.04
1830378	5 times amount to neutralize acid number in 1732591-2	<.018	0.2
1830364	12.5 times amount to neutralize acid number in 1732591-2	.032	0.5
1830358	12.5 times amount to neutralize acid number in 1732591-2	.045	0.5
1830352	12.5 times amount to neutralize acid number in 1732591-2	060.	0.5
1830380	Amount to neutralize acid number in 1732588-4 <sup>b</sup>	<.018	0.2
1830380	4 times amount to neutralize acid number in 1732588-4	<.018	9.0
1830380	12.5 times amount to neutralize acid number in 1732588-4	<.018	2.6
1830382	Amount to neutralize acid number in 1732588-3	.40	90.0
1830382	4 times amount to neutralize acid number in 1732588-3	<.018	0.24
1830382	8 times amount to neutralize acid number in 1732588-3	<.018	0.48
1830385	5 times amount to neutralize acid number in 1732588-3	060.	0.3

<sup>a</sup>Main fraction of caustic:MqO/methanol distillation, acid number of 0.39. <sup>b</sup>post fraction of caustic/methanol distillation, acid number of 1.58. <sup>C</sup>Main fraction of caustic/methanol distillation acid number of 0.45.

#### 3.6.1 Slurry Treatment

Oil samples from the main distillate of the used oil distilled with NaOH/methanol (RF N/M = reformulation fraction NaOH/methanol distillation), were slurried at two different temperatures with eight adsorbents at 10% by weight to oil. Samples were also slurried with Ca(OH)<sub>2</sub> and magnesium oxide (MgO) at various concentrations to maximize lowering of the acid number. Oil slurried with 0.5% by weight Ca(OH)<sub>2</sub> was also slurried with bleaching clays for examination. Table 8 summarizes the adsorbent study scheme. Table 9 summarizes thin layer chromatography (TLC), high performance liquid chromatography (HPLC), and acid number analyses of the distillate treated with bleaching clays at two different temperatures (70°C and 150°C) to determine minimum treatment temperature. The data in Table 9 suggest the use of attapulgus or fuller's clay for additive/degradation product removal at 70°C.

Acid number lowering of the distillate with basic material was examined and the data are shown in Table 10. HPLC and TLC results indicate no differences in additive/degradation product removal by varying the basic material level or between  $Ca(OH)_2$  and MgO. Data indicates use of  $Ca(OH)_2$  as most effective in lowering the acid number.

In another study, the RF N/M distillate was treated with 0.5% by weight  $Ca(OH)_2$  and then slurried with bleaching clays at 70°C. The data shown in Table 11 suggest the use of attapulgus or fuller's clay for treatment.

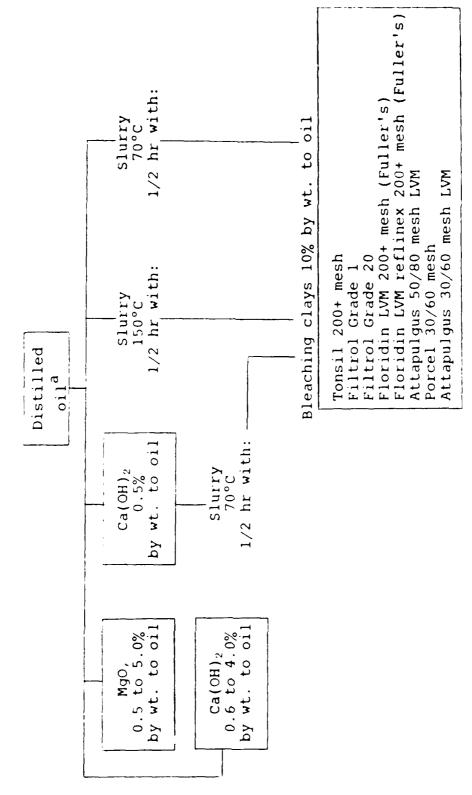
When acid number needs lowering, the data generated by this adsorbent study suggest a slurry at room temperature for 16 hr with the stoichiometric level of caustic to neutralize the acid number, followed by a slurry with fuller's or attapulgus bleaching clays at 70°C for 0.5 hour.

The adsorbent study scheme suggested by the data generated with the RF N/M distillate, which will be used with the distillates from the remaining three scaleup distillations is shown in Table 12.

Data presented in Table 13, for main distilled fraction (MDF) from NaOH/ $H_2O$  in isopropyl alcohol (IPA) distillation, suggest the use of enough  $Ca(OH)_2$  to neutralize the acid number, followed by treatment with fuller's earth. Data in Table 14 for MDF from NaOH/MgO distillation indicate the use of the stoichiometric level of  $Ca(OH)_2$  to neutralize the acid number followed by treatment with either fuller's earth or attapulgus clay.

Data in Table 15 (MDF from no treatment) suggest slurry with the stoichiometric amount of  $Ca(OH)_2$  to neutralize, followed by reslurry with fuller's earth.

TABLE 8. ADSORBENT STUDY SCHEME



<sup>a</sup>Main fraction distillate of NaOH/methanol distillation.

TABLE 9. SUMMARY OF DISTILLATE TREATED WITH VARIOUS BLEACHING CLAYS

	Acid	No.
Bleaching Clays	Slurry heated	Slurry heated
10% by wt. to Oil	to 70°C	to 150°C
Tonsil 200+ mesh	0.17	0.16
Filtrol Grade 1 200+ mesh	0.24	1.77
Filtrol Grade 20 200+ mesh	0.25	2.26
Floridin 200+ mesh (Fullers)	0.14	0.09
Floridin reflinex (Fullers)	0.10	0.10
200+ mesh		
Attapulgus 50/80 mesh LVM	0.17	0.10
Porcel 30/60 mesh	0.10	0.15
Attapulgus 30/60 mesh LVM	0.11	0.18

## Thin Layer Chromatography

Slurry	heated to 70°C		Slurry heated to 150°C	
Decreasing removal of material	Filtrol 1 Filtrol 20 Attapulgus 30/60	+	Tonsil Floridin reflinex Attapulgus 30/60	+
	Tonsil All Fullers Attapulgus 50/80	0	Floridin 200+ Attapulgus 50/80 Filtrol 1	U
<b>\undersigned</b>	Porcel	-	Filtrol 20	0-

Clays rated + at 150°C are slightly better than clays rated + at 70°C.

# High Pressure Liquid Chromatography

Slurry heated to 70°C	Slurry heated to 150°C
Decreasing Filtrol 1 removal of Filtrol 20 -	Attapulgus 50/80 + Porcel 4 Clays 0 Filtrol 1 Filtrol 20 -

Comparison of clays rated + at 150°C are equivalent to clays rated 0 at 70°C.

<sup>&</sup>lt;sup>a</sup>Main fraction from NaOH/methanol distillation.

b Subjective test, with + meaning the greatest removal of material.

TABLE 10. USE OF BASIC MATERIAL TO LOWER ACID NUMBER AFTER DISTILLATION

 $Ca(OH)_2$  as a base MgO as a base Resulting acid Resulting acid Level of number of the Level of number of the treated oil treated oil addition addition 0.48 0.02 Stoichiometric amount Stoichiometric amount for 0.0 acid number for 0.0 acid number (0.06% by wt. to oil)(3.0% by wt. to oil)0.5% by wt. to oil 0.06 0.5% by wt. to oil 0.35 2.0% by wt. to oil 0.02 2.0% by wt. to oil 0.14 4.0% by wt. to oil 4.0% by wt. to oil 0.10 0.03

TABLE 11. SUMMARY OF DISTILLATES SLURRIED WITH 0.5% BY WEIGHT Ca(OH)<sub>2</sub> AND THEN WITH VARIOUS BLEACHING CLAYS

Bleaching Clays Slurried at 70°C	Acid No.
No Bleaching Clay treatment	.0.09
Porcel	< 0.09
Attapulgus 30/60	-0.09
Attapulgus 50/80	0.00
Floridin (Fullers) 200+	< 0.09
Floridin reflinex	< 0.09
Tonsil	<0.09
Filtrol Grade 1	<0.08
Filtrol Grade 20	< 0.09

#### High Pressure Liquid Chromatography

Decreusing Attapulgus 50/80
removal of Attapulgus 30/60
material Floridin 200+ (Fullers)

Remaining clays equal to or worse than oil before treatment.

#### Thin Layer Chromatography

Tonsil
Floridin reflinex
Floridin 200+ mesh
Attapulgus 30/60
Attapulgus 50/80
Filtrol Grade 1
Filtrol Grade 20
Porcel

<sup>&</sup>lt;sup>a</sup>Slurry for 16 hours at room temperature; main fraction from NaOH/methanol distillation.

bOriginal acid number before treatment: 0.52.

aMain fraction from NaOH/methanol distillation.

bSubjective test, with + meaning greatest removal of material.

ADSORBENT STUDY SCHEME TO USE WITH DISTILLATES OF THE REMAINING 3 SCALE-UP DISTILLATIONS TABLE 12.

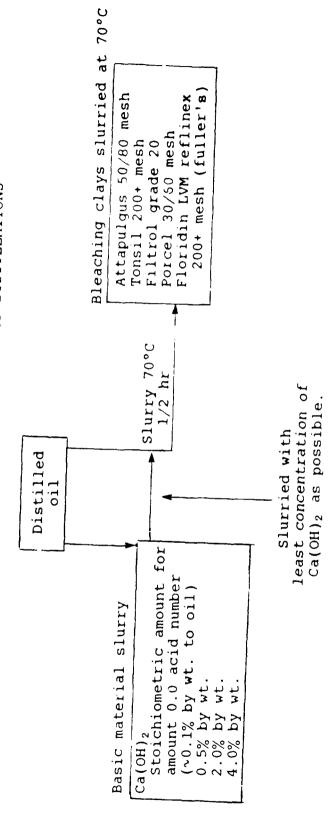


TABLE 13. SUMMARY OF MAIN FRACTION DISTILLATE FROM NaCH/WATER: ISOPROPYL ALCOHOL DISTILLATION TREATED WITH VARIOUS BLEACHING CLAYS

Bleaching clays 10% by weight slurried at 70°C	Acid number oil treated with clay	Acid number of oil treated with 0.5% by weight Ca(OH) <sub>2</sub> and then with clay
No clay treatment	2.18	ု <u>(</u> ) . မ <b>စ်</b>
Tonsil	1.47	·0.09 · 0.18
Attapulgas	1.66	.0.09
Fullers (Floridin-reflinex)	1.18	<0.09
Filtrol 20	1.83	< 0.09

Lowering of acid number with Ca(OH) <sub>2</sub>	Acid number
Quantity to neutralize acid no. $(0.28\%)$	<0.09
0.5% by weight $Ca(OH)_2$	<0.06
2.0% by weight $Ca(OH)_2$	<0.09

Mo difference in removal by changing  ${\rm Ca}({\rm OH})_2$  treatment level by HPLC and TLC.

#### Thin Layer Chromatography

Oil	+ adsorbents	$Oil + Ca(OH)_2 + adsorbents$
Decreasing removal of material	Filtrol 20 Tonsil Attapulgas Fullers (reflinex)	Tonsil Fullers (reflinex) Filtrol 20 Attapulgas

 $Ca(OH)_{\circ}$  + bleaching clays better than bleaching clays only.

#### High Pressure Liquid Chromatography

Oil -	+ adsorbents	Oil + $Ca(OH)_2$ + adsorbents
Decreasing removal of material	Attapulgas Tonsil Fullers (reflinex) Filtrol 20	Tonsil Fullers (reflinex) Filtrol 20 Attapulgas

 $Ca(OH)_2$  + Tonsil better than attapulgas alone.

TABLE 14. SUMMARY OF MAIN FRACTION DISTILLATE FROM 75/25 NaOH/MgO IN METHANOL, DISTILLATION TREATED WITH VARIOUS BLEACHING CLAYS

Bleaching clays 10% by weight slurried at 70°C	Acid number oil treated with clay	Acid number of oil treated with 0.5% by weight Ca(OH)2 and then with clay
No clay treatment	0.35	<0.08
Tonsil	0.12	<0.09
Filtrol 20	0.18	0.11
Fullers (reflinex)	0.09	<0.09
Attapulgas	0.13	<0.09

Lowering of acid number with Ca(OH)2	Acid number
Quantity to neutralize acid no. (0.042%)	<0.080
0.5% by weight	<0.080
2.0% by weight	<0.085

No difference in removal by changing  $Ca(OH)_2$  treatment level by HPLC and TLC.

Thin Layer Chromatography a

011 -	+ adsorbents	Cil + Ca(OH) <sub>2</sub> + adsorbents	-
	Tonsil	Tonsil +	
Decreasing	Tonsil Attapulgas Fullers (reflinex)	Tonsil + Fullers + Attapulgas 0  ▼ Filtrol 20 -	
removal of	Fullers (reflinex)	Attapulgas 0	
material '	Filtrol 20 -	♦ Filtrol 20 -	

Treatment with  $Ca(OH)_2$  + clay better than clay alone.

## High Pressure Liquid Chromatography a

Oil 4	adsorbents	Cil + Ca(OH) <sub>2</sub> + adsorbents
Decreasing removal of material	Fullers (reflinex) Tonsil Attapulgas Filtrol 20	Fullers (reflinex) Attapulgas fonsil Filtrol 20

Treatment with  $Ca(OH)_2 \rightarrow clay$  slightly better than clay alone.

<sup>&</sup>lt;sup>a</sup>Subjective test, with + meaning greatest removal of material.

TABLE 15. SUMMARY OF MAIN FRACTION DISTILLATE FROM DISTILLATION TREATED WITH NO BASIC MATERIAL, TREATED WITH VARIOUS BLEACHING CLAYS

Bleaching clays 10% by weight slurried at 70°C	Acid number oil treated with clay	Acid number of oil treated with 3% by weight $Ca(OH)_2$ and then with clay
No bleaching clay treatment	1.84	0.17
		- · - ·
Tonsil 200+ mesh	1.24	0.24
Porcel	0.60	0.22
Fullers (Floridin-reflinex)	0.73	0.22
Attapulgas 50/80	1.40	0.17
Filtrol grade 20	1.43	0.17

Lowering of acid number with $Ca(OH)_2$	Acid number
0.5% by weight	0.22
2% by weight	0.17
Quantity to neutralize acid no. (2.1%)	0.17
3% by weight	0.17
4% by weight	0.16

No difference in removal by changing  ${\rm Ca(OH)}_2$  treatment level by HPLC and TLC.

Thin Layer Chromatography

oil	+ adsorbents	····	$0il + Ca(0H)_2 + adsorbents$	
Decreasing removal of material	Attapulgas	+ 0+ 0	Tonsil Filtrol grade 20 Fullers (reflinex) Attapulgas Porcel	+ O+ O-

 $0il + Ca(0H)_2 + adsorbents$  better than oil + adsorbents.

## High Pressure Liquid Chromatography

0il + a	dsorbents	0	$Oil + Ca(OH)_2 + adsorbents$
Decreasing removal of material	Tonsil Fullers Attapulgas Filtrol 20 Porcel	* >0 >-	Fullers Tonsil Filtrol 20 Attapulgas Porcel

Oil +  $Ca(OH)_2$  + adsorbents = oil + adsorbents.

<sup>&</sup>lt;sup>a</sup>Subjective test, with + meaning greatest removal of material.

A comparison of acid number, HPLC, and TLC data for the reclaimed oil from the four large-scale distillations suggests the following decreasing order of effectiveness for the additive/degradation removal processes:

- NaOH/MeOH Ca(OH)<sub>2</sub> slurry attapulgus clay slurry
- NaOH:MgO/MeOH Ca(OH)<sub>2</sub> slurry fuller's or attapulgus clay slurry
- 3. Either NaOH/H<sub>2</sub>O:IPA Ca(OH)<sub>2</sub> slurry fuller's earth slurry, or use of no basic material - Ca(OH)<sub>2</sub> slurry fuller's earth slurry

We selected NaOH/MeOH -  $Ca(OH)_2$  slurry - attapulgus clay slurry as a viable reclamation procedure on which to base our continuing studies.

Previous studies had shown that attapulgus clay slurried in oil at approximately 70°C was the ideal treatment for additive/degradation product removal. We further defined the optimum slurry temperature through the use of HPLC, TLC, and acid number. Results are presented in Table 16. The data show a wide range of temperatures that could be used, but to keep energy costs low and oil viscosity down (to aid in filtration), we chose to clay treat at 50°C for 30 min to 1 hr.

The attapulgus clay-to-oil ratio were determined individually for each batch processed in the pilot plant; these results were then used to determine a general clay-to-oil ratio.

TABLE 16. VARYING ATTAPULGUS CLAY TREATMENT TEMPERATURE

Conditions (slurry			
for 30 min)	Acid number	_TLC <sup>a</sup>	HPLCa
40 40505		0+	
90-105°C	<0.018	O	O
70-77°C	<0.018	0	0
50-55°C	<0.018	0	0
30-40°C	<0.018	0	0

1830373 - Main fraction NaOH:MgO/methanol distillate treated with Ca(OH)<sub>2</sub> and 10% by weight clay to oil.

<sup>a</sup>Subjective test, with + meaning the greatest removal of material.

#### 3.6.2 Chromatographic Adsorbent Studies

We examined the use of a column (93 cm x 1 cm) packed with 50/80 mesh attapulgus clay heated from  $70-100^{\circ}\text{C}$  for comparison with the 10% by weight clay-to-oil slurry procedure. The MDF from NaOH/MeOH treated with 0.5% by weight  $\text{Ca}(O\text{H})_2$  was used, to provide a direct comparison with the slurry studies. Acid number data presented in Figure 49 for both procedures show that the slurry treatment is more efficient; HPLC and TLC data bear out this conclusion.

We extended this study by varying column temperature and oil retention time. Figures 50, 51, and 52 present sample acid number vs. total volume through column by varying column conditions. The data suggest that a ≤7.5-min retention time and a column temperature of 65-80°C will treat (determined by HPLC and TLC) a greater volume of oil than a slurry of 10% by weight clay to oil. Nevertheless, the 10% by weight clay presently used for slurrying is probably an excess, and column adsorption is not as attractive as the data suggest because of the extra work involved in packing columns on a large scale and the need for high pressure pumps. We will therefore use the slurry adsorption procedure in our reclamation process.

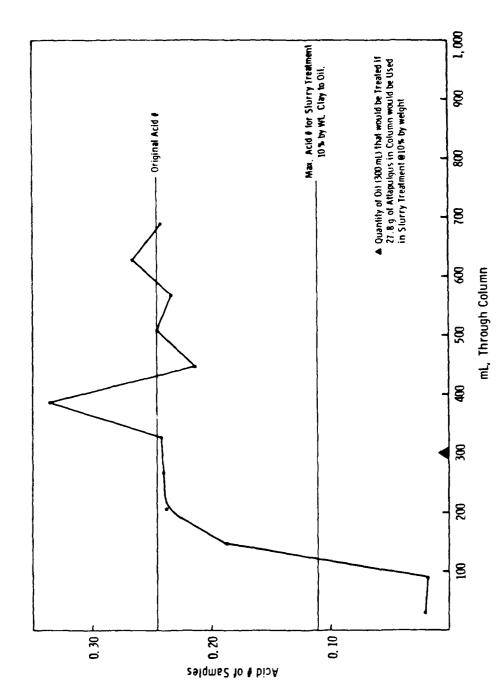
#### 3.7 USE OF ACTIVATED CHARCOAL AS AN ADSORBENT

The original reclamation process included an activated charcoal treatment step of questionable value. We again examined the use of charcoal in this program. A sample of a distillate treated with Ca(OH)<sub>2</sub> and attapulgus clay was treated with 6% by weight charcoal to oil for 30 min at 70°C and examined by HPLC and TLC. The only perceptible change to the oil was a very slight color improvement, which may not improve the total reclamation process.

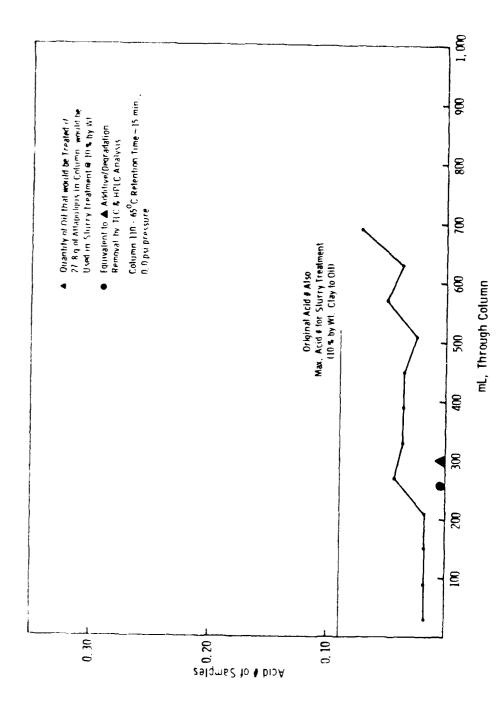
#### 3.8 RECLAIMED BASE STOCK EVALUATION

In the earlier work on this contract we concentrated on identifying a distillation procedure, selecting a clay adsorbent for additive/degradation removal with a treatment procedure, and identifying the level of Ca(OH)<sub>2</sub> treatment required to lower the acid number. The next step was to reclaim a small quantity of oil with our modified procedure and reformulate it with additives to perform a few selective tests to verify our overall process. Table 17 presents our reclaimed oil test scheme.

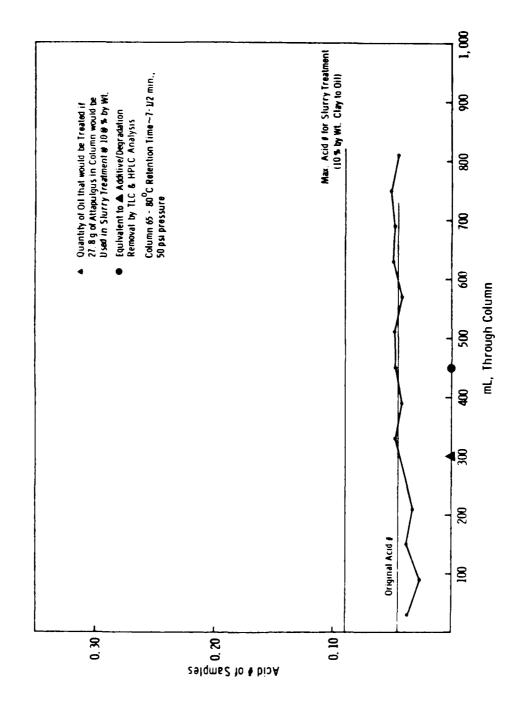
To determine if precut, main cut, and postcut distillates could be combined as reclaimed base stock, distillates from NaOH/methanol distillation were combined, reformulated and foam tested. Table 18 lists the resulting foam test volumes, which suggest that it may be possible to combine all fractions for recovery. To further verify the reclamation process and the idea of combining the distillate



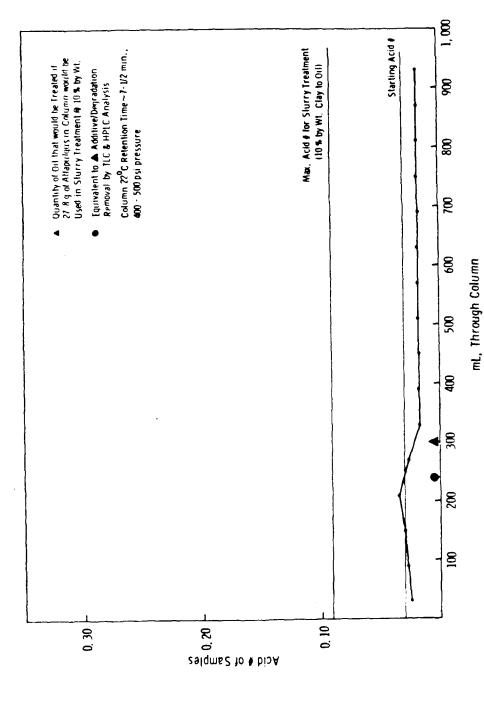
Main fraction from NaOH/MeOH distillation, pretreated with Ca(OH)2, followed by chromatographic treatment with attapulgus clay. Figure 49.



Main fraction from NaOH:MgO/MeOH distillation, pretreated with Ca(OH)<sub>2</sub>, followed by chromatographic treatment with attapulgus clay. Figure 50.



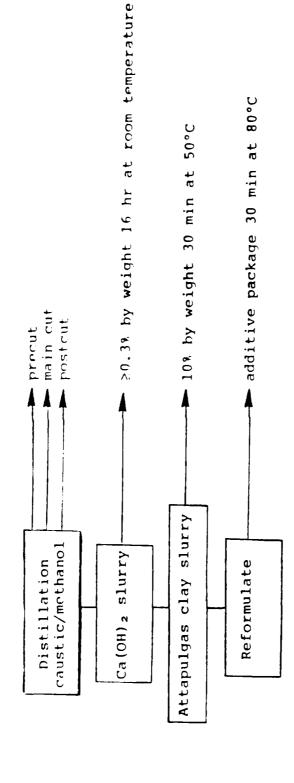
Main fraction from NaOH:MgO/MeOH distillation, pretreated with Ca(OH)2, followed by chromatographic treatment with attapulgus clay. Figure 51.



Main fraction from NaOH:MgO/MeOH distillation, pretreated with Ca(OH)2, followed by chromatographic treatment with attapulgus clay. Figure 52.

RECLAIMED OIL TEST SCHEME TABLE 17.

1



- main cut Foam test 3 formulated samples

main cut + appropriate percent of precut main cut + appropriate percent of pre- and pre- and

pre- and postcut

main cut + precut
main cut + precut and postcut Alcor testing of 2-2.6 liter samples - main cut

# Tests:

Silver and bronze corrosion, 50 hr @ 450°F @ 392°F Viscosities at 210°F and -65°F Corrosion and oxidation 48 hr Lead corrosion, 1 hr @ 325°F Deposition number test Static 'Jaming Acid number

TABLE 18. REFORMULATED BASE STOCK FOAM VOLUMES

Sample	Distilled oil sample numbers	Foam volume
1	Reclaimed 1732588-3 (main cut)	10 mL
2	Reclaimed 1732588-3+ 6% (1732588-2) (precut)	15 mL
3	Sample 2 reformulated	15 mL
4	Sample 3 + reformulated 1732588-4 (postcut)	15 mL

fractions, two 2.6-liter samples had been reclaimed. They were reformulated with additives and sent to Alcor Testing Laboratory for tests outlined in our reclaimed oil test scheme.

The two reclaimed samples of combined NaOH/methanol distillates had the vapor temperature ranges shown in Table 19.

TABLE 19. VAPOR TEMPERATURE RANGES OF COLLECTED DISTILLATES IN THE ALCOR TESTED SAMPLES

	Vapor
	temperature
Sample	range
1830396-C/M-23	135 <b>-</b> 285°C
1830397-C/M-234	135-305°C

The results of the selected MIL-L-7808H tests by Alcor are included in Appendix B.

In general, the reclaimed/reformulated (RR) samples performed as well as the formulated virgin base stocks from various manufacturers used earlier to verify the additive package. We have shown by experimentation in the laboratory that the 100-120 mL foam volumes of the RR samples were due to inadvertent addition of excess (3%) PANA additive. In the corrosion and oxidation stability test, the RR samples do not meet the MIL-L-7808H requirements with respect to percent change in viscosity and total acid number change. However, this is attributed to the probable presence of MIL-L-7808G oils in the samples which do not meet these requirements even as new oils. This deficiency could be corrected by dilution with a MIL-L-7808 virgin base stock if necessary. In conclusion, we feel that the developed reclamation process is viable.

#### 3.9 LARGE SCALE RECLAMATION - PILOT PLANT SCALE

In this section we'll discuss scale-up of the reclamation process from laboratory to pilot plant scale. Also included in this section are laboratory studies that were initiated as difficulties arose.

#### 3.9.1 Pilot Plant Distillations (25 gallon scale)

Figure 53 shows a schematic diagram of the distillation set-up in the plant. The equipment was cleaned with soap, water, acetone and a chlorinated solvent. Virgin basestock was distilled in the equipment for final clean out. After each distillation, the still was cleaned with acetone, water and then acetone to remove still bottoms and NaOH. The initial batches distilled encountered problems with bump-over, so we decreased the distillation rate. Appendix G and I contains the data for each batch and procedure sheets for a typical batch distilled. No problems were encountered with the 0-79- series of used oils, though some were distilled 2 times, due to foaming caused by Ca(OH)<sub>2</sub> treatment. Distillates from the 0-82- series had considerable problems with high foam test volumes.

#### 3.9.1.1 Distillate Foam Test Failure

Foam test failure of distillates from the 0-82- series of used oil batches have been uncorrectable. Table I-1 (page 227) lists data from those distillations. The data compared with earlier 0-79- used oil samples processed, are considerably different. There is significantly more still bottoms and still toppings/pre-cut material The earlier cuts in the distillations contain kerosene types of material.

Table 20 presents data from a foam test study involving addition of JP-4 jet fuel and hydraulic fluid to a low foaming formulated oil. The data suggests that the used oils maybe contaminated by these fluids and their associated additives. Treatment of the distillates with  $Ba(OH)_2 \cdot H_2O$  and fullers earth did not lower the foam test volume.

Metal analysis of these oils, indicate that used oil 0-82-6 maybe contaminated with motor oil. This oil has very high levels of lead (25 ppm) and zinc (76 ppm), not normally, found in 7808H oils.

Use of HPLC, TLC and IR as a screening method for contamination was unsuccessful. Appendix J contains the chromatograms for HPLC, GC and IR. As can be seen the HPLC and IR resemble the other used oil samples, the GC chromatograms do show a significant difference. Comparison of chromatograms (180-350°C program) with the 0-79- used oils show additional peaks buried in the initial solvent peak. Comparing the chromatograms with their distillates, show the removal of the extra peaks. The GC program was changed to 100-350°C for better separation of the peaks buried under the initial solvent peak, these chromatogram are also in Appendix J. Also included are samples of JP-4 and hydraulic fluids. GC appears to be a good way to screen each used oil sample before processing.

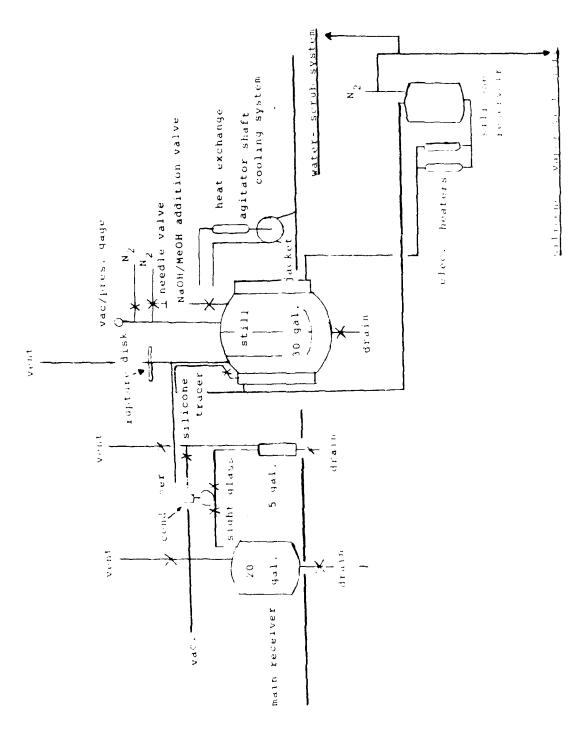
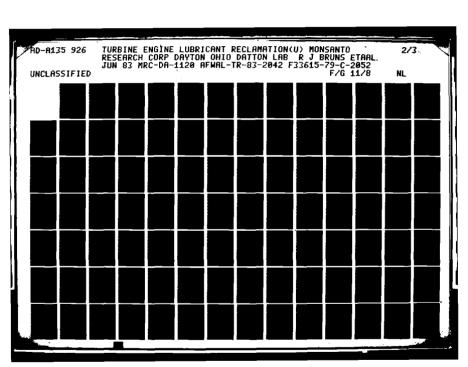


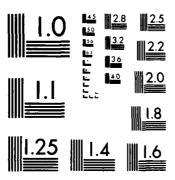
Figure 53. Plant distillation equipment.

TABLE 20. FOAM TEST RESULTS WITH JP-4 AND HYDRAULIC FLUIDS2

NB#	Treated samples	_	test mL treatment	Foam test mL after treatment
2776315A	2% JP-4		50	50
2776315B	Extra JP-4 additives (2%)		50	50
2776315C <sup>1</sup>	Oil from 2276315B heated to $320^{\circ}\text{C}$ for 7 hr.		50	225
2276316A	Ten drcps each of 2 different hydraulic fluids - ester based.		45	50
2276316B	2276316A plus additional 10 drops each of the hydraulic fluids.		50	9.,

Simulate distillation conditions.
 In 200 mL of low foaming formulated Emery basestock.





MICROCOPY RESOLUTION TEST CHART
NATIONAL BUREAU OF STANDARDS 1963-A

Used oil 0-79-2 was also unreclaimable, a GC chromatogram, Figure J-20, does not indicate any contamination. Since the used oil is only one out of 15 that does not pass the foam test the sample in a plant processing situation could easily be diluted with others.

Used oil 0-79-2 was also distilled without caustic (stoichiometric amount to neutralize the acid no.) to show that caustic does indeed lower, or decrease the acid number compared to no treatment. The acid number of the caustic treated oil is almost 1/2 that of the non-treated oil. It appears that the most dramatic effects of acid number lowering occurs on oils having starting acid numbers of 0.75 to 1.0 or greater (see Appendix I).

Also included in Appendix J are the IR and HPLC chromatograms of JP-4 and the hydraulic fluids.

#### 3.9.2 Use of Basic Material to Lower Acid No.

Figure 54 shows a schematic diagram of the adsorbent treatment in the plant. The equipment was prepared similar to the distillation equipment. The cotton filter bag (for the bag filter) was washed 2 times and rinsed several times, with the last 3 rinses being deionized water.

Initially we used  $Ca(OH)_2$  to lower the acid number and started getting inconsistant and poor foam test results. The problem was traced to contamination from the equipment and also,  $Ca(OH)_2$  itself was found to cause foaming. Treatment with attapulgas clay was not beneficial. Lab studies, using  $Ba(OH)_2 \cdot H_2O$  in place of  $Ca(OH)_2$  to lower the acid number in our reclamation process, were consistently successful. The use of  $Ba(OH)_2 \cdot H_2O$  resulted in 0.00 acid numbers compared to 0.02 to 0.2 with  $CA(OH)_2$  treatment. The foam test results after  $Ba(OH)_2 \cdot H_2O$  treatment were comparable to foam volumes prior to treatment. We determined that the minimum level of  $Ba(OH)_2 \cdot H_2O$  for treatment was 0.77% by weight or the stoichiometric amount to neutralize the acid number.

Most oils after  $Ba(OH)_2 \cdot H_2O$  [also  $Ca(OH)_2$ ] treatment had after 1 day of standing, turned from a crystal clear to a cloudy liquid with some precipitate. By raising our treatment temperature from  $20\,^{\circ}\text{C}$  to  $50\,^{\circ}\text{C}$  we were able to increase the precipitation (ppt) rate so it could be removed along with the  $Ba(OH)_2 \cdot H_2O$  during filtration. The acid number remained at zero after increased temperature treatment.

The earlier processed oils that had developed high foam test volumes due to  $Ca(OH)_2$  treatment and equipment contamination were successfully lowered by redistillation and treatment with  $Ba(OH)_2 \cdot H_2O$ .

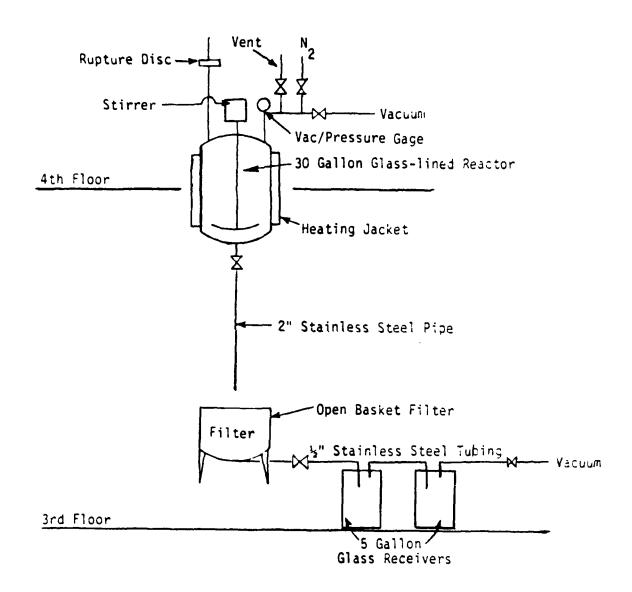


Figure 54. Equipment design for  $Ba(OH)_2 \cdot H_2O$  adsorbent treatment in Pilot Plant.

Two reclaimed basestocks 0-79-7 and -14 and to some extent 0-79-12 and-10 had developed precipitate after a couple of months storage prior to reformulation. Various samples were prepared from 0-79-14 to solve the problem. Table 21 describes the various treatments and their results. In the plant we first filtered the batches and then heated them to 60°C prior to reformulation.

TABLE 21. SAMPLES PREPARED TO SOLVE PRECIPITATE PROBLEMS IN RECLAIMED BASESTOCK 0-79-14 (1997651)

Sample No.	******	<u> Preatment</u>	Storage	results*
2000251A	Control - only filter	ed		ppt -
2000251B	Held at 50°C for 16 he	ours/filter		No ppt
2000251C	Held at 60°C for 16 he	ours/filter		No ppt
2000251D	50°C for 16 hours with	h Ba(OH) $_2 \cdot H_2 O/f$	ilter	ppt
2000251E	60°C for 16 hours with	h Ba(OH) $_2 \cdot H_2$ O/f	ilter	ppt
2000251F	Fullers earth @ 50°C	for 1/2 hour/fi	lter	ppt

<sup>\*</sup>Samples stored for approximately 1-1/2 months, except F-only 1 month.

#### 3.9.3 Adsorbent Treatment

Two used oils 0-79-11 and 0-79-9 (first group sent out for testing) were formulated without attapulgas clay treatment, sent to Alcor Testing Labs for MIL-L-7808H testing. It was possible that we would be able to exclude clay treatment from the process. Originally the clay was used to remove trace metals and surfactants, but trace metals are lowered through distillation and  $Ba(OH)_2 \cdot H_2O$  treatment. Attapulgus clay also has been shown not to remove troublesome surfactants in the oil. Later it was found that the clay also threw the accelerated storage stability test out of specifications. The results of the MIL-L-7808H testing indicated no need for clay treatment, so none of the reclaimed oils in the plant were clay treated.

We also had looked at ion exchange resins in the lab [Rexyn 101(H), Amberlyst 15, and Amberlite 200] to remove surfactants that may have caused foaming. However, no beneficial effects were found.

#### 3.9.4 Reformulation

Figure 55 shows a schematic diagram of the reformulation equipment in the plant. The equipment was prepared similar to the distillation equipment. The Filterite inline cotton filter was washed with 5 gallon of virgin basestock. The first quantity of virgin

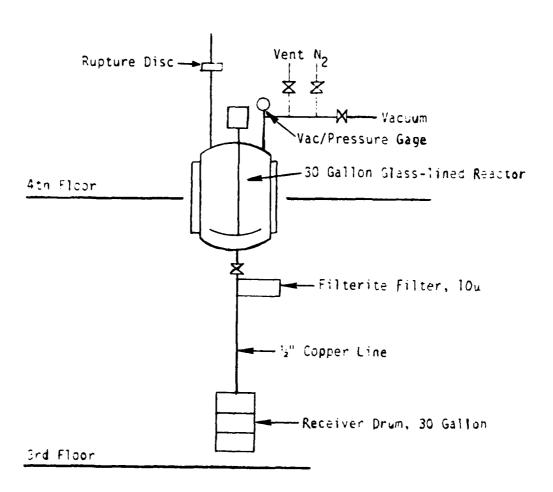


Figure 55. Equipment design for oil reformulation.

oil through the filter had a high foam test volume, while a later sample had a low volume. The unlined steel drums were washed with acetone and dried with a nitrogen stream.

Each reclaimed basestock was analyzed prior to reformulation by HPLC for DODPA, PANA, and TCP, major additives in the 7808H oils. Appendix C lists the analytical conditions for the analysis. The major additive levels in the oils were taken into consideration for the reformulation. Table 22 lists the additive and their level of addition. The lower level additives are assumed at a low concentration in the reclaimed oil.

Prior to reformulating a batch in the plant, we did a reformulation in the laboratory for foam testing and acid number determination. We found that there was a correlation between the constituent level of the major additives and foam test level. This suggested that TCP maybe interacting with PANA to form an amide which is known to lower surface tension and act as a surfactant. We finally did reformulation studies on each batch to maintain a low foam volume. Tables 23 and 24 bring out the difficulty in balancing different additive levels. This difficulty is a result of not knowing all the additives and their levels in the oil. We finally maintained a 2% antioxidant level with DODPA (no PANA addition) and did not necessarily add TCP. By this time we had some test results back on load bearing tests and had satisfactory results at lower levels of TCP.

We found it unnecessary to add virgin base stock for viscosity correction to any batch.

In the plant the additives were dissolved at 80°C/1 hr.

#### 3.9.5 MIL-L-7808H Test Results

The 7808H test results for the 10 batches of used oil reclaimed in the pilot plant are in Appendix E. Table 25 summarizes the major problems encountered with these batches.

Initially foaming results were high. High test volumes were traced to major additive levels as discussed in Section 3.9.4.

TABLE 22. ADDITIVE LEVELS ADDED TO THE RECLAIMED BASESTOCKS

Additives	Percent by weight
Tricresyl phosphate (TCP)	2.0
4,4'-Dioctyldiphenylamine (DODPA)	1.0
Phenyl-α-naphthylamine (PANA)	1.0
Benzotriazole	0.1
Triphenyl phosphite	0.1
Quinizarin	0.05
Antioxidant 703	0.2

FOAM TEST RESULTS OF REFORMULATION STUDIES TABLE 23.

THE STATE OF THE S

Table   Tabl																•	
Harabara   Harabara	-			36-06	200003	147)				-0	74-8 (2	000248)			1997	28°5	1397650
1.32 0 0 0.66 0 0.4 0.8 1.0 0.4 0.4 0.4 0.4 0.4 0.4 0.4 1.0 1.0 1.0 1.0 1.3 0.2 0.2 0.2 0.2 0.5 0.5 0.5 0.5 0.2 0.2 0.2 0.2 0.2 0.2 0.2 0.2 0.2 0.2							l	e,	=	#2	#3	#4	#2g	#6	=	<b>\$</b>	=
1.32 0 0 0.66 0 0.44 0.8 1.04 1.04 1.04 1.05 1.04 1.04 1.04 1.04 1.04 1.04 1.04 1.04	Percent		2	2			Ì.			4	4	G	4.0	1.0	1.0	9.1	7.0
0.2         0.2         0.2         0.5         0.5         0.26         0.26         0.26         0.26         0.26         0.26         0.26         0.26         0.49         0.27         0.27         0.27           0.1         0.2         0.2         0.2 <th>TCP</th> <td>1.32</td> <td>0</td> <td>0</td> <td>9.0</td> <td>9</td> <td><del>*</del>.</td> <td></td> <td>?</td> <td>•</td> <td>:</td> <td>,</td> <td></td> <td></td> <td>•</td> <td></td> <td></td>	TCP	1.32	0	0	9.0	9	<del>*</del> .		?	•	:	,			•		
1e         0.3         0.3         0.3         0.23         0.1	DODPA	0.5	0.2	0.2	0.2	5.0	. 0	0.5	0.26	0.26	0.26	0.49	0.49		1.24	1.24	າ. -
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. 703 0.2 1.5 0.2 0.2 0.2 0.2 0.2 0.2 0.1 0.2 0.2 0.2 0.2 0.2 0.2 0.2 0.2 0.2 0.2	Tree contracts		0.05	0.05	0.05	0.05	0.05	0.05	0.05	0.05		0.05	0.05	0.05			
>200 45 20 70 10 15 65 65 80 80 10 20	Antioxidant 703		1.5	0.2	0.2	0.2	0.2	0.2	0.2	0.1	0.2	0.2	0.2	0.5			
45 20 /0 10 TO 23 S\$						9	1	3	5.5	9	90	2	20	95			
	Foam test level	>200	45	20	2	3	ا 1	3	١								

<sup>a</sup>Formulation used in pilot plant batch. b\_Level of additives added to samples, % wt to oil wt. Smajor additive level in basestock before reformulation. Number one samples were results of first analysis, number two-results of the second.

FOAM TEST RESULTS OF REFORMULATION STUDIES TABLE 24.

D Jie 11-d	0-74-7 200626.4 Additive level before formula-	e [ #	0-74-14 Zuntzak Addatave level before formula-	<u>-</u>	2	~ *	<b>∵</b>	© . 3ac	3*	0-79-12 2006275 Additive level before formula-		2 #	3.34	<del>*</del>	0-79-12 2000273 Additive level before formula- tion	i i	e 7.
4.4	8 0	1.20	00 1	6.0	0	c		0	5.0	1.19	18 0	0	0	4	1.24	0	0
D: DPA	1 32	0.18	1.20	0.3	O	0	o o	0.54	0.54	1.29	0.21	0.21	0.35	0.35	1.46	0	0 24
PENA	0.27	0.23	0.26	0.24	0	0 74	ε	э	0	0.36	0.14	0.14	0	0	0.03	0 2	3
E-nZett 120le	٥	7 0	0	1.0	0.1	0	- n	<b>1</b> 0	1.0	0	1.0	0.1	0.1	0.1	0	1.0	0 1
3.41	_	0.1	0	0.1	0.1	1.0	0.1	- 0	1.0	•	0.1	0.1	0.1	0.1	0	0 1	0
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furnished as overlaw Polot Flant batch. There is an effective object to sample of with only of  ${\bf M}$ 

TABLE 25. SUMMARY OF MIL-L-7808H TEST PROBLEMS

	Test groups	Treatments	Problems
	1st Group of 2		
	1997659 (0-79-9) 1997658 (0-79-11)	Distilled Ba(OH) <sub>2</sub> ·H <sub>2</sub> O	Some foaming - traced to higher level of TCP & PANA, shown in lab to cause foaming
	2nd Group of 2		Elastomer results slightly off
	1997695 (0-79-6) 1997693 (0-79-13)	Distilled $Ba(OH)_2 \cdot H_2O$	Foaming - OK Elastomer results are better (lower level of TCP)
		Attapulgus clay Also 0-79-13 has 2 x's level of guinizarin	Accelerates storage stability test ~288 normal additive level - ~190 with 2 times quinizarin
91	Group of 6-1 liter3		
l	Sample for accelerated storage	Assorted	Attapulsus clay appears to adversely increase accelerated storage test results
	2000247 (0-79-15) <sup>1</sup> 2000248 (0-79-8) <sup>1</sup>	Distilled Ba(OH)2·H2O	Some static foam problems
	4th Group of 22		
	2000267 (0-79-7) 2000268 (0-79-14) <sup>1</sup>	Distilled Ba(OH) $_2\cdot H_2O$	Accelerated storage stability test for 0-79-14 Both batches had heavy precipitate prior to reformulation
	5th Group of 22		
	2000275 (0-79-12) <sup>1</sup> 2000273 (0-79-10)	Distilled Ba(OH) <sub>2</sub> ·H <sub>2</sub> O	Both batches had light precipitate prior to reformulation
	$^{1}$ Oil distilled two times. $^{2}$ Contains two times the amount of Ethyl 703 $^{3}$ See Table 26.	es. le amount of Ethyl 703.	

Some elastomer results were high on the earlier batches. In the later batches the TCP level was lowered and there was a significant improvement.

The corrosion and oxidation results of some batches at 96 hours are higher than expected. The data is similar to results from a diester basestock provided by Rohm and Haas. MIL-L-7808G (di & triester) oils would not be expected to have the same stability as the MIL-L-7808H oils.

The accelerated storage stability results had been inconsistent throughout this program. It was learned after the completion of the work phase, that miscalculations by an outside testing laboratory had resulted in erroneously high test results. These results led to studies to improve the storage results. Table 25 displays the various treatments and results of the reclaimed batches. Also, we did a modified accelerated storage test of a sample consisting of 3 of the 5 formulated virgin basestocks used to verify MRC's additive package earlier in this program. The results are listed below:

Accelerated storage test on sample 1 1/2 years in storage	Specification	Results
0.0 hr 168 hr	150 max.	0.3 69.9

We also worked up 6 samples by various treatments for accelerated storage testing. Table 26 summarizes the treatments and results of these samples. The samples indicate that the attapulgus clay we had used to treat the 2nd group (Table 25) in the pilot plant may have been instrumental in increasing the accelerated storage test results.

Still working with erroneous test results, we then processed the remaining batches in the plant with no clay treatment and doubled the ethyl antioxidant 703 additive level. We then went back at the end of the program and added additional antioxidant 703 (to bring the level to 0.20%) to the attapulgus clay treated batches (see Table 25) to correct their storage results. Table 27 contains the results of retesting. As can be seen, the test results were improved.

We conclude from the revised data, the attapulgus clay treated batches would have passed the accelerated storage test without clay treatment, we then have 90% passing on the test.

TABLE 26. ACCELERATED STORAGE TEST SAMPLES/RESULTS

			esults
NBP	Treatment	25 max. 48 hrs.	150 max. 168 hrs.
2000227A	Treated with attapulgus clay pre- viously dried @ 500°C, formulated with standard additive package.	2.1	42.1
2000227B	Treated with attapulgus clay as received and used in Pilot Plant formulated with standard additive package.	21.7	106.4
2000227C	Treated with Fullers earth previously dried @ 500°C as in earlier contract (AFAPL-TR-78-50), formulated with standard additive package.	8.3	17.6
2000227D	Treated with Fullers earth as received, formulated with standard additive package.	0.4	50.5
2000227E	No clay treatment, 2 times normal level of Ethyl Antioxidant 703 (storage stability additive).	0.5	47.6
2000227F	No clay treatment, 2 times normal level of quinizarin (anti-lead corrosion additive).	3.0	48.1

TABLE 27. ACCELERATED STORAGE STABILITY TEST RESULTS

Addition of Extra Ethyl 703 to Attapulgus
Clay Treated Batches

Comments	NBP	48 hrs	168 hrs
Original	1997693(0-79-13) 1997695(0-79-06)	12.7 57.8	190.7 288.5
10 months later, extra Ethyl 703 added and retested	1997693(0-79-13) 1997695(0-79-06)	3.0 30.6	85.7 136.6

The ten reclaimed oil batches were blended together in pairs for 5 bearing deposition tests, to lower testing costs. Table 28 lists the pairs blended and reasoning for the combinations. Generally all results were within specifications, a few were 4 to 6% over, but we feel no reason for concern (see Appendix E). Groups 4 and 5 had significantly higher total sludge during testing then the other groups. The inlet/outlet screens in the test equipment had a sludge similar to the precipitates formed in the plant. There was no precipitate prior to testing. 90-96% of the sludge was removed on the 1st screen weighings. Examination of plant data on these oils, show that they were hydroxide treated below 45°C. We believe that the sludge formation would have been less if the hydroxide treatment temperature would have been higher.

TABLE 28. BATCH BLENDING FOR BEARING 1 OSITION TESTS

Group	NBP	Used oil	C ents
1	2000248 2000247	(0-79-08) (0-79-15)	Good batches.
2	2000267 2000268	(0-79-07) (0-79-14)	Batches had developed heavy precipitation on sitting prior to reformulation.
3	2000273 2000275	(0-79-10) (0-79-12)	Batches had developed light precipitation on sitting prior to reformulation.
4	1997693 1997695	(0-79-13) (0-79-06)	Poor accelerated storage results; batches were attapulgus clay treated.
5	1997659 1997658	(0-79-09) (0-79-11)	

## 3.9.6 The Optimized Process

The following discussion details the process, as developed and optimized in the pilot plant. The operation discussed follow those outlined in Figure 56, Section 5. The materials used are those specified in Appendix H. Not all the operations discussed were necessary for the oils reclaimed in this program, i.e., phase separation, but they are included to keep the process as general as can be envisioned being required.

It is assumed that the used oils will arrive at the processing plant in 55 gallon drums. A 500 mL sample should be removed. A acid number and a GC/IR analysis should be run on the sample.

A acid no. >10.0 or a unusual GC or IR pattern would constitute rejection of the drum. Next, a lab distillation of the sample, followed by a foam test. If after 2 distillations a sample does not pass the foam test, discard the drum. NaOH will not need to be added to the sample for the lab distillations.

The next step will be to remove solid insoluble materials present (e.g., wear metals, sludges), by pumping through a 100  $\mu$ m cartridge (cotton) type filter, into the still. These materials are present in very low amount, usually less than 0.2% by weight.

The oil is allowed to sit overnight and then a small quantity is removed from the bottom of the still for identification of a insoluble liquid phase. Next a stoichiometric addition of NaOH/methanol (usually 100 g NaOH/½ gallon methanol) solution to neutralize the acid no. (see Appendix F) is added to the still (short path distillation). The oil is distilled at 15 mm. The forerun is collected up to vapor temperature of 135°C and discarded. The main cut is collected from a vapor temperature of 135°C to a pot temperature of 325°C. There is an average of 85% recovery after distillation.

The distillate (main cut) is foam tested and a acid no. determined. A foam test over 20 mL indicate the distillate should be redistilled.

The oil is then treated with  $Ba(OH)_2 \cdot H_2O$  at 55-60°C for 15-18 hrs. The amount of hydroxide added is 0.77% by wt (see Appendix F), unless if a larger amount is necessary to stoichiometrically neutralize the acid no. The oil is filtered to remove the hydroxide; filter aid maybe necessary to prevent a slimy and slow filtering cake. Losses are usually 5%.

At this point the reclaimed base stock should have a acid number of 0.0, but if over 0.1, retreat with the hydroxide. A foam test >35 mL indicates the oil should be redistilled. A value >35 mL at this point would probably result in a >100 mL after reformulation.

Next a HPLC analysis of the oil for major additive levels. A sample should be reformulated in the lab and foam tested. Viscosities should also be determined.

The plant batch can then be reformulated and heated to  $80^{\circ}$ C for 1 hour, filtered through a 10 µm inline cotton filter and pumped into a storage tank. A poor foam test or acid no. >0.3 indicates need for a redistillation. Viscosities out of specs. can be corrected through use of virgin oil.

Another assumption is the reclaimed oil will be kept in a storage tank until results of MIL-L-7808H tests. The oil will then be packaged into sealed containers for delivery.

## 4. DISPOSAL

In this section we'll discuss the disposal of material generated by the reclamation process.

# 4.1 NON-RECLAIMABLE OIL

Oils that do not pass the initial screening or prove at a later time unreclaimable, can be sold to other reclaimers for use as a plasticizer.

# 4.2 STILL BOTTOMS

The majority of the additives distill over, leaving a very low additive level in the still bottoms. The bottoms are similar to materials the asphalt producers use.

# 4.3 STILL TOPPINGS, WASTE ACETONE

Still toppings consisting of toluene, kerosene, low molecular weight oil and other similar material, can be mixed with fuel oil for on site use. The acetone from still clean-outs could also be disposed of, in the same manner.

# 4.4 Ba(OH)<sub>2</sub>·H<sub>2</sub>O/FILTER AID

Filter aid mixed with  $Ba(OH)_2 \cdot H_2O$  will need to be buried in an EPA approved land fill.

# 5. COST ANALYSIS - ENGINEERING STUDY

Two cost analyses were to be performed. One based on a process intended for maximum yield and one for maximum cost effectiveness.

The present process operates at maximum yield and maximum cost effectiveness. The pre-cut range was selected for removal of low boiling contaminants. The still bottom consists of undistillable/unrecoverable material.

The process developed in the earlier contract had included clay and charcoal treatment. These materials are ideal for recycling to lower costs, but has since been found unnecessary. Therefore, only one cost analysis is presented.

A cost analysis of the reclamation process was carried out to estimate the cost of reclaiming used oils in batch size of 1,000, 2,500, 5,000, and 10,000 gallons. The operations involved in the processing are shown in Figure 56.

The capital cost estimates for installation of a process in an existing plant were made using a 1,000 gallon batch size as the basic unit and escalating the costs to other sizes using an escalation factor of 0.6, an average value used frequently in the types of calculation [1]. The standard equation for such types of calculation, where cost is the cost

 $cost = cost_o \left(\frac{s}{s_o}\right)^x$ 

for 1,000 gallon scale, S is the size factor and x the escalation factor, was used. The capital cost data and assumptions are presented in Tables 29 and 30.

Installation costs can vary considerably depending on several factors, e.g., level of instrumentation, amount of piping required, etc. Consequently, the capital costs can vary.

The cost analysis was made based on the assumption that a plant was set up and operating continuously over a year period. This assumes an adequate used oil supply. No cost is included for collection and delivery of used oil to reclamation site.

<sup>[1]</sup> Peters, M. S., and Timmerhaus, K. D., "Plant Design and Economics for Chemical Engineers," 2nd edition, 1968 McGraw Hill Book Company, New York, NY.

Popper, H., editor, "Modern Cost-Engineering Techniques," 1970, McGraw-Hill Book Company, New York, NY.

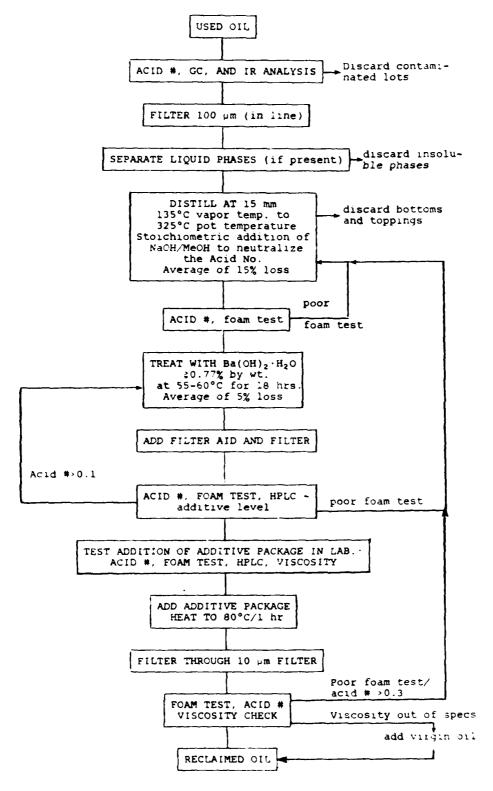


Figure 56. Reclamation process.

TABLE 29. CAPITAL COST DATA
(1,000 gal scale, 1 shift, 40 hr/wk)

# PLANT

	Equipment requirements (1,000gallons)	Cost, \$ est. 1982	Installation,
,	Filtor (bos)	2 500	1 000
1	Filter (bag)	2,500	1,000
1	Gas still	75,000	75,000
1	Vacuum jet	18,000	3,000
2	Jacketed agitator vessel	30,000	120,000
1	Condensor	8,000	_*
1	Receiver	10,000	5,000
1	Bottling equipment	18,000	2,500
28	Product storage tanks	98,000	84,000
		249,500	285,500

<sup>\*</sup>Installation included in with jacketed agitation vessels
TOTAL PLANT \$550,000

LAB

Equipment requirements	Cost, \$ est. 1982
HPLC Auto sampler/GC IR Installation/other equipment	12,000 16,000 12,000 15,000
	55,000

The projected recovery efficiency for this process is 73%, other estimated costs are presented in Table 31, along with cost breakdown for the four batches sizes. The cost assigned to utilities is considered to be a very crude estimate and was derived by calculating the cost of energy required for distillation and multiplying the answer by 5 to cover the costs of the remaining operation cost for filtration, pumping materials, etc.

Not included in the above analysis is the cost incurred if virgin base stock must be added to adjust the viscosity. This is treated here as a separate case since this addition was found not to be necessary in our study.

# TABLE 30. ASSUMPTIONS (1,000 gal. batches)

Waste oil arrives in 55 gal drums

Each drum is tested by GC, IR, acid # determination and lab distillation/foam test

Each plant batch requires 8 hrs to distill and 8 hrs to clean-up and recharge

 $H_{\overline{y}}$  50 g-cal/g

Specific gravity 0.93

2 Month turnaround time for commercial testing ... need 800 gal. storage tanks

40 hr work week - 1 shift

100 batches/yr, 2 days/batch, 20% downtime

\$130,000 labor costs/yr

90% of processed batches will be good

85% recovery on each batch due to distillation

95% recovery on each batch due to Ba(OH)<sub>2</sub>·H<sub>2</sub>O treatment

1,000 gal still can only contain 800 gal of actual oil

15 gal removed out of each batch for testing

56,790 actual gal processed each yr

Capital costs spread out over 3 years

Qualification tests \$5,000 each

Qualified batches are packaged in pt or qt cans

TABLE 31. OPERATING COSTS (\$/gallon of oil)

	1,000 gal scale (56,790 gal/yr)	2,500 gal scale (143,850 gal/yr)	5,000 gal scale (289,200 gal/vr)	10,000 gal scale (579,900 gal vr)
Labor: 1 shift/40 hr wk	2.29	1.04	0.59	0.33
Materials				
Ba(OH)2.H20	$6.72 \times 10^{-2}$			
Filteraid	8.6 x 10 <sup>-3</sup> —			
Additives				
(0.3%) DODPA	5.95 x 10 <sup>-2</sup>			
(0) PANA	•			
(0.5%) TCP	×			
(0.1%) TPP	2.67 x 10 <sup>-2</sup>			
(0.1%) Benzotriazole	×			
(0.05%) Quinizarin	×			
(0.2%) Ethyl 703	×			
Utilities	0.38			•
Capital related costs	0.07	0.07	0.07	0.07
Capital costs	3.55	2.32	1.76	1.33
Disposal $[Ba(OH)_2 \cdot H_2O]$	0.08	0.08	0.08	0.08
Qualification tests	8.80	3.48	1.73	0.86
Miscellaneous	1.00	1.00	1.00	1.00
No. of people	Ŋ	9	7	Ø
Total cost/gallon	16.54	8.74	5.98	4.42
Labor: 3 shifts/40 hr wk	170,370 gal/yr			
No. of people	15			
Total cost/gallon	16.97			

Labor costs split equally between plant and lab.

 $^{
m b}$  Additive levels are based on what actually was added to successfully reclaimed oil.

<sup>c</sup>Maintenance.

d Spread over 3 years.

## 6. CONCLUSIONS

- 1. A feasible process for reclamation of synthetic ester turbine engine oils was developed in an earlier program. The main features of the process consist of distillation to recover base stock, treatment with adsorbents to remove degradation products, and addition of additives to upgrade the oil to meet specifications.
- MRC's proposed additive package has been shown to be satisfactory in five different base stock formulations.
- 3. The use of NaOH during distillation helps lower the acid no., therefore decreasing the amount of  $Ba(OH)_2 \cdot H_2O$  to use in the next step.
- 4. Treatment with  $Ba(OH)_2 \cdot H_2O$  to lower the acid number depends on the following rule of thumb:  $\ge 0.77\%$  by weight  $Ba(OH)_2 \cdot H_2O$  to oil. (A greater amount of  $Ba(OH)_2 \cdot H_2O$  may be required to neutralize larger acid numbers.)
- 5. The use of bleaching clay and activated charcoal was found unnecessary.
- 6. An extensive used oil screening method had to be devised for identification of contaminated lots to eliminate those used oil samples not suitable for reclamation.
- 7. Extensive laboratory evaluation on each reclaimed basestock was necessary to determine optimum additive replenishment concentrations without causing high foam.
- 8. Ten 25 gallon batches of used oil were reclaimed. MIL-L-7808H test results on the batches were generally good. However, none of the ten batches completely passed all the tests.
- 9. A cost analysis of the process was carried out. Assuming a half million quarts of oil to be reclaimed each year, each gallon could be reclaimed for \$8.74.

# 7. RECOMMENDATIONS

The process as now defined has been shown to be feasible for reclamation of used MIL-L-7808H oils. However, the following are areas in which it is recommended that further effort be placed for improvement and refinement of the process.

- 1. Additional 15-25 gallon plant reclamations for additional data.
  - a. Additional MIL-L-7808H test results.
  - b. Statistics on precipitation formation in processed oil.
  - c. Correlate GC information of used oil with lab distillation/ foam test data to lower initial screening costs.
  - d. To develop a better understanding between reformulation, foaming and accelerated storage stability results.
- 2. Two or three 500 gallon plant reclamations for additional scale-up information.

## **BIBLIOGRAPHY**

Gunderson, R. C.; Hart, A. W.; et al. Synthetic Lubricants, New York, Reinhold Publishing Corp. 1962.

- "Reclamation of Synthetic Turbine Engine Oil Mixtures," Technical Report AFAPL-TR-78-50, AF Aero Propulsion Lab, AF Wright Aeronautical Laboratories, AF Systems Command, Wright-Patterson Air Force Base, OH 45433.
- Ronney, W. M. Lubricant Additives, Chemical Technology Review No. 104, Park Ridge, NJ, Noyes Data Corp. 1978.
- Ronney, W. M. Synthetic Oils and Greases for Lubricants, Chemical Technology Review No. 72, Park Ridge, NJ, Noyes Data Corp. 1976.
- "Turbine Engine Lubricant Reclamation," Technical Interim Report AFWAL-TR-81-2053, Aero Propulsion Laboratory, AF Wright Aeronautical Laboratories, AF Systems Command, Wright Patterson Air Force Base, OH 45433.

# APPENDIX A

# MIL-L-7808H TEST RESULTS FOR SELECTED FORMULATED VIRGIN BASE STOCKS

1732506 (Hatco base stock)

1732510 (Stauffer base stock)

1732511 (Rohm and Haas base stock) (Plexol)

1732508 (APL-furnished ATL 9148 base stock)

1732509 (APL-furnished ATL 9149 base stock)

# ADDITIVE LEVELS USED IN VIRGIN BASE STOCKS

Additives	Percent by weight
Tricresyl phosphate	2.0
4,4'-Dioctyldiphenylamine	1.0
Phenyl- $\alpha$ -naphthylamine	1.0
Benzotriazole	0.1
Triphenyl phosphite	0.1
Quinizarin	0.05
Antioxidant 703	0.1

PHYSICAL & CHEMICAL PROPERTIES  Neutralization Number 0.30 Max. 0.14 Viscosity @ 210°F, cs 3.0 Min. 3.0 Viscosity @ 100°F, cs Report 11.5  VISCOSITY STABILITY @ -65°F  Original Oil, 35 Min., cs 17,000 Max. 11,341 After 3 Hours, cs 17,000 Max. 11,633
Neutralization Number 0.30 Max. 0.14 Viscosity @ 210°F, cs 3.0 Min. 3.0 Viscosity @ 100°F, cs Report 11.5  VISCOSITY STABILITY @ -65°F  Original Oil, 35 Min., cs 17,000 Max. 11,341
Viscosity @ 210°F, cs 3.0 Min. 3.0 Viscosity @ 100°F, cs Report 11.5  VISCOSITY STABILITY @ -65°F  Original Oil, 35 Min., cs 17,000 Max. 11,341
Viscosity @ 210°F, cs 3.0 Min. 3.0 Viscosity @ 100°F, cs Report 11.5  VISCOSITY STABILITY @ -65°F  Original Oil, 35 Min., cs 17,000 Max. 11,341
Viscosity @ 100°F, cs Report 11.5  VISCOSITY STABILITY @ -65°F  Original Oil, 35 Min., cs 17,000 Max. 11,341
Original Oil, 35 Min., cs 17,000 Max. 11,341
Original Oil, 35 Min., cs 17,000 Max. 11,341
722
1. /22
Aller 3 mours. Ls
Viscosity Change, % 6.0 Max. +2.6
FOAMING CHARACTERISTICS, STATIC
176°F, Vol. after 30 Min. Aeration, ml 100 Max. 35
Collapse time, seconds 60 Max. 6.8
FOAMING CHARACTERISTICS, DYNAMIC
176°F Volume @ 1000 cc air. cc 100 Max. 10
votame & root of birty of
Contapo di inc.
Volume & 1500 cc arr, cc
Collapse time, immutes
Volume & 2000 Cc all, Cc
Conapse time, minutes
230°F Volume @ 1000 cc air. cc 100 Max. 10
Volume & 1000 to all, to
Collapse time, minutes 60 Max. 31.8 sec.  Volume @ 1500 cc air, cc 150 Max.
Collapse time, minutes 60 Max. 31.8 sec.
Volume @ 2000 cc air, cc Report 20
Collapse time, minutes 60 Max. 35.5 sec.

Sample Monsanto #1732511		Page 2.
	Specification	Results
FA ELASTOMER COMPATIBILITY, 72 H	OURS @ 347°F	
% Swell Tensile Strength, % Change Elongation, % Change Hardness, No. Change	2.0 - 25.0 50 Max. 50 Max. 20 Max.	+15.8 -14.0 +15.6 -5
LEAD CORROSION, 1 Hour @ 325°F		
Weight Change, mg/in <sup>2</sup>	6 Max.	0.0
SILVER & BRONZE CORROSION, 50 HO	URS @ 450°F	
Silver Weight Change, mg/in <sup>2</sup> Bronze AMS4616, mg/in <sup>2</sup>	3. 0 Max. 3. 0 Max.	-0.1 0.0
DEPOSITION NUMBER (see attached data	sheet)	
Deposit Number Viscosity Change, % TAN Change Oil Consumption	1.5 Max. Report Report Report	0.67 55.6 22.88 175 cc
RYDER GEAR TEST (see attached data s	heet)	
2 Determinations, ppi	2400 Min.	2631

生まれる (Manager Manager Manage

Specification Results

# CORROSION AND OXIDATION STABILITY, 96 HOURS @ 392°F

		16     24     40     48     64     72     88     96       4.3     +5.2     +13.0     +24.3     +56.5     +74.8     +113.9     +129.6       6.64     1.16     5.58     12.26     24.30     27.46     38.26     40.74       5.9       6.2
-0.030 -0.148 +0.010 -12.247 +0.053 +0.050	None None Mg None	16 +4.3 + 0.64
Report Report Report Report Report Report	Report Report Report Report	Report Report Report Report Report
Corrosion: Steel, mg/cm² Silver, mg/cm² Aluminum, mg/cm² Magnesium, mg/cm² Bronze AMS4616, mg/cm² Titanium, mg/cm²	Appearance of Metal Specimens: Pitting Etching Corrosion Staining	Oxidation: Viscosity @ 100°F, % Change Viscosity @ 210°F, % Change Total Acid Number, Change Evaporation Loss, % Sludge, Volume, %

SAMPLE Monsanto #1732509	Date 4/07/80	
	Specification	Results
PHYSICAL & CHEMICAL PROPERTIES		
Neutralization Number	0.30 Max.	0.17
Viscosity @ 210°F, cs	3.0 Min.	3.5
Viscosity @ 100°F, cs	Report	14.8
VISCOSITY STABILITY @ -65°F		
Original Oil, 35 Min., cs	17,000 Max.	16,642
After 3 Hours, cs	17,000 Max.	16,954
Viscosity Change, %	6.0 Max.	+1.9
FOAMING CHARACTERISTICS, STATIC		
176°F, Vol. after 30 Min. Aeration, ml	100 Max.	115
Collapse time, seconds	60 Max.	7.4
FOAMING CHARACTERISTICS, DYNAMIC		
176°F		
Volume @ 1000 cc air, cc	100 Max.	10
Collapse time, minutes	60 Max.	12.7 sec.
Volume @ 1500 cc air, cc	150 Max.	<b>3</b> 0
Collapse time, minutes	60 Max.	14.7 sec.
Volume @ 2000 cc air, cc	Report	30
Collapse time, minutes 230°F	60 Max.	18.9 sec.
Volume @ 1000 cc air, cc	100 Max.	10
Collapse time, minutes	60 Max.	10 13.1 sec.
Volume @ 1500 cc air, cc	150 Max.	20
Collapse time, minutes	60 Max.	14.2 sec.
Volume @ 2000 cc air, cc	Report	<b>2</b> 0
Collapse time, minutes	60 Max.	14.7 sec.

Sample Monsanto #1732509		Page 2.
	Specification	Results
FA ELASTOMER COMPATIBILITY, 72	HOURS @ 347°F	
% Swell	2.0 - 25.0	+15.8
Tensile Strength, % Change	50 Max.	-11.4
Elongation, % Change	50 Max.	+35. 9
Hardness, No. Change	20 Max.	0
LEAD CORROSION, 1 Hour @ 325°F		
Weight Change, mg/in <sup>2</sup>	6 Max.	0.0
SILVER & BRONZE CORROSION, 50 H	OURS @ 450°F	
Silver Weight Change, mg/in <sup>2</sup>	3.0 Max.	0.0
Bronze AMS4616, mg/in <sup>2</sup>	3.0 Max.	0. 1
DEPOSITION NUMBER (see attached da	ta sheet)	
Deposit Number	1.5 Max.	0.56
Viscosity Change, %	Report	112.8
TAN Change	Report	14.41
Oil Consumption	Report	150 cc
RYDER GEAR TEST (see attached data	sheet)	
2 Determinations, ppi	2400	3282

Results Specification

Corrosion: Steel, mg/cm <sup>2</sup>	Report	-0.024							
Aluminum, mg/cm <sup>2</sup>	Report	0.000							
Magnesium, mg/cm <sup>2</sup>	Report	-0.036							
Bronze AMS4616, mg/cm <sup>2</sup>	Report	+0,010							
Titanium, mg/cm <sup>2</sup>	Report	-0.020							
M50 Steel, mg/cm <sup>2</sup>	Report	+0. 030							
Appearance of Metal Specimens:									
Pitting	Report	None							
Etching	Report	None							
Corrosion	Report	None							
Staining	Report	None							
Oxidation:		16	24	40	48	64	72	88	96
Viscosity @ 100°F, % Change	e Report	<del>1</del> 6.1	+7.4	+9.5	+10.1	+9.5 +10.1 +12.2 +12.8 +14.2	+12.8	+14.2	+14.2
Viscosity @ 210°F. % Change									+8.6
Total Acid Number, Change		0.51	0.83		1.17 1.23	1.57	1.69	1.69 1.91	2.30
Evaporation Loss. %									3.8
									c

SAMPLE Monsanto #1732510	Date 4/07/80		
	Specification	Results	
PHYSICAL & CHEMICAL PROPERTIES			
Neutralization Number	0.30 Max.	0. 25	
Viscosity @ 210°F, cs Viscosity @ 100°F, cs	3.0 Min. Report	3. 1 12. 8	
VISCOSITY STABILITY @ -65°F			
Original Oil, 35 Min., cs	17,000 Max.	13,925	
After 3 Hours, cs	17,000 Max.	13,484	
Viscosity Change, %	6.0 Max.	+3.1	
FOAMING CHARACTERISTICS, STATIC			
176°F, Vol. after 30 Min. Aeration, ml	100 Max.	<b>3</b> 0	
Collapse time, seconds	60 Max.	3.8	
FOAMING CHARACTERISTICS, DYNAMIC			
176°F Volume @ 1000 cc air. cc	100 Max.	10	
Collapse time, minutes	60 Max.	9.1 sec.	
Volume @ 1500 cc air, cc	150 Max.	10	
Collapse time, minutes	60 Max.	9.3 sec.	
Volume @ 2000 cc air, cc	Report	10	
Collapse time, minutes	60 Max.	10.3 sec.	
230°F			
Volume @ 1000 cc air, cc	100 Max.	10	
Collapse time, minutes	60 Max.	8.5 sec.	
Volume @ 1500 cc air, cc	150 Max.	10	
Collapse time, minutes	60 Max.	8.2 sec.	
Volume @ 2000 cc air, cc	Report	10	
Collapse time, minutes	60 Max.	8.9 sec.	

Sample Monsanto #1732510		Page 2.
	Specification	Results
FA ELASTOMER COMPATIBILITY, 72 HO	URS @ 347°F	
% Swell	2.0 - 25.0	+17.6
Tensile Strength, % Change	50 Max.	-11.9
Elongation, % Change	50 Max.	+17.2
Hardness, No. Change	20 Max.	- 5
LEAD CORROSION, 1 Hour @ 325°F		
Weight Change, mg/in <sup>2</sup>	6 Max.	-0.1
SILVER & BRONZE CORROSION, 50 HOU	RS @ 450°F	
Silver Weight Change, mg/in <sup>2</sup>	3.0 Max.	-0.1
Bronze AMS4616, mg/in <sup>2</sup>	3. 0 Max.	-0.2
DEPOSITION NUMBER (see attached data	sheet)	
Deposit Number	1.5 Max.	<b>0</b> . <b>4</b> 0
Viscosity Change, %	Report	82.0
TAN Change	Report	14.85
Oil Consumption	Report	125 cc
RYDER GEAR TEST (see attached data she	et)	
2 Determinations, ppi	2400 Min.	2671

The second of th

Specification Results

# CORROSION AND OXIDATION STABILITY, 96 HOURS @ 392.F

	96	+14.8	49.7 2.89 5.2 0.0
	88	+14.1	2. 33
	72	+10.7 +11.7 +14.1	1.61 1.77 2.05
	64	+10.7	1.77
	48	+9.4	
	40	+7.0	1. 43
6 0 9 0 5 C 8	24	+7.0	0.87
+0. 069 -0. 040 +0. 016 +0. 030 +0. 115 +0. 050 -0. 028 None None None	9]	+6.3	0. 71
Report Report Report Report Report Report Report Report	1	Report	Report Report Report Report
Steel, mg/cm <sup>2</sup> Silver, mg/cm <sup>2</sup> Silver, mg/cm <sup>2</sup> Aluminum, mg/cm <sup>2</sup> Magnesium, mg/cm <sup>2</sup> Bronze AMS4616, mg/cm <sup>2</sup> Titanium, mg/cm <sup>2</sup> M50 Steel, mg/cm <sup>2</sup> Appearance of Metal Specimens: Pitting Etching Corrosion Staining	Oxidation:	Viscosity @ 100 F. % Change	Total Acid Number, Change Evaporation Loss, % Sludge, Volume, %

SAMPLE Monsanto #1732506	Date	4/07/80	<u> </u>
	Specification	Result	<u>s</u>
PHYSICAL & CHEMICAL PROPERTIES			
Neutralization Number	0.30 Max.	<b>0</b> . <b>2</b> 0	
Viscosity @ 210°F, cs	3.0 Min.	<b>3</b> . 0	
Viscosity @ 100°F, cs	Report	12.5	
VISCOSITY STABILITY @ -65°F			
Original Oil, 35 Min., cs	17,000 Max.	12,136	
After 3 Hours, cs	17,000 Max.	12,561	
Viscosity Change, %	6.0 Max.	3.5	
FOAMING CHARACTERISTICS, STATIC			
176°F, Vol. after 30 Min. Aeration, ml	100 Max.	15	
Collapse time, seconds	60 Max.	3. 2	
FOAMING CHARACTERISTICS, DYNAMIC			
176°F		• •	
Volume @ 1000 cc air, cc	100 Max.	10	
Collapse time, minutes	60 Max.	7. 9	sec.
Volume @ 1500 cc air, cc	150 Max.	10	
Collapse time, minutes	60 Max.	8.8	sec.
Volume @ 2000 cc air, cc	Report	10	
Collapse time, minutes	60 Max.	8.5	sec.
230°F			
Volume @ 1000 cc air, cc	100 Max.	10	
Collapse time, minutes	60 Max.	8.4	sec.
Volume @ 1500 cc air, cc	150 Max.	10	
Collapse time, minutes	60 Max.	8.5	sec.
Volume @ 2000 cc air, cc	Report	10	
Collapse time, minutes	60 Max.	7.6	sec.

Sample Monsanto #1732506		Page 2.
	Specification	Results
FA ELASTOMER COMPATIBILITY, 72 H	OURS @ 347°F	
% Swell Tensile Strength, % Change Elongation, % Change Hardness, No. Change	2.0 - 25.0 50 Max. 50 Max. 20 Max.	+12. 4 +2. 1 +9. 3 -5
LEAD CORROSION, 1 Hour @ 325°F		
Weight Change, mg/in <sup>2</sup>	6 Max.	0.0
SILVER & BRONZE CORROSION, 50 HO	URS @ 450°F	
Silver Weight Change, mg/in <sup>2</sup> Bronze AMS4616, mg/in <sup>2</sup>	3.0 Max. 3.0 Max.	0.0 -0.2
DEPOSITION NUMBER (see attached data	a sheet)	
Deposit Number Viscosity Change, % TAN Change Oil Consumption	1.5 Max. Report Report Report	0.52 48.8 13.10 110 cc
RYDER GEAR TEST (see attached data s	heet)	

2 Determinations, ppi

2400 Min. 2919

Specification Results

# CORROSION AND OXIDATION STABILITY, 96 HOURS @ 392°F

	96 112. 8 116. 7 2. 91 3. 8
	16     24     40     48     64     72     88     96       5     6     +6, 4     +8, 8     +8, 8     +10, 4     +11, 2     +12, 0     +12, 8       6     0, 24     0, 42     1, 14     1, 38     1, 82     2, 04     2, 44     2, 9       3     8     1, 14     1, 14     1, 38     1, 182     2, 04     2, 44     2, 9
	72 11. 2 + 2. 04
	1.82
	1.38
	40 +8.8 1.14
	24 +6. 4 0. 42
+0. 010 -0. 069 -0. 030 -0. 022 -0. 032 -0. 026 -0. 026 None None None	16 + 5 6 + 0 . 24
Report Report Report Report Report Report Report Report Report	Report Report Report Report
Corrosion:  Steel, mg/cm² Silver, mg/cm² Aluminum, mg/cm² Magnesium, mg/cm² Bronze AMS4616, mg/cm² Titanium, mg/cm² M50 Steel, mg/cm² Appearance of Metal Specimens: Pitting Etching Corrosion Staining	Oxidation: Viscosity @ 100°F, % Change Viscosity @ 210°F, % Change Total Acid Number, Change Evaporation Loss, % Sludge, Volume, %
118	

SAMPLE Monsanto #1732508	Date	4/07/80
	Specification	Results
PHYSICAL & CHEMICAL PROPERTIES		
Neutralization Number	0.30 Max.	0. 22
Viscosity @ 210°F, cs	3.0 Min.	3.4
Viscosity @ 100°F, cs	Report	14, 2
VISCOSITY STABILITY @ -65°F		
Original Oil, 35 Min., cs	17,000 Max.	14,462
After 3 Hours, cs	17,000 Max.	14,799
Viscosity Change, %	6.0 Max.	+2, 3
FOAMING CHARACTERISTICS, STATIC  176°F, Vol. after 30 Min. Aeration, ml	100 Max.	100
Collapse time, seconds	60 Max.	<b>5. 1</b>
FOAMING CHARACTERISTICS, DYNAMIC 176°F		
Volume @ 1000 cc air, cc	100 Max.	10
Collapse time, minutes	60 Max.	18.9 sec.
Volume @ 1500 cc air, cc	150 Max.	20
Collapse time, minutes	60 Max.	16.4 sec.
Volume @ 2000 cc air, cc	Report	20
Collapse time, minutes	60 Max.	19.5 sec.
230°F		
Volume @ 1000 cc air, cc	100 Max.	10
Collapse time, minutes	60 Max.	12.9 sec.
Volume @ 1500 cc air, cc	150 Max.	10
Collapse time, minutes	60 Max.	15, 2 sec.
Volume @ 2000 cc air, cc	Report	20
Collapse time, minutes	60 Max.	13.9 sec.

Sample Monsanto #1732508		Page 2.
	Specification	Results
FA ELASTOMER COMPATIBILITY, 72 H	HOURS @ 347°F	
% Swell Tensile Strength, % Change Elongation, % Change Hardness, No. Change	2.0 - 25.0 50 Max. 50 Max. 20 Max.	+16.1 -12.1 +12.5 -5
LEAD CORROSION, 1 Hour @ 325°F		
Weight Change, mg/in <sup>2</sup>	6 Max.	0.0
SILVER & BRONZE CORROSION, 50 HC	OURS @ 450°F	
Silver Weight Change, mg/in <sup>2</sup> Bronze AMS4616, mg/in <sup>2</sup>	3.0 Max. 3.0 Max.	0.1 0.1
DEPOSITION NUMBER (see attached data	a sheet)	
Deposit Number Viscosity Change, % TAN Change Oil Consumption	1.5 Max. Report Report Report	0.51 118.3 11.96 135 cc
RYDER GEAR TEST (see attached data s	heet)	
2 Determinations, ppi	2400 Min	2815

Specification Results

# CORROSION AND OXIDATION STABILITY, 96 HOURS @ 392°F

			<i>;</i>
	96	+11.9	1. 60 4. 6 0. 0
	88	+11.9	1.36
	72	+11.2	1. 24
	64	+9.8	1.08
	48	+8, 5	0.84
		+8.5	0.74
		+6.3	0.46
None None	None	+5.6	0.42
Report Report Report	Report	Report Report	Report Report Report
Pitting Etching Corrosion	Staining Oxidation:	Viscosity @ 210°F, % Change Viscosity @ 210°F, % Change	Total Acid Number, Change Evaporation Loss, % Sludge, Volume, %
	Pitting Report Etching Corrosion Report	Pitting Etching Corrosion Staining Oxidation:	Pitting         Report         None           Corrosion         Report         None           Staining         Report         None           Oxidation:         16 24 40 48 64 72 88           Viscosity @ 100°F, % Change         Report         +5.6 +6.3 +8.5 +8.5 +9.8 +11.2 +11.9 +           Viscosity @ 210°F, % Change         Report         +5.6 +6.3 +8.5 +8.5 +9.8 +11.2 +11.9 +

# APPENDIX B

# MIL-L-7808H TEST RESULTS FOR LAB RECLAIMED OIL

1830396-C/M-23 (Reclaimed oil, NaOH/methanol, vapor temperature distillation 135-285°C)

1830397-C/M-234 (Reclaimed oil, NaOH/methanol, vapor temperature distillation 135-305°C)

# ADDITIVE LEVELS USED IN RECLAIMED BASE STOCK

Additives	Percent by weight
Tricresyl phosphate	2.0
4,4'-Dioctyldiphenylamine (DODPA)	1.0
Phenyl- $\alpha$ -naphthylamine	1.0 <sup>a</sup>
Benzotriazole	0.1
Triphenyl phosphite	0.1
Quinizarin	0.05
Antioxidant 703	0.10

<sup>&</sup>lt;sup>a</sup>Both samples inadvertently had more (3%) than 1% PANA added.

CUSTOMER Monsanto Research Corpo	Date 10/16/80						
		Results					
	Specification	•	•				
PHYSICAL & CHEMICAL PROPERTIES		6C/M-23	7C/M-234				
Neutralization Number (T.A.N.) Viscosity @ 210°F, cs	0.30 Max. 3.0 Min.	0. 22 3. 0	0. 20 3. 1				
VISCOSITY STABILITY @ -65°F							
Original Oil, 35 Minutes, cs After 3 Hours, cs Viscosity Change, %	17,000 Max. 17,000 Max. 6.0 Max.	9,777	10,428 10,340 -0.8				
FOAMING CHARACTERISTICS, STATIC							
176°F, Volume after 30 Min. Aeration, ml Collapse Time, seconds	100 Max. 60 Max.	100 4. 2	120 5. 1				
LEAD CORROSION, 1 HOUR @ 325°F							
Weight Change, mg/in <sup>2</sup>	6 Max.	+0.1	+0.1				
SILVER & BRONZE CORROSION, 50 HOURS @ 450°F							
Silver Weight Change, mg/in <sup>2</sup> Bronze AMS4616, mg/in <sup>2</sup>	3.0 Max. 3.0 Max.	-0.1 -0.3	-0.1 -0.5				
DEPOSITION NUMBER (see attached data sheets)							
Deposit Number Viscosity Change, % T.A.N. Change Oil Consumption	Report	0.53 119.3 25.04 100 cc	0.41 +38.3 11.42 50 cc				

R	_	•	u	1	ŧ	8
- 1	C	3	u	1	ι	3

	Specification	183039- 6C/M-23	183039- 7C/M-2
CORROSION & OXIDATION STABILIT	Y, 48 HOURS @ 392	<u>•</u> F	
Corrosion:			
Steel, mg/cm <sup>2</sup>	10.3	0.010	
	<u>+</u> 0. 2	-0.010	-0.099
Silver, mg/cm <sup>2</sup>	<u>+</u> 0. 2	-0.139	-0.050
Aluminum, mg/cm <sup>2</sup>	+0.2	-0.073	-0.040
Magnesium, mg/cm <sup>2</sup>	<u>+</u> 0. <b>4</b>	-0.050	-0.030
Bronze AMS4616, mg/cm <sup>2</sup>	<u>+</u> 0.4	-0.119	-0.010
Titanium, mg/cm <sup>2</sup>	<u>+0.2</u>	-0.040	-0.080
M-50 Steel, mg/cm <sup>2</sup>	<u>+</u> 0. 2	-0.080	-0.046
Appearance of Metal Specimens:			
Pitting	Report	None	None
Etching	Report	None	None
Corrosion	Report	None	None
Staining	Report	None	None
Oxidation:			
Viscosity @ 100°F, Initial	Report	11.4	12.0
Viscosity @ 100°F, % Change	-5 to +25	+25.4	+23, 3
Viscosity @ 210°F, Initial	Report	3.0	3. 1
Viscosity @ 210°F, % Change	Report	+13.3	+12.9
Total Acid Number, Initial	Report	0.22	0, 20
Total Acid Number, Change	4.0 Max.	5.03	4.80
Evaporation Loss, %	Report	2. 5	2. 1
Sludge, Volume, %	Report	0.0	0.0

# APPENDIX C

# ANALYTICAL CONDITIONS

Gas Chromatography
High Performance Liquid Chromatography
Thin Layer Chromatography

# ANALYTICAL CONDITIONS

### GAS CHROMATOGRAPHY

Column: 3% Dexsil 300 on Chromasorb

W 80/100 0.125 in. OD 12 ft

length acid washed and silanized

Sample: 10% in cyclohexane

Sample size:  $1 \mu L$ 

Helium flow rate: 40 cc/min

Inlet temperature: 300°C

FID temperature: 400°C

Temperature programs: 180-350°C at 4°C/min

100-350°C at 4°C/min

# HIGH PERFORMANCE LIQUID CHROMATOGRAPHY

# General Conditions

Column: Partisil PXS 10/25 PAC

Sample: 10% in cyclohexane

Sample size:  $10 \mu L$ 

Mobile phase: 100% isooctane TO 70/30

methylene chloride/isooctane,

2 mL/min.

Program: Linear, 20 min

Detector: UV @ 254 nm

Chart: 0.5 cm/min

# Specific Conditions

## PANA AND DODPA ANALYSIS

Program: Program #7 on Water's Solvent

Programmer, 7 min., 2 mL/min.

Mobil phase: 100% isooctane to 95/5 methylene

chloride/isooctane

#### ANALYTICAL CONDITIONS (continued)

#### TCP ANALYSIS

Program:

Program #6 on Water's Solvent Programmer, 15 min., 2 mL/min.

Mobil phase:

100% isooctane to 90/10 methylene

chloride/isooctane

Sample:

100  $\mu$ L of 30% oil

Area determination of peaks was found to be more accurate by calculating the areas by hand. More reproduceable data may be obtained with the use of second generation analytical software.

#### THIN LAYER CHROMATOGRAPHY

Plate:

Silica gel F-254 activated

at 100°C for 1 hr

Sample:

30  $\mu$ L 10% solution oil in

cyclohexane

Mobile phase:

30/70 heptane/toluene

Lamp:

254 nm UV

#### APPENDIX D

HIGH PERFORMANCE LIQUID CHROMATOGRAMS OF ORIGINAL 15 USED OILS AS RECEIVED

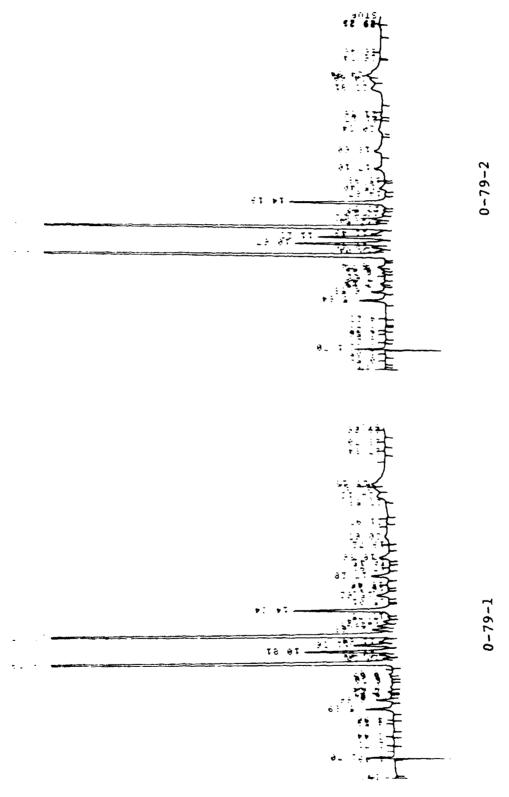


Figure D-1. High performance liquid chromatograms of used oil as received.

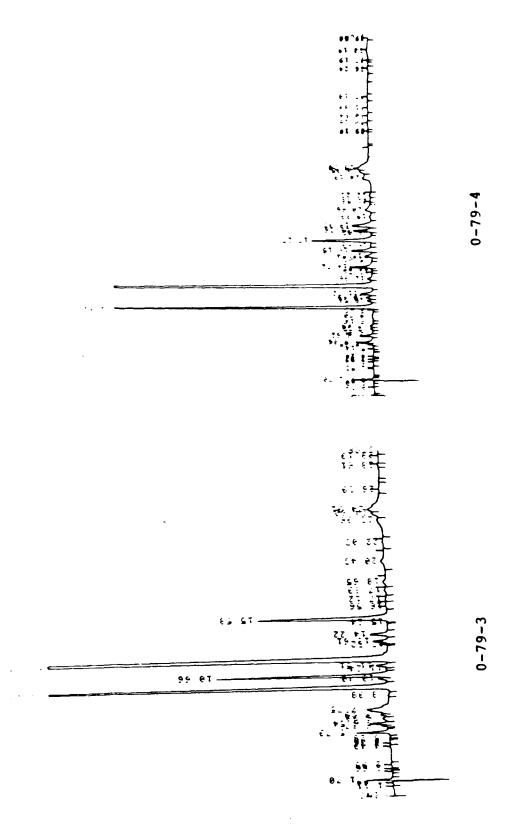


Figure D-2. High performance liquid chromatograms of used oil as received.

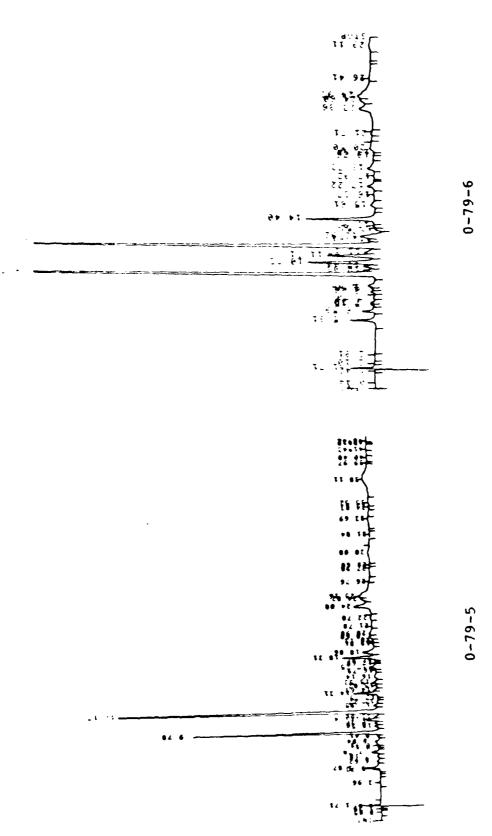
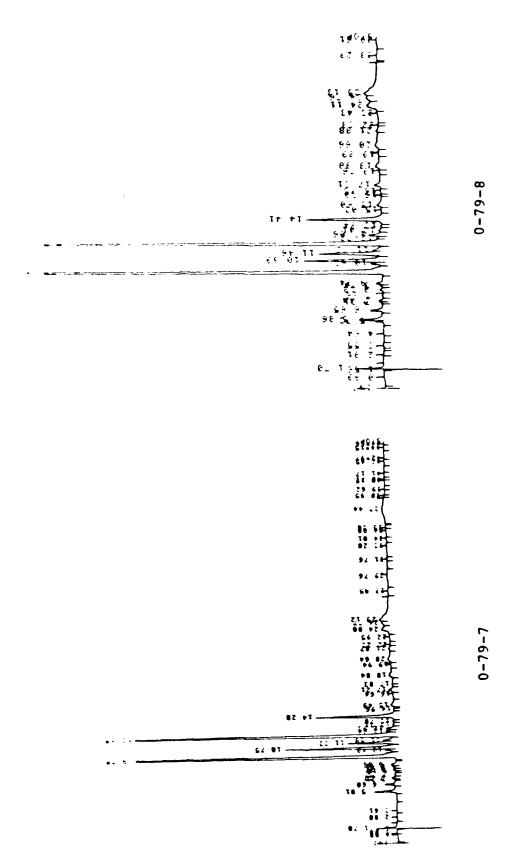
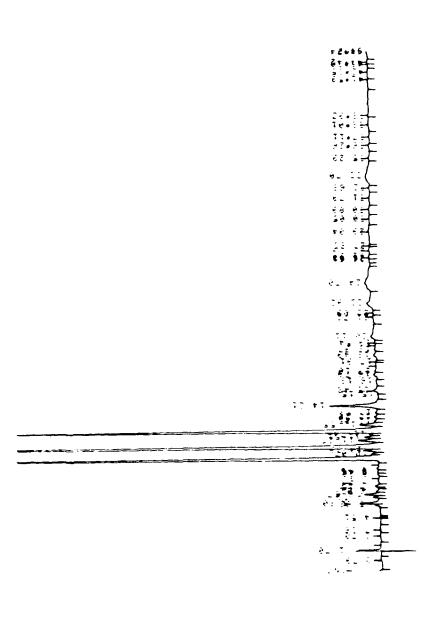


Figure D-3. High performance liquid chromatograms of used oil as received.



High performance liquid chromatograms of used oil as received.



High performance liquid chromatogram of used oil as received. Figure D-5.



Figure D-6. High performance liquid chromatograms of used oil as received.

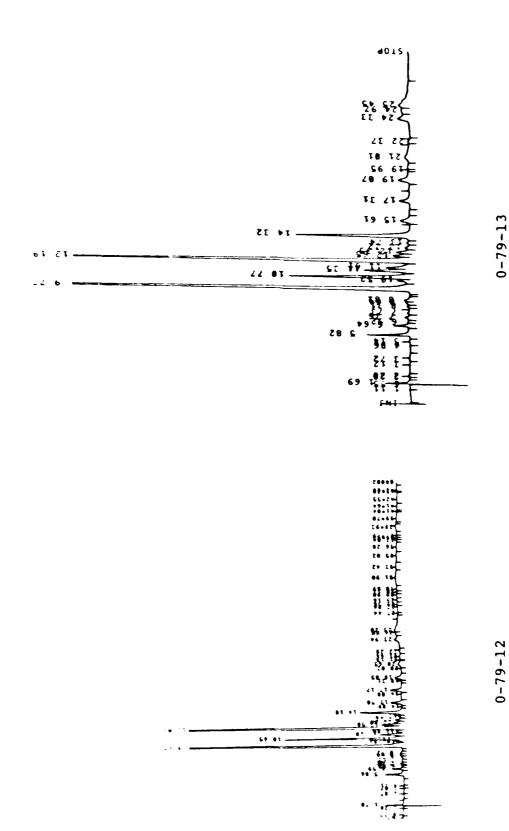


Figure D-7. High performance liquid chromatograms of used oil as received.

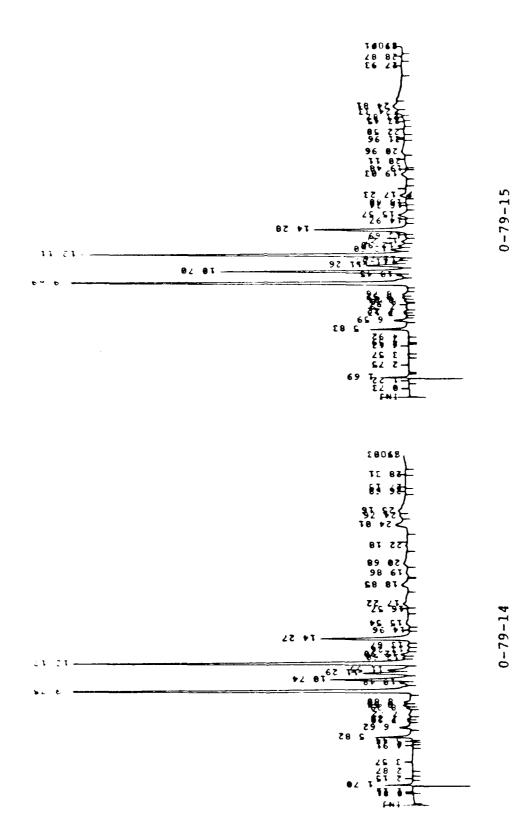


Figure D-8. High performance liquid chromatograms of used oil as received.

#### APPENDIX E

MIL-L-7808H TEST RESULTS FOR LARGE SCALE (25 GALLON) RECLAIMED OIL BATCHES

LEVEL OF ADDITION (% by wt), ADDED TO LARGE SCALE RECLAIMED OILS

2000248 (0-79-8) Original	D 40 10	2000275 (0-79-12) Orig- inal	ne 1.19 23 1.29 ne 0.36 1 1 1 20
			1.24 none 1.46 0.23 0.3 none 0.1 0.1 0.05
1997693 (0-79-13) Orig- inal Added level	none 2 0.15 1 0.25 0 0.1 0.1	2000273 (0-79-10) Orig ina Added leve	none 1. 0.24 1. none 0. 0.1 0.05
1997695 (0-79-6) Orig- inal Added level	none 2.0 0.25 1.24 0.25 0.25 0.1 0.1 0.05	2000268 (0-79-14) Orig- inal	none 1.0 0.57 1.2 none 0.26 0.1 0.1 0.05
1997659 (0-79-9) Orig- inal	1.2 0.8 0.4 1.01 0.2 0.19 0.1 0.05	2000267 (0-79-7) Orig- inal Added level	1.2 0.8 0.18 1.32 0.23 0.27 0.1 0.1 0.05
1997658 (0-79-11) Original	0.7 1.3 none 1.4 0.2 0.31 0.1 0.0 0.0	2000247 (0-79-15) Orig- inal Added level	0.8 0.68 0.5 1.3 none 0.2 0.1 0.05
	Tricresyl phosphate 4,4'-dioctyldiphenylamine (DODPA) Phenyl-alpha-naphthylamine (PANA) Benzotriazole Triphenyl phosphite Quinizarin Antioxidant 703		Tricresyl phosphate 4,4'-dioctyldiphenylamine (DODPA) Phenyl-alpha-naphthylamine (PANA) Benzotriazole Triphenyl phosphite Quinizarin Antioxidant 703

#### GEAR LOAD CARRYING RATING

#### Method ASTM D1947 (one gear, two determinations) (Ryder Gear)

NB number	Used oil	Test (PPI)	Specifications
1997693	(0-79-13)	2935	2400 PPI min.
1997695	(0-79-6)	2697	
1997659	(0-79-9)	2364	
1997658	(0-79-11)	2816	
2000248	(0-79-8)	2587	
2000247	(0-79-15)	2407	
2000275	(0-79-12)	2787	
2000273	(0-79-10)	3092	
2000267	(0-79-7)	2967	
2000268	(0-79-14)	2813	

BEARING DEPOSITION TEST SUMMARY

	2000248 (0-79-8) & 2000247 (0-79-15) Test No. 48-312	2000267 (0-79-7) & 2000268 (0-79-14) Test No. 48-313	2000273 (0-79-10) & 2000275 (0-79-12) Test No. 48-314	1997693 (0-79-10) & 1997695 (0-79-6) Test ilo, 48-316	1997659 (0-79-9) & 1997658 (0-79-11) Test No. 48-317
Bearing Deposition Test (see enclosed copies)					
Overall rating, max. 60	47.1	63.9	33.0	39.7	46.6
Viscosity, % change, max. 25%	+7.3	+15.3	+8.1	7.6+	+8.7
TAN change, max. 2.5	2.66	2.63	1.32	1.93	1.77
Oil consumption	4.6	3.5	7.1	13.8	2.1
Total sludge, max. 2 grams	0.598	999.0	0.418	2.132	1.337

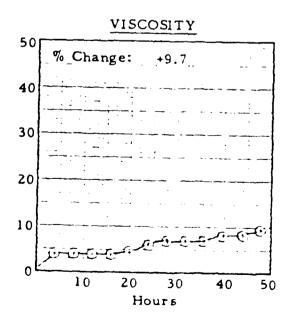
#### BEARING DEPOSITION TEST SUMMARY Test No. 48-316

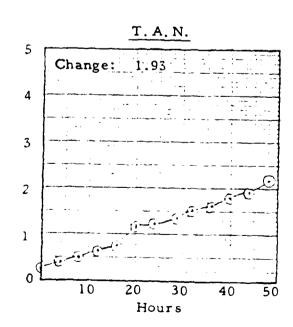
# MONSANTO RESEARCH CORPORATION

Sample No. 1997693 (0-79-13) & 1997695 (0-79-6)

(Rig No. 1 )

DEPOS	SITS	SLUD	GF
End Cover	0.0	Inlet Screen	<del></del>
Spacer & Nut	14.0	Outlet Screen	.762
Heater Front	99.6		1.370 OTAL 2.132
Heater Rear	86.4	•	OIAL 2.132
Seal Plate	10.0		
Bearing	28.0		
OVERALL RATING 39.7		OIL CONSUMPTION,	ml/hr 13.8
SUMP	<u>*</u>	<u>META</u> 1	LS
Sides	Lt. Varnish	Aluminum	-0.020
Bottom	Lt. Varnish	Silver	0.000
		Bronze	-0.020
* Not included	in the overall	Steel	+0.040
rating above.		M-50 Steel	-0.040
		Waspaloy	+0.059
		Titanium	+0.020





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# BEARING DEPOSITION TEST SUMMARY Test No. 48-317 MONSANTO RESEARCH CORPORATION Sample No. 1997659 (0-79-9) & 1197658 (0-79-11)

(Rig No. \_1\_\_)

DEPOSITS	
End Cover	1.2
Spacer & Nut	14.0
Heater Front	102.3
Heater Rear	119.1
Seal Plate	0.0
Bearing	43.0

OIL CONSUMPTION, ml/hr
OVERALL RATING 46.6

Sides Lt. Varnish Aluminum

Lt. Varnish

\* Not included in the overall rating above.

**Bottom** 

 Aluminum
 +0.020

 Silver
 +0.040

 Bronze
 0.000

 Steel
 -0.040

 M-50 Steel
 0.000

 Waspaloy
 +0.020

 Titanium
 0.000

SLUDGE

TOTAL

.392

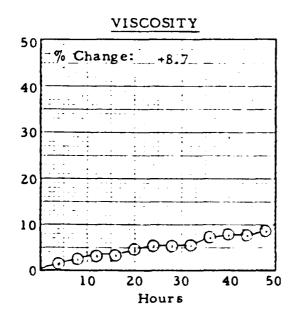
.945

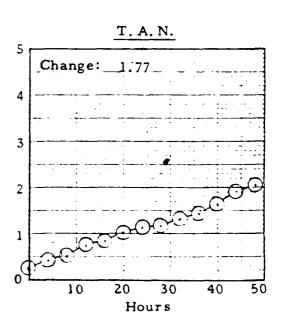
1.337

2.1

Inlet Screen

Outlet Screen





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(This sheet may be detached for filing)

#### BEARING DEPOSITION TEST SUMMARY

#### MONSANTO RESEARCH CORPORATION

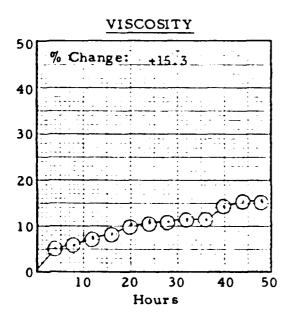
Test No. 48-313

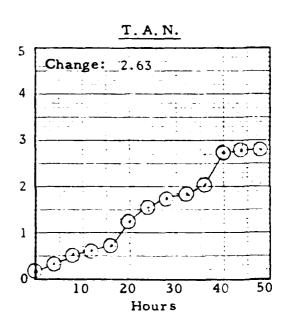
Sample No. 2000267 (0-79-7) & 2000268 (0-79-14)

(Rig No. 2 )

DEPOSITS		SLUDGE	
End Cover	1.2	Inlet Screen	.356
Spacer & Nut	29.0	Outlet Screen	.310
Heater Front	151.2	TOTAL	<b>.6</b> 66
Heater Rear	102.9		
Seal Plate	11.0		
Bearing	88.0		
		OIL CONSUMPTION, ml/hr	3.5
OVERALL RATING	63.9		
SUMP*		METALS	
Sides	Lt. Varnish	Aluminum	0.000
Bottom	Lt. Varnish	Silver	-0.040
		Bronze	+0.020
* Not included in the overall		Steel	0.000
rating above.		M-50 Steel	-0.020
<u> </u>		Waspaloy	0.000

Titanium





+0.020

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#### BEARING DEPOSITION TEST SUMMARY Test No. 48-312

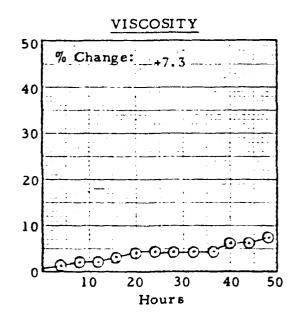
### MONSANTO RESEARCH CORPORATION

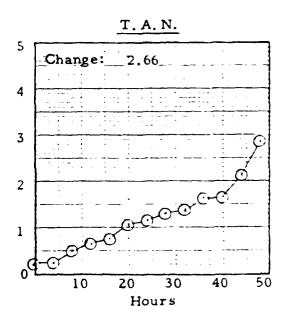
Sample No. 2000248 (0-79-8) & 2000247 (0-79-15)

(Rig No. 1 )

DEPOSITS		SLUDGE	
End Cover	1.5	Inlet Screen	.262
Spacer & Nut	14.0	Outlet Screen	.336
Heater Front	77.1	TOTAL	.598
Heater Rear	142.2		
Seal Plate	10.0		
Bearing	37.5		
-		OIL CONSUMPTION, ml/hr	4.6
OVERALL RATING	47.1		

SUMP*		METAL	S
Sides	Lt. Sludge	Aluminum	0.000 +0.040
Bottom	Lt. Varnish	Silver Bronze	+0.040
w Was in aludad is	o the eventl	Steel	0.000
* Not included in	the overall	M-50 Steel	0.000
rating above.		Waspaloy	+0.020
		Titanium	0.000





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### BEARING DEPOSITION TEST SUMMARY

#### MONSANTO RESEARCH CORPORATION

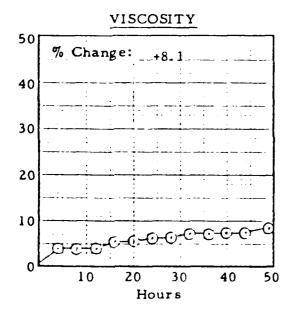
Test No. 48-314

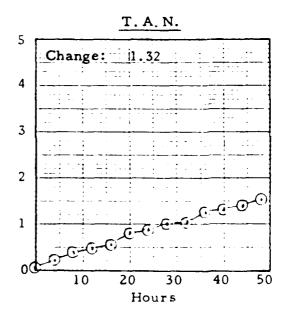
Sample No. 2000273 (0-79-10) & 2000275 (0-79-12)

(Rig No. \_1\_\_)

DEPOSITS		SLUDGE		
End Cover	0.0	Inlet Screen	.183	
Spacer & Nut	14.0	Outlet Screen	.235	
Heater Front	82.2	TOTAL	.418	
Heater Rear	64.2			
Seal Plate	10.0			
Bearing	27.5			
_		OIL CONSUMPTION, ml/hr	7.1	
OVERALL RATING	33.0			

SUM	1P*	META	LS
Sides	Lt. Varnish	Aluminum	-0.020
Bottom	Lt. Varnish	Silver	-0.020
		Bronze	+0.079
* Not included in the overall		Steel	-0.020
		M-50 Steel	+0.079
rating abov	above.	Waspaloy	-0.040
		Titanium	-0.040





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(This sheet may be detached for filing)

#### ALCOR Inc. 10130 Jones Maltsberger Road San Antonio, Texas 78284 (512) 349-3771

CUSTON	ER Monsanto Research Corporation	DATE August 19,	1981
SAMPLE	1997659, Batch 0-79-9		
		Specification	Results
	AL & CHEMICAL PROPERTIES		
	tralization Number (T.A.N.)	0.30 Max	0.17
	cosity @ 210°F, cs	3.0 Min	3.1
	cosity @ 100°F, cs	Report	12.8
Fla	sh Point (COC), °F	400.0 Min	435
Evaj	poration, 6.5 Hrs @ 400°F, Wt. Loss, %	30.0 Max	18.2
	SEDIMENT		
Free	cipitation, m1/200 m1	0.005 Max	0.000
	ILATE CONTAMINATION		
Cont	tamination, mg/ltr	10 Max	0.01
FOAMING	CHARACTERISTICS, STATIC		
176	F, Volume after 30 min. aeration, ml	100 Max	35 ml
Coll	apse time, sec.	60 Max	8.1 sec
	CHARACTERISTICS, DYNAMIC		-
176°F,	Volume at 1000 cc air, cc	100 Max	10 ml
	Collapse time, min.	60 Max	61.7 sec
	Volume @ 1500 cc air, cc	150 Max	10 ml
	Collapse time, min.	60 Max	61.3 ml
	Volume @ 2000 cc air, cc	Report	20 ml
_	Collapse time, min.	60 Max	58.5 sec
230°F,		100 Max	10 ml
	Collapse time, min.	60 Max	50.9 sec
	Volume @ 1500 cc air, cc	150 Max	15 ml
	Collarse time, min.	60 Max	55.6 sed
	Volume @ 2000 cc air, cc	Report 60 Max	30 ml 51.6 se <b>c</b>
	Collapse time, min.	ou max	51.0 500
	RROSION, 1 Hr @ 325°F	6 Max	0.0
Wt.	Change, rg/in-	O MAX	0.0
	& BRONTE CORR 10N, 50 Hrs @ 450°F	2 0 1/	0.1
	er kt. Charge, mg/in <sup>2</sup>	3.0 Max	- 0.1 - 0.5
Bron	ze, AMS:016, pg/ip2	3.0 Max	- 0.5

	Specification	Results
H ELASTOMER COMPATIBILITY, 168 Hrs @ 158°F		
% Swell	12.0-35.0	+28.5
FA ELASTOMER COMPATIBILITY, 72 Hrs @ 347°F		
% Swell	2.0-25.0	+16.4
Tensile Strength, % Change	50 Max	- 1.6
Elongation, % Change	50 Max	+24.6
Hardness, No. Change	20 Max	-10
QVI ELASTOMER COMPATIBILITY, 72 Hrs @ 302°F		
% Swell	2-30	+ 7.4
Tensile Strength, % Change	50 Max	-59.2
Elongation, % Change	50 Max	-34.3
Hardness, No. Change	20 Max	- 5
FS ELASTOMER COMPATIBILITY, 72 Hrs @ 302°F		
% Swell	2-25 Max	+ 5.2
Tensile Strength, % Change	50 Max	-58.9
Elongation, % Change	50 Max	-30.4
Hardness, No. Change	20 Max	- 5
DEPOSITION NUMBER (see attached data sheet)		
Deposit Number	1.5 Max	0.16
Viscosity Change, %	Report	73.4
TAN Change	Report	15.65
Oil Consumption	Report	125
ACCELERATED STORAGE STABILITY @ 230°F		
Lead Wt. Loss, mg/in <sup>2</sup>		
48 Hrs	25 Max	<del>+1.4</del> (+ (1.7)*
168 Hrs	150 Max	<del>-257.0</del> (123.8,*
VISCOSITY STABILITY, 3 Hrs @ -65°F		
Original Oil, cst	17,000 Max	12,805
3 Hrs, cst	17,000 Max	12,691
Viscosity Change, %	6.0 Max	0.9
VISCOSITY STABILITY, 72 Hrs @ -65°F		
Original Oil, cst	17,000	12,805
72 Hrs, cst	17,000	11,039
Viscosity Change, %	6.0 Max	13.8
WORKMANSHIP		
Clear, Transparent	Report	Clear

<sup>\*</sup>Corrected results

ALCOR Inc.
10130 Jones Maltsberger Road
San Antonio, Texas 78284
(512) 349-3771

DATE August 19, 1981

CUSTOMER	MIL-L-7808H QUALITY CONFORMANCE	CUSTOMER Monsanto Research Corporation
		CUSTOMER

1997659, Batch 0-79-9

SAMPLE

	96 12.8 24.2 3.1 19.4 0.17 3.83 2.1
	23.4
	20.3
	64 +18.0 3.71
	14.1
	+13.3
LM als	24 +7.8 1.17
0.000 -0.010 0.000 -15.940 -0.059 0.000 -0.010 None None None None None None None	16 +7.0 0.51
Specification  Meport Report	Report Report Report Report Report Report Report
CORROSION & OXIDATION STABILITY, 96 Hrs @ 39  Corrosion: Steel, mg/cm² Silver, mg/cm² Aluminum, mg/cm² Report Report Report Report Report Report Report Answersium, mg/cm² Appearance of Metal Specimens: Pitting Etching Corrosion Staining Report Report Report Report Report Report Report Report Report	Oxidation:  Viscosity @ 100°F, Initial Viscosity @ 100°F, % Change Viscosity @ 210°F, Initial Viscosity @ 210°F, % Change Total Acid Number, Initial Total Acid Number, Change Evaporation Loss, % Sludge, Volume, %

# MIL-L-7808 DEPOSITION TEST



Ţ	r. ĉ	:	1	Ur	٠,5	$1 \in A$	ANT	
_						<b>-</b> .	A	

DESIGNATION Monsanto, Sample 1997659, Batch 0-79-9

TEST NO 4464 DATE 8/3/81 OPERATOR Pavlicek, Trawick

Coking Tube 590°F · Oil In 300°F · Oil Flow 300 cc/min · Air Flow 300 cc/min

	ī ————	T	TEM	PERATURES	·F			[	ſ	
Fr. LRS	TIME	01L 1N	OILOUT		STUBE	CAP AIR	HEATER VOLTS	AIR ROTO-	DIL	FILTER PRESSURE
	<u> </u>	ļ. <u> </u> -	3	2	VAP SP	1		<u> </u>		
START	16.15	100	115	590	510	116	195	2.5	0_	1.0
	± <	1-00	15C	FG ()	100	11:	155	25	0	1.1
1	5 5 5	  - 24	<u>ر ۽ نہ ا</u>	1. 4	c-,-	116	185	7,5		10
<u> </u>	- 5		150	500	555	111	185	2 =		1.
2	2 . <	] 	415	500	555	1 . /2	185	25		1 75
1	4 <		450	- 6.	r==	1.6	155	25	Ç.	1,5
3	20.5	310	1:50	590	x50	116	185	2.5	0	1.0
!	45	366	445	-90	555	116	185	2.5		1,0
4	2 - 5	200	445	590	5.50	116	185	2.5	<u> </u>	1.0
	.: 5	70 C	445	590	545	116	185	2.5	<u> </u>	1.0
5	246	200	445	590	2 415	116	105	2.5	0	1.0
		200	445	590	540	116	155	2.5	0	1.0
6	2   5	3 C C	445	590	545	116	155	3.5	0	1. 1:
	1.5	310	455	590	550	116	190	2.5	$\mathcal{C}_{-}$	1, C
7		300	455	590	550	166	190	25	0	1,0
-	,, e	700	1.00	590	55C	116	196	25	c	1. C
8	0315	200	455	590	550	116	190	2.5	0_	1.0
	1.5	300	1150	5%	555	116	196	2.5	<u> </u>	1.0
9	5, 4 5	. C :	455	500	550	71 &	19:	5.5	•	1
· - i		35 0	450	59C	<<<	116	190	2.5	0	1.0
10	5015	: ! !	1150		r.K.C	114	19.8	96		1, 0
1		116	1	-91	470	116	192	2.5		1.0
	76.0	3 5 3	.: 25	5%	555	116	190	2,5	(	1.0
	- '	210	465	195	500	116	101	2.5	0	1.0
12	0715	300	455	590	550	116	190	2.5	0	1.0

OH IEVEL VARNISH VARNISH CLEAN LIGHT Viscosity TAN

. 17 12.8 Before 15.82 22.2 After 15.65 9.4 Change

Test Section A Tube No. 2 Oil Cons., cc 125 Cooler Sludge Clean

> Coking Tube, gm .015 Filter Sludge, gm .013

DEPOSIT RATING .16

150

#### ALCOR Inc. 10130 Jones Maltsberger Road San Antonio, Texas 78284 (S12) 349-3771

CUSTOMER Monsante Research Corporation	DATE August 19,	1981
SAMPLE: 1997658, Batch 0-79-11		
	Specification	Results
PROSICAL & CHEMICAL PROPERTIES  Neutralization Number (T.A.N.)  Viscosity @ 210°F, cs  Viscosity @ 100°F, cs  Flash Point (COC), °F  Evaporation, 6.5 Hrs @ 400°F, Wt. Loss, %	0.50 Max 3.0 Min Report 400.0 Min 30.0 Max	0.17 3.2 12.5 435 17.8
TRACE SEPIMENT Precipitation, m1/200 m1	0.005 Max	0.000
PARTICULATE CONTAMINATION  Contamination, mg/ltr	10 Max	0.02
FOAMING CHARACTERISTICS, STATIC  176°F, Volume after 30 min. aeration, ml Collapse time, sec.	100 Max 60 Max	30 ml 4.5 sec.
FOAMING CHARACTERISTICS, DYNAMIC  176°F, Volume @ 1000 cc air, cc Collapse time, min. Volume @ 1500 cc air, cc Collapse time, min. Volume @ 2000 cc air, cc Collapse time, min.  230°F, Volume @ 1000 cc air, cc Collapse time, min. Volume @ 1500 cc air, cc Collapse time, min. Volume @ 2000 cc air, cc Collapse time, min. Volume @ 2000 cc air, cc Collapse time, min.	100 Max 60 Max 150 Max 60 Max Report 60 Max 100 Max 60 Max 150 Max 60 Max Report	10 ml 50.5 sec. 10 ml 51.0 sec. 10 ml 31.6 sec. 10 ml 28.9 sec. 10 ml 50.4 sec. 10 ml 19.7 sec.
LEAD COLROSION, 1 Hr @ 325°F Wt. Change, mg/in <sup>2</sup>	6 Max	0.0
SILVER & BRONZE CORROSION, 50 Hrs @ 450 F Silver Wt. Change, mg/in <sup>2</sup> Bronze, AMS4616, mg/in <sup>2</sup>	3.0 Max 3.0 Max	-0.1 6.0

	Specification	Results
H ELASTOMER COMPATIBILITY, 168 Hrs @ 158°F		
% Swell	12.0-35.0	+26.8
EA ELASTOMED COMPATIBILITY 72 Has @ 747°E		
FA ELASTOMER COMPATIBILITY, 72 Hrs @ 347°F % Swell	2.0-25.0	+16.3
Tensile Strength, % Change	50 Max	- 4.8
Elongation, % Change	50 Max	+58.5
Hardness, No. Change	20 Max	-10
QVI ELASTOMER COMPATIBILITY, 72 Hrs @ 302°F		
% Swell	2-30 Max	+ 8.9
Tensile Strength, % Change	50 Max	-42.5
Elongation, % Change	50 Max	-25.7
Hardness, No., Change	20 Max	- 5
FS ELASTOMER COMPATIBILITY, 72 Hrs @ 302°F		
% Swell	2-25 Max	+ 5.8
Tensile Strength, % Change	50 Max	-46.1
Elongation, % Change	50 Max	-26.1
Hardness, No., Change	20 Max	- 5
DEPOSITION NUMBER (see attached data sheet)		
Deposit Number	1.5 Max	0.48
Viscosity Change, %	Report	95.2
TAN Change	Report	21.77
Oil Consumption	Report	150
ACCELERATED STORAGE STABILITY @ 230°F		
Lead Wt. Loss, mg/in <sup>2</sup>		
48 Hrs	25 Max	- 2.8 ( 1.4) <sup>*</sup>
168 Hrs	150 Max	-275.3 (132.7)*
VISCOSITY STABILITY @ 3 Hrs @ -65°F		
Original Oil, cst	17,000 Max	11,417
Viscosity Change, %	6.0 Max	0.3
3 Hrs, cst	17,000 Max	11,381
VISCOSITY STABILITY, 72 Hrs @ -65°F		
Original Oil, cst	17,000 Max	11,417
Viscosity Change, %	6.0 Max	11.5
72 Hrs, cst	17,000 Max	10,100
WORKMANSHIP	Penort	Clear
Clear, Transparent	Report	Cicai

<sup>\*</sup>Corrected results

ALCOR Inc. 10130 Jones Maltsberger Road San Antonio, Texas 78284 (512) 349-3771

SAMPLE: 1997	1997658, Batch 0-79-11	0-11								
		Specification	Results							
CORROSION & OXIDATION STABILITY, 96 Hrs @	ON STABILITY, 9	06 Hrs @ 392°F								
Steel, mg/cm2		Report	-0.010							
Silver, mg/cm <sup>2</sup>		Report	-0.040							
Aluminum, mg/cm <sup>2</sup>		Report	0.000							
Magnesium, mg/cm <sup>2</sup>	21	Report	-8,822							
I Bronze, AMSJolo, mg/cm2	mg/cm <sup>2</sup>	Report	000.0							
		Report	+0.010							
M-50 Steel, mg/cm <sup>2</sup>	2 <u>5</u>	Report	0.000							
Appearance of Metal Specimens:	Specimens:									
Pitting		Report	None							
Etching		Report	None							
Corrosion		Report	Magnesium	m						
Staining		Report	All Metals	als						
Oxidation:			16	24	40	48	64	72	88	96
<b>@</b> ,		Report								12.5
Viscosity @ 100°F, Viscosity @ 210°F, Viscosity @ 210°F	F, S Change F, Initial FF Change	Report Report Report	+6.4	+7.2	+11.2	+12.0	+24.0	+39.2	+71.2	+84.8 3.2 +46.9
Total Acid Number, Total Acid Number, Evaporation Loss,	• • •	Report Report Report	ς. σ.		t .	2.81	9.65	17.11	23.71	25.67

# MML-L-7808 DEPOSITION TEST



TEST LUBRICANT

DESIGNATION Monsanto, Sample 1997658, Batch 0-79-11

TEST NO.	4465	DATE	8/4/81	OPERATOR	Pavlicek,	Revet,	Trawick	•
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Coking Tube 590°F · Oil In 300°F · Oil Flow 300 cc/min · Air Flow 300 cc/min

			TEI	MPERATURES			]	1	T	T
HOURS	TIME	OILIN	OILOUT		NO TUBE	CAB AIR	HEATER	AIR ROTO	OIL	FILTER PRESSURE
ļ	<u> </u>	1	<b>)</b>	2	VAP SP.	<u> </u>				
START	1325	1300	125	590	545	116	190	25	0	90
	1415	300	250	540	550	11/4	190	2.5		.50
1	AC	300	450	590	550	116	190	2.5	10	90
-	15 15	1300	440	500	550	116	190	25	0	90
2	45	300	440	590	555	116	150	25	0	.90
	1615	300	445	590	560	116	190	25	٥	.90
3	15	300	4.45	590	540	116	190	2 <	0	, 5 ~
	1715	200	445	590	560	1112	190	25	0	. 55
4	15	300	450	590	565	11.5	190	3<		.90
	1815	300	450	F90	560	, <u>-</u>	190	2.5		97
5	4.5	300	445	540	555	11.5	1917	2.5		. 4/7
	1915	300	AAS	590	550	1' -	190	2.5		.40
6	<u> </u>	300	440	591)	545	,	190	75		.40
	201=	300	425	5,40	555	)/_	120	25	0	.90
7	4-5	300	450	5,5,3	560	17.5	190	25		.9,
	27:5	300	450	541)	560	115	19/2	2.5	C	.76
8	45	300	460	590	550	1/-	190	25	0	.90
	2215	200	455	590	560	116	190	2.5	0	190
9	15	200	450	590	560	116	19.5	2,5	0	,90
	23 -	300	455	590	560	116	195	2.5	0	1.00
10	15	300	-150	590	565	1/6	195	2.5	0	1.00
	24.5	300	455	F 90	560	116	195	2.5	0	1.00
11	10	3CC	450	590	565	116	195	2.5	0	1.00
	0115	300	450		560	116	195	2.5	0	110
1 2	15	300	450	590	560	116	195	2.5	0	1.0

4)	CIT	BL	<u> </u>	BL
	CLEAN	ACK COKE	MODERATE	BLACK COKE
	OIL I	LEVEL		

	Viscosity	TAN
Before	12.5	.17
After	24.4	21.94
Change	11.9	21.77

+95 2

Test Section	D	Tube No.	3
Oil Cons., cc	150	Cooler Sludg	ge Light

Coking Tube, gm_	.046
Filter Sludge, gm	.022

DEPOSIT RATING .48	

154

#### AlCOR Inc. 10130 Jones Maltsberger Road San Antonio, Texas 78284 (512) 349-3771

CUSTOMER	Monsanto Research Corporation	DATENovember.	4, 1981
SAMPLE	1997693, Batch 0-79-13		
	•	Specification	Results
PHYSICAL &	CHEMICAL PROPERTIES		
	iration Number (T.A.N.)	0.30 Max	0.25
Viscosi	ty @ 210°F, cs	3.0 Min	3.3
	ty @ 100°F, cs	Report	12.5
	oint (COC), °F	400.0 Min	440
Evapora	tion, 6.5 Hrs @ 400°F, Wt. Loss, %	30.0 Max	5.7
TRACE_SEDI		0 005 May	0.0
Precipi	tation, m1/200 ml	0.005 Max	0.0
	E CONTAMINATION		
Contami	nation, mg/ltr	10.0 Max	0.0
	ARACTERISTICS, STATIC		
176°F,	Volume after 30 min. aeration, ml	100.0 Max	10 ml
Collaps	e time, sec.	60.0 Max	1
	ARACTERISTICS, DYNAMIC		-
	ume @ 1000 cc air, cc	100.0 Max	10 ml
	lapse time, min.	60.0 Max	40.3 sec.
	ume @ 1500 cc air, cc	150.0 Max	10 ml
	lapse time, min.	60.0 Max	49.5 sec.
	ume @ 2000 cc air, cc	Report	10 ml
	lapse time, min.	60.0 Max	41.9 sec.
	ume @ 1000 cc air, cc	100.0 Max	10 ml
	lapse time, min	60.0 Max	48.6 sec.
	ume @ 1500 cc air, cc	150.0 Max	10 ml
	lapse time, min.	60.0 Max	48.2 sec.
	ume @ 2000 cc air, cc	Report	10 ml
Col	lapse time, min.	60.0 Max	52.1 sec.
	SION, 1 Hr @ 325°F		
Wt. Cha	nge, mg/in <sup>2</sup>	6.0 Max	0.4
	RONZE CORROSION, 50 Hrs @ 450°F		
	wit. Change, mg/in <sup>2</sup>	3.0 Max	0.0
Bronze,	AMS4616, mg/in <sup>2</sup>	3.0 Max	-0.4

•	Specification	Results
H ELASTOMER COMPATIBILITY, 168 Hrs @ 158°F % Swell	12.0-35.0	+27.6
FA ELASTOMER COMPATIBILITY, 72 Hrs @ 347°F % Swell Tensile Strength, % Change Elongation, % Change Hardness, No., Change	2.0-25.0 50.0 Max 50.0 Max 20.0 Max	+13.7 - 0.8 +13.8 - 5
QVI ELASTOMER COMPATIBILITY, 72 Hrs @ 302°F % Swell Tensile Strength, % Change Elongation, % Change Hardness, No., Change	2.0-30.0 50.0 Max 50.0 Max 20.0 Max	+ 5.2 -56.9 -34.3 - 5
FS ELASTOMER COMPATIBILITY, 72 Hrs @ 302°F Swell Tensile Strength, % Change Elongation, % Change Hardness, No., Change	2.0-25.0 50.0 Max 50.0 Max 20.0 Max	+ 5.3 -51.3 -40.6 - 5
DEPOSITION NUMBER (see attached data sheet) Deposit Number Viscosity Change, % TAN Change Oil Consumption	1.5 Max Report Report Report	.80 107.2 23.09 125
ACCELERATED STORAGE STABILITY @ 230°F  Lead Wt. Loss, mg/in2  48 Hrs  168 Hrs	25.0 Max 150.0 Max	<del>26.2</del> ( 12.7)* <del>395.7</del> (190.7)*
VISCOSITY STABILITY @ 3 Hrs @ -65°F Original Oil, cst Viscosity Change, % 3 Hrs, cst	17,000 Max 6.0 Max 17,000 Max	10,946 2.0 10,718
VISCOSITY STABILITY, 72 Hrs @ -65°F Original Oil, cst Viscosity Change, % 72 Hrs, cst	17,000 Max 6.0 Max 17,000 Max	10,946 17.2 9,066
WORKMANSHIP Clear, Transparent	Report	Clear

<sup>\*</sup>Corrected results

ALCON The.
10130 Jones Waltsberger Road
San Antonio, Texas 78284
(512) 849-8711

Bishod Sud	Monsanto Research Corporation	oration			DATE	Novemb	DATE November 4, 1981	12		
SAMPLE	1997693, Batch 0-79-13	20								
		Specification	Results							
h war sounds	CRANCE ON INTERNATION STABILLITY, 96 IIrs @ 392°F	6 Ilrs @ 392°F								
Since Time/Cite	รูพว/มี <b>ม</b>	Report	+0.010							
Stiver, mg/cm2	n; g / c:n; 2	Report	0.000							
With the think	Alternam, mg/cm2	Report	-0.010							
	Magnesium, mg/cm2	Report	-10.554							
Pronze	Bronze, AMS4016, mg/cm2	Report	000.0							
	m, mg/cm2	Report	000.0							
M-50 Stc	M-50 Steel, mg/cm2	Report	-0.010					`.		
Appearance	Appearance of Metal Specimens:									
200 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		Report	Ng.							
Stehing		Report	None							
Corrosion	on	Report		•	-					
Staming	£.	Report	N-50, B	Bz, Ag, Steel	r a a					
Oxidation:			16	24	40	48	64	72	88	96
Viscosity	ty @ 100°F, Initial	Report Report	+6.4	+8.0	+12.8	+15.2	+18.4	+20.0	+40.0	12.5 +13.2
Viscosity Viscosity	(a) 210 (b)	Report Report								3.3
Total A		Report	0.47	1.15	2.95	3.51	5.03	7.51	14.67	0.25 14.79
Tyanora Majara	Traporation Loss, 5	Report								3.2 0.0
	,									

# MIL-L-7808 DEPOSITION TEST



Monsanto Research, Sample 1997693, Batch 0-79-13 マイスこうじゃいごうへて DESTUNATION

TEST NO. 4472 DATE 9/22/81 OPERATOR Trawick, Revet, Pavlicek Colong Tube 590°F · Oil In 300°F · Oil Flow 300 cc/min · Air Flow 300 cc/min 15 MATERIAL SES \*F NO TUBE HE ATER AIR HOTO-METER OIL FRITER FRISSIPE 718/1 4 . R S 10175 LAP SP 0930 1 500 START . . . 500 1-5 111 11 CC 16 C 55 30 41-16 565 5 116 Ċ. 550 ج ۾ ۽ 200 1::5 5 1. 5 566 116 0 <u>. (</u> 2 36 -: ( 570 ت ت 4155 590 200 1250 126 2.00 20 **つ** 3 . . . 155 2.5 20 300 ے ترسیر 0,7 90  $\overline{\phantom{a}}$ 190 1.0 115 300 570 2.5 -- 50 11%  $\sim$ 150 2.5 1.60 570 116 1,\_ 200 م جن **-**ج بتر 1-1 1:1 -0 0 ∴ \_\_\_ 5 2000 0  $\Omega_{-}$ 450 ~ ~ ~ 115 590 25 1,0 ¿. 0 11/2  $\dot{C}$ 150 130 20 さうう 0 <u>-</u> ----1.5-, ... 2.1 7 530 500 وسبئه المريد ر. من<sup>د</sup> ا 4 C 0 ے جسر 67.0 120 2-1.6 711 300 4 10 2 -90 55 700 450 190 2.5 770 16 0 1.0 < -,-16 م میتند 570 116 10 150 2.5  $\bigcirc$ 9 10 150 X70 570 190 2.5 200 116 <u>^</u> 150 7.0 1.0 = 00 2000 1 - 12 -,5 0 \_- -130 10 ے مربر `` 1.C 2.5 115 0 ٠٠/2 .7 -2.70 سروب م ---Ć. سر، ت  $\overline{\phantom{a}}$ 50 -11 برسیرے 775 <u>C</u> -ہر بیمس J-100 150 77. 1.--\_-\_\_\_ 10 <u>- سرم</u> 12 1.1. 25 OU LEVEL

OIL	. Linkel	-		
CLFAN	HEAVY	LIGIT	VARNISH	
	1	Vieco	situ	TAN

Test Section	on A	Tube No.	1
Oil Cons.,	cc 125	Cooler Sluc	dge Light

Coking Tube, gm .076 Filter Sludge, gm .035

DEPOSIT RATING .80

TAN 158 Viscosity Before 12.5 .25 After 25.9 23.34 23.09 13.4 Change

#### ALCOR Inc. 10130 Jones Maltsberger Road San Antonio, Texas 78284 (512) 349-3771

CUSTOMER Monsanto Research Corporation	DATE November	4, 1981
SAMPLE 1997695, Batch G-79-6		
	Specification	Results
PHYSICAL & CHEMICAL PROPERTIES		
Neutralization Number (T.A.N.)	0.30 Max	0.22
Viscosity @ 210°F, cs	3.0 Min	3.3
Viscosity @ 100°F, cs	Report	12.4
Flash Point (COC), °F	400.0 Min	445
Evaporation, 6.5 Hrs @ 400°F, Wt. Loss, %	30.0 Max	5.8
TRACE SEDIMENT		
Frecipitation, m1/200 ml	0.005 Max	0.00
PARTICULATE CONTAMINATION		
Contamination, mg/ltr	10.0 Max	0.00
FOAMING CHARACTERISTICS, STATIC		
176°F, Volume after 30 min. aeration, ml	100.0 Max	10 ml
Collapse time, sec.	60.0 Max	2
FOAMING CHARACTERISTICS, DYNAMIC	-	
176°F, Volume @ 1000 cc air, cc	100.0 Max	10
Collapse time, min.	60.0 Max	7.8 sec.
Volume @ 1500 cc air, cc	150.0 Max	10
Collapse time, min.	60.0 Max	8.8 sec.
Volume @ 2000 cc air, cc	Report	10
Collapse time, min.	60.0 Max	8.5
230°F, Volume @ 1000 cc air, cc	100.0 Max	10
Collapse time, min	60.0 Max	8.9 sec.
Volume @ 1500 cc air, cc	150.0 Max	10
Collapse time, min.	60.0 Max	10.1
Volume @ 2000 cc air, cc	Report	10
Collapse time, min.	60.0 Max	10.5
LEAD CORROSION, 1 Hr @ 325°F		
Wt. Change, mg/in <sup>2</sup>	6.0 Max	0.0
SILVER & BRONZE CORROSION, 50 Hrs @ 450°F		
Silver Wt. Change, mg/in2	3.0 Max	+0.2
Bronze, AMS4616, mg/in <sup>2</sup>	3.0 Max	-0.1

	Specification	Results	
H ELASTOMER COMPATIBILITY, 168 Hrs @ 158°F	12.0-35.0	+26.0	
FA ELASTOMER COMPATIBILITY, 72 Hrs @ 347°F			
% Swell	2.0-25.0	+12.6	
Tensile Strength, % Change	50.0 Max	0.0	
Elongation, % Change	50.0 Max	+13.8	
Hardness, No., Change	20.0 Max	- 5	
QVI ELASTOMER COMPATIBILITY, 72 Hrs @ 302°F			
% Swell	2.0-30.0	+ 5.3	
Tensile Strength, % Change	50.0 Max	-41.9	
Elongation, % Change	50.0 Max	-20.7	
Hardness, No., Change	20.0 Max	- 5	
FS ELASTOMER COMPATIBILITY, 72 Hrs @ 302°F			
% Swell	2.0-25.0	+ 5.2	
Tensile Strength, % Change	50.0 Max	-48.2	
Elongation, % Change	50.0 Max	-40.6	
Hardness, No., Change	20.0 Max	- 5	
DEPOSITION NUMBER (see attached data sheet)			
Deposit Number	1.5 Max	.45	
Viscosity Change, %	Report	113.7	
TAN Change	Report	26.44	
Oil Consumption	Report	75	
ACCELERATED STORAGE STABILITY @ 230°F			
lead Wt. Loss, mg/in <sup>2</sup>			-
48 Hrs	25.0 Max	$\frac{119.9}{}$	(57.8)*
168 Hrs	150.0 Max	<del>598.7</del>	(288.5)*
VISCOSITY STABILITY @ 3 Hrs @ -65°F			
Original Oil, cst	17,000 Max	10,351	
Viscosity Change, %	6.0 Max	-0.2	
3 Hrs, cst	17,000 Max	10,331	
VISCOSITY STABILITY, 72 Hrs @ -65°F		• •	
Original Oil, cst	17,000 Max	10,351	
Viscosity Change, %	6.0 Max	-0.6	
72 Hrs, cst	17,000 Max	10,289	
WORKMANSHIP			
Clear, Transparent	Report	Clear	

<sup>\*</sup>Corrected results

ALCOR Inc.
10130 Jones Waltsberger Road
San Antonio, Texas 78284
(512) 349-3771

# MIL-L-7808H QUALITY CONFORMANCE

CHSTONER Monsanto Research Corporation	rporation		DATE		November 4, 1981	1981	
SWOLE 1997695, Batch 0-79-6	9						
	Specification	Results					
YTION STABILLITY,	96 Hrs @ 392°t						
ortori, mg/cm2	Report	+0.119					
Silver, mg/cm <sup>2</sup>	Report	-0.069					
Aluminum, mg/cm2	Report	+0.020					
	Report	-7.950					
19 Brenze, MS4609, mg/cm2	Report	+0.050					
Treation, mg/end	Report	0.000					
M-50 Steel, mg/cm2	Report	+0.108				•	
Appearance of Metal Specimens:							
Pitting	Report	Mg.					
Etching	Report	None					
Corrusion	Report						
Staining	Report	M-50, Bz, Ste	Steel, Mg				
Oxidation:		16 24	40	48	64	7.2	α α
Viscosity @ 100°F, Initial	Report						
	Report	+6.5 +7.3	+12.1	+14.5 +	+29.0	+35.5	+45.2
Viscosity @ 2'09', Initial	Report						
-	Report						
Total Acid Number, 'n. tal	Report						
Total Acid Nations, Camer	Report	0.90 i.74	3.54	4.66	12.70	15.26	15.50
Evaporation Loss, S	Report						
Sissist Volume, 'saissis	Report						

96 12.4 +47.5 3.3 +27.3 n.22 16.06 2.9

# MIL-L-7808 DEPOSITION TEST



INSTITUTE ANT PURSTONATION	Monsanto Research, Samp	24 1997695, Fatch 0-79-6
TEST NO4473	DATE _9/23/81	OPERATOR Trawick, Revet, Pavlicek

Colong	Tube 590°F	· Oil	In 300	'F · C	Dil Flow	300 c	/inin ·	Air Fl	ow 300	cc/min
1			T	Orrac HES. To Fo	. FF	AP 4.6	h- 231#	*: F FOTO.	C::	,
H . K5	1.54	01.15	CHECK	2	VAT SP	4	· 7 S	(	LEVEL.	r * 1 ' * . = E
- KTART	12=~	<u>.</u>	550	196	5:2	15	23	2.6	-	<i>y A</i>
!		3:0	300	595	12	115	156	2.5	0	1.0
1	-	(2-5	350	592	5,50	1.5	185	2_<		10
	_	- > -	370	111	6.6			5.5	_	C
2		· · ·	= 1-3	7	- 130	:		2.5		
	• • • •	=7	<b></b> 0		1.0	16	5'-1	75	0	
3		ī.c.			150	115	75	2.5	C	15
		<b>=</b>	·	: 12 C		,-	سمه تیج	<u>۔ کہ تے</u>		
4			مريد سي المريد سي	2°C	,		75	25	 	
	<del>-</del>			.0.			<u> </u>	<u>من من من</u>	·	2
5			.7.5	. '\$	-			 : :		1 . 2
-	·	,	7	: 77,7			15	سر اسری 2	<u> </u>	
t t	 - 1- 1		-7	.30°	سو .	1,5	75	5.5	(3	1.6
		775	570	: 0 ()	(40	115	15-7	2.1		
7	- ^	٠	5-1-A	<u> </u>	0	1.5	1-5	<u> </u>	<u>)</u>	3,
	-,-		500	530	7:0	. 5	<u> </u>	26	_0	1.3
5	- ·	200	31.5	500			· ɔ ᢩ -			
	2500	3.00	- < 5	<u> </u>	50 C			<u> </u>		ļ , <u>-</u>
9	•			, ,	<u>~</u>		<u> </u>		<u></u>	
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162

c:	LIFY	FL	
CLEAN	RI ACK COM	HEAVY	BLACK
Befo	re	Viscosity 12.4	1 AN . 22

	Viscosity	lan
Before	12.4	. 22
After	26.5	26.66
Change	14.1	26.44

Test Section D Tube No. 3
Oil Cons., cc 75 Cooler Sludge Clean

Coking Tube, gm .044
Filter Sludge, gm .014

F POSIT RATING .45

#### ALCOR Inc. 10130 Jones Malisberger Road San Antonio, Texas 78284 (512) 349-3771

CUSTOMER	Monsanto Research Company	DATE3/9/82_	
SAMPLE	2000248, Batch 0-79-8		
		Specification	Results
FHD SICAL &	CHEMICAL PROPERTIES		
	zation Number (T.A.N.)	0.30 Max	0.08
	y @ 210°F, cs	3.0 Min	3.2
	y 6 100°F, cs	Report	12.5
	int (COC), °F	400.0 Min	<b>43</b> 5
	ion, 6.5 Hrs @ 400°F, Wt. Loss, %	30.0 Max	15.1
TRACE SELL		0.005.14	0.000
Precipit	ation, m1/200 ml	0.005 Max	0.000
	CONTAMINATION		0.0
Contamin	ation, mg/ltr	10.0 Max	0.0
	RACTERISTICS, STATIC		••
	olume after 30 min. aeration, ml	100.0 Max	90
Collapse	time, sec.	60.0 Max	26.4 sec
	RACTERISTICS, DYNAMIC		
	me @ 1000 cc air, cc	100.0 Max	30 59.3 sec
	apse time, min.	60.0 Max	45
	me @ 1500 cc air, cc	150.0 Max	63.0 sec
	apse time, min.	60.0 Max	25
	me @ 2000 cc air, cc	Report	65.6 <b>se</b> c
Coll	apse time, min.	60.0 Max	45
	me @ 1000 cc air, cc	100.0 Max	47.6 sec
	apse time, min	66.0 Max	50
	me @ 1500 cc air, cc	150.0 Max	48.3 <b>s</b> ec
	apse time, min.	60.0 Max	50
	me @ 2000 cc air, cc	Report	61.8 sec
Co11	apse time, min.	60.0 Max	01.0
LEAD CORROS	ION, 1 Hr @ 325°F		0.3
Wt. Chan	ge, mg/in <sup>2</sup>	6.0 Max	+ 0.1
	ONZE COEROSION, 50 Hrs @ 450°F		<b>.</b>
	t. Change, mg/in <sup>2</sup>	3.0 Max	- 0.1
Bronze, AMS4616, mg/in <sup>2</sup>		3.0 Max	0.0

# MIL-L-7808H QUALITY CONFORMANCE

	Specification	Results
H ELASTOMER COMPATIBILITY, 168 Hrs @ 158°F Swell	12.0-35.0	+26.1
FA ELASTOMER COMPATIBILITY, 72 Hrs @ 547°F % Swell Tensile Strength, % Change Flongation, % Change Hardness, No., Change	2.0-25.0 50.0 Max 50.0 Max 20.0 Max	+16.2 -11.8 +10.9 -10
QVI ELASTOMER COMPATIBILITY, 72 Hrs @ 302°F % Swell Tensile Strength, % Change Elongation, % Change Hardness, No., Change	2.0-30.0 50.0 Max 50.0 Max 20.0 Max	+14.6 -19.5 -14.0 - 5
FS ELASTOMER COMPATIBILITY, 72 Hrs @ 302°F % Swell Tonsile Strength, % Change Elongation, % Change Hardness, No., Change	2.0-25.0 50.0 Max 50.0 Max 20.0 Max	+ 5.1 -50.4 -42.4 -10
Deposit Number Viscosity Change, % TAN Change Oil Consumption	1.5 Max Report Report Report	1.0 +85.6 23.04 125
ACCELERATED STORAGE STABILITY @ 230°F  Lead Wt. Loss, mg/in <sup>2</sup> 48 Hrs  168 Hrs	25.0 Max 150.0 Max	± 3.6 (+ 1.7)* -68.1 ( 32.8)*
VISCOSITY STABILITY @ 3 Hrs @ -65°F Original Oil, cst Viscosity Change, % 3 Hrs, cst	17,000 Max 6.0 Max 17,000 Max	10,914 - 1.6 10,738
VISCOSITY STABILITY, 72 Hrs @ -65°F Original Oil, cst Viscosity Change, % 72 Hrs, cst	17,000 Max 6.0 Max 17,000 Max	10,914 - 1.6 10,741
WORKMANSHIP Clear, Transparent	Report	Clear

<sup>\*</sup>Corrected results

# MIL-L-780RH QUALITY CONFORMANCE

CUSTOMER	Monsanto Research Company	отрапу			DATE	DATE 3/9/82	.82			
SAMPLE	200248, Batch 0-79-8	-8								
		Specification	Results							
CORROSION & C	CORROSION & OXIDATION STABILITY, 96 HES @	Hrs @ 392°F								
Steel, mg/cm2	/cm <sup>2</sup>	Report	-0.024							
Silver, mg/cm <sup>2</sup>	g/cm2	Report	-059							
Aluminum, mg/cm <sup>2</sup>	ng/cm <sup>2</sup>	Report	-0.001							
	, mg/cm <sup>2</sup>	Report	-6.903							
Bronze, P	$AMS4616_{3}$ mg/cm <sup>2</sup>	Report	-5.409							
	Titanium, mg/cm² M-50 Steel mg/cm²	Report	-0.010							
)		3 10/100						•		
Appearance of	Appearance of Metal Specimens:									
Pitting		Report	None							
Etching		Report	None							
Corrosion		Report	Bz, Mg							
Staining		Report	None							
Oxidation:			16	24	40	48	64	72	88	96
Viscosity	@ 100°F,	Report	i,	Ĺ	Š		c t	•		12.5
Viscosity @ 210°F,		Report	p. *	2 n +	٠,٢	+10.4	+35.8	7:15+	7.16+	+104.8 3.2
Viscosity @ 210°F, Total Acid Number.	@ 210°F, % Change d Number, Initial	Report Report								+ 62.5 0.08
rotal Acid Number,		Report	1.32	2.04	3.12	3.80	16.52	20.40	26.08	30.00
Cyaporation Loss, Clase Volume %	on Loss, %	Report								o. c
		3 10/02								

# MIL-L-7808 DEPOSITION TEST



TEST LUBRICANT					SAN ANTONIU TEXAS 1828	14 512 349-3771
DESIGNATION_	Monsanto Research,	Sample	2000248,	Batch 0-79-8		

TEST NO. 4503 DATE 2/18/82 OPERATOR Revet, Trawick, Pavlicek

Coking Tube 590°F · Oil In 300°F · Oil Flow 300 cc/min · Air Flow 300 cc/min

		1	TEM	PERATURES,	•F					T
HOURS	TIME	OIL IN	OIL OUT	COKIN	VAP SE	CAB AIR	HEATER VOLTS	AIR ROTO- METER	LEVET OIL	FILTER PRESSURE
START	1330	300	425	550	5 3 5	116	185	23	0	0.6
	14 50	360	440	= 50	645	116	150	23	0	0.9
1	30	300	430		550	116	190	ج.د	0	1.0
	15.00	300	430	590	550		190	2.2	0	1.0
2		300	430	590	<del>5</del> 55	116	197		0	
	30	i	430		555	116	190	23	0	1.0
3	11: (-1	300	430	590	555	116	190	2.3	0	1.0
	1700	300	430	500	223	116	190	2.3	0	1,0
4		300	430	590	5.55	116	190	2.3	0	1,0
·	35	300	430	590	555		190	2.3		1.6
5	18 00	300	430	590	555	116	190	2.3	c	1.0
			430	590	155	116	190	2.3	<del> </del>	1.0
<u>6</u>	19 00	300 300	430	590	550	116	190	2.3	0	1.0
	3:		430	540	550	116	190	2.3		1.0
<del></del>	2000	300	430	590	570	116	190	2.3	0	1.0
		300	430	590	570	116	190	2.3	0	1.0
8	2100	300	430	590	570	116	190	2.3		1.0
	3.0	300	430	590	<del>-</del>	116	190	2.3	0	1.0
9	2200	300		59C	570	116	190	2.5	0	1,0
	30_	300	430	59C	560		190	23		<del>                                     </del>
10	2300	<del></del>	430		550	116		· · · · · · · · · · · · · · · · · · ·	0	1, 6
10	30	300	430	591	5.50	116	196	2,3	0	1,0
11	1400	300	430	590	555	116	170	2.3		1,0
	30	300	430	590	550	116	190	2,3	0	1,0
1 2	0100	300	·	590	550	116	190	2.3		1.0
	OIL LEVEL	31.0	430	591	550	116	198	2.3		1.0

C	IL LEVE	. <b>L</b>		
CLEAN	HEAVY VARNISH	BLACK	COKE	!
Bef	ore	Visco	sity	TAN 30.

% Change

Test Section	D	Tube No	), 4	
Oil Cons., cc_	125	Cooler	Sludge_L	ight

Coking Tube, gm .092
Filter Sludge, gm .079

DEPOSIT RATING 1.00

Before 12.5 .08 166
After 23.2 23.12
Change 10.7 23.04

+85.6

# MIL-L-7808H QUALITY CONFORMANCE

CUSTOMER Monsanto Research Company	DATE3/9/82	
SAMPLE 2000247, Batch 0-79-15		
	Specification	Results
PHYSICAL & CHEMICAL PROPERTIES		
Neutralization Number (T.A.N.)	0.30 Max	0.06
Viscosity @ 210°F, cs	3.0 Min	3.2
Viscosity @ 100°F, cs	Report	12.4
Flash Point (COC), °F	400.0 Min	435
Evaporation, 6.5 Hrs @ 400°F, Wt. Loss, %	30.0 Max	17.5
TRACE SEDIMENT		0.000
Precipitation, m1/200 m1	0.005 Max	0.000
PARTICULATE CONTAMINATION		
Contamination, mg/ltr	10.0 Max	+0.01
FOAMING CHARACTERISTICS, STATIC		
176°F, Volume after 30 min. aeration, ml	100.0 Max	120
Collapse time, sec.	60.0 Max	29.1 sec
FOAMING CHARACTERISTICS, DYNAMIC		
176°F, Volume @ 1000 cc air, cc	100.0 Max	30
Collapse time, min.	60.0 Max	46 sec
Volume @ 1500 cc air, cc	150.0 Max	50
Collapse time, min.	60.0 Max	50.5 sec
Volume @ 2000 cc air, cc	Report	30 58.9 sec
Collapse time, min.	60.0 Max	30
230°F, Volume @ 1000 cc air, cc	100.0 Max	43.5 sec
Collapse time, min	766.0 Max	35
Volume @ 1500 cc air, cc	150.0 Max	43.7
Collapse time, min.	60.0 Max	30
Volume @ 2000 cc air, cc	Report	46.1
Collapse time, min.	60.0 Max	
LEAD CORROSION, 1 Hr @ 325°F		0.2
Wt. Change mg/in <sup>2</sup>	6.0 Max	<del>±0.2</del> (+0.1)*
SILVER & BRONZE CORROSION, 50 Hrs @ 450°F		
Silver Wt. Change, mg/in <sup>2</sup>	3.0 Max	-0.1 0.0
Bronze, AMS4616, mg/in <sup>2</sup>	3.0 Max	0.0

<sup>167</sup> 

# MIL-L-7808H QUALITY CONFORMANCE

	Specification	Results
H ELASTOMER COMPATIBILITY, 168 Hrs @ 158°F	12.0-35.0	+27 8
FA FLASTOMER COMPATIBILITY, 72 Hrs @ 347°F % Swell Tensile Strength, % Change Elengation, % Change Hardness, No., Change	2.0-25.0 50.0 Max 50.0 Max 20.0 Max	+15.0 -25.3 - 9.4 -10
QVI ELASTOMER COMPATIBILITY, 72 Hrs @ 302°F % Swell Tensile Strength, % Change Elongation, % Change Hardness, No., Change	2.0-30.0 50.0 Max 50.0 Max 20.0 Max	+17.2 - 9.2 - 2.9 0
FS ELASTOMER COMPATIBILITY, 72 Hrs @ 302°F % Swell Tensile Strongth, % Change Elongation, % Change Hardness, No., Change	2.0-25.0 50.0 Max 50.0 Max 20.0 Max	+ 6.6 -31.9 -25.8 -10
DEPOSITION NUMBER (see attached data sheet) Deposit Number Viscosity Change, % TAN Change Oil Consumption	I.5 Max Report Report Report	.71 +79.8 23.38 125
ACCELERATED STORAGE STABILITY @ 230°F  Lead Wt. Loss, mg/in <sup>2</sup> 48 Hrs  168 Hrs	25.0 Max 150.0 Max	11.9 (5.7)* 88.5 (42.7)*
VISCOSITY STABILITY @ 3 Hrs @ -65°F  Original Oil, cst Viscosity Change, % 3 Hrs, cst	17,000 Max 6.0 Max 17,000 Max	11,213 - 3.5 10,823
VISCOSITY STABILITY, 72 Hrs @ -65°F Original Oil, cst Viscosity Change, % 72 Hrs, cst	17,000 Max 6.0 Max 17,000 Max	11,213 -19.0 9,085
WORKMANSHIP Clear, Transparent	Report	Clear

<sup>\*</sup>Corrected results

# MIL-L-7808H QUALITY CORPORMANCE

COSTORIES FORESAMED RESEALCH COMPANY	company			DANE.	78/6/8	78,			1
SAMPLE 2000247, Batch 0-79-15	9-15								1
	Specification	Results							
CORROSION & OXIDATION STABILITY,	, 96 Hrs @ 392°F								
Corrosion:	Renort	+0 010							
Silver, mg/cm2	Report	-0.024							
Aluminum, mg/cm <sup>2</sup>	Report	+0.007							
	Report	-11.702							
9 Brenze, AMS4616, mg/cm2	Report	-1.510							
	Report	+0.008							
M-50 Steel, mg/cm2	Report	+0.024					•		
Appearance of Metal Specimens:									
Pitting	Report	None							
Etching	Report	None							
Corrosion	Report	Bz, Mg							
Staining	Report	None							
Oxidation:		16	24	40	48	64	72	88	96
ty @ 100°F,	Report				,				12.4
Viscosity @ 100%r, % Change Viscosity @ 210%r Triffial	Report	+6.5	+6.5	+6.7	+10.5	+25.6	+40.3	+75.8	+87.1
Viscosity @ 210°F, % Change	Report								+50.0
umber,	Report								0.06
Total Acid Number, Change	Report	1.18	1.68	2.84	3.42	12.34	16.32	20.88	25.42
Evaporation Loss, %	Report								6.1
Sludge, Volume, %	Report								0.0

# MIL-L-7808 DEPOSITION TEST



TEST LUBRICANT
DESIGNATION Monsanto Research, Sample 2000247, Batch 0-79-15

TEST NO. 4502 DATE 2/17/82 OPERATOR Revet, Trawick, Pavlicek

Coking Tube 590°F · Oil In 300°F · Oil Flow 300 cc/min · Air Flow 300 cc/min

	<del></del>	T	TEM	PERATURES.	• • •					
HOURS	TIME	OIL IN	OIL OUT		TUBE	CAB AIR	HEATER	AIR ROTO-	OIL	FILTER
, ACCAS	11	1	3	2	VAP SP	4	VOLTS	METER	LEVEL	PRESSURE
START	14 = 5	300	430	550	<b>630</b>	116	195	3 رو	0	1.0
	1515	200	435	550	C25	116	195	23		1.0
1	25	300	430	540	530	116	170	2.3	0	1.0
	11 5	300	43-	590	535	116	190	2,3	0	1.0
2	35	300	4-0	-50	530	116	170	22	_0_	1.0
		300	430	590	530	116	190	2.3	0	1.0
3	>5	300	430	590	530	116	190	2.3	6	1.0
	18 05	300	430	590	530	116	190	2.3	0	1.6
	35	300	420	590	.530	116	100	2.3	0	10
·	1005	300	430	590	530	116	100	23	0	1.0
<u> </u>	-; ,-	300	430	200	530	116	190	2.3	C	,0
	: cc5	300	435	590	550	16	196	23	0	1.0
6	35	300	435	590	550	116	190	2.3	0	10
	2105	300	435	590	550	116	190	2.5	0	1.0
7	3.5	30c	435	5º C	550	116	120	23	0	10
	3215	360	435	590	535	116	190	2.3	6	1.0
8	35	300	435	590	525	116	190	2.3	0	-1.0
	2305	300	435	596	525	116	190	2.3	C	1,0
9	35	300	1/35	590	530	116	190	2.3	c	1.0
	2405	300	435	590	520	116	190	2.3	0	1.0
10	35	300	435	595	-30	116	140	2.3	0	1.0
	0105	300	435	590	530	116	190	2.3	0	1.0
11	35	300	435	590	530	116	190	2.3	2	1.0
	0205	300	435	590	525	116	191	23	0	1.0
12	35	300	435	-90	525	116	150	2,3		1.0

CLEAN CLEAN BLACK COKE HEAVY VARNISH VARNISH COKE	<u> </u>	OI	L LEVEL			
				HEAVY	VARNISH	RROWN COKE

Test Section C Tube No. 3
Oil Cons., cc 125 Cooler SludgeClean

Coking Tube, gm .067
Filter Sludge, gm .037

DEPOSIT RATING .71

 Viscosity
 TAN 170

 Before
 12.4
 .06

 After
 22.3
 23.44

 Change
 9.9
 23.38

1 Change +79.8

Specification	Results
0.70	0.11
	3.1
	12.1
•	12.1 420
30.0	18.9
0.005	0.000
10.0	0.1
•	
	65.0
60.0	17.7 sec
100.0	40.0
60.0 Max	34.0 sec
150.0	80.0
60.0 Max	59.9 sec
Report	90.0
60.0 Max	63.2 sec
100.0	30.0
	60.0
	55.3 sec
	60.0
60.0 Max	56.5 sec
-	
6.0	<del>2.0</del> (1.()*
~ ^	
	-0.1
5.0	-0.1
	0.30 3.0 Report 400.0 30.0 0.005 10.0 100.0 60.0 100.0 60.0 Max 150.0 60.0 Max Report 60.0 Max

<sup>\*</sup>Corrected results

	Specification	Results
H ELASTOMER COMPATIBILITY, 168 Hrs @ 158°F		
Swell, % , max	12.0 to 35.0	+28.0
FA ELASTOMER COMPATIBILITY, 72 Hrs @ 347°F		
Swell, %, max	2.0 to 25.0	+14.9
Tensile Strength, % Change, max	50.0	+32.0
Elongation, % Change, max	50.0	+45.0
Hardness, No., Change, max	20.0	- 5.0
QVI ELASTOMER COMPATIBILITY, 72 Hrs @ 302°F		
Swell, %, max	2.0 to 30.0	+12.9
Tensile Strength, % Change, max	50.0	-45.0
Elongation, % Change, max	50.0	-14.3
Hardness, No., Change, max	20.0	-10.0
FS ELASTOMER COMPATIBILITY, 72 Hrs @ 302°F		
Swell, %, max	2.0 to 25.0	+ 6.2
Tensile Strength, % Change, max	50.0	-52.8
Elongation, % Change, max	50.0	-42.0
Hardness, No., Change, max	20.0	- 5.0
DEPOSITION NUMBER (see attached data sheet)		
Deposit Number, max	1.5	.33
Viscosity Change, %	Report	+76.9
Total Acid Number (T.A.N.)	Report	18.59
Oil Consumption, cc	Report	90
ACCELERATED STORAGE STABILITY @ 230°F		
Lead Wt. Loss, mg/in <sup>2</sup> , max		
48 Hrs, max	25.0	<del>11.8</del> ( 5.7)*
168 Hrs, max	150.0	<del>257.2</del> (123.9)*
VISCOSITY STABILITY, 3 Hrs @ -65°F		
Original, cSt, max	17,000	11,041
Viscosity Change, %, max	6.0	+1.6
3 Hrs, cSt	17,000	11,218
	<b>,</b>	,
VISCOSITY STABILITY, 72 Hrs @ -65°F	17.000	11 041
Original, cSt, max	17,000	11,041
Viscosity Change, %, max	6.0	-13.4
72 Hrs, cSt, max	17,000	9,737
WORKMANSHIP		<b>C1</b>
Clear, Transparent	Report	Clear

<sup>\*</sup>Corrected results

ALCOR Inc. 10130 Jones Maltsherger Road San Antonio, Texas 78284 (512) 349-3771

			-	96 12.1 +104.1 3.1 + 61.3 0.11
				+94.2
7/8/82				72 +60.3 21.09
ш.				64 +42.1 14.53
DATE				48 +16.5 4.83
		vo l		40 +11.6 3.15
		Results -0.006 -0.020 +0.006 -14.121 0.018 -0.006	None None Mg None	24 +7.4 1.75
		32°F		16 +6.6 0.73 5.1 0.0
Monsanto Research Company	2000273 (-0-79-10)	STABILITY, 96 Hrs @ 392°F g/cm <sup>2</sup>		Initial % Change Initial % Change Initial Change
Mo	20	cm <sup>2</sup> /cm <sup>2</sup> mg/cm <sup>2</sup> mg/cm <sup>2</sup> system mg/cm <sup>2</sup> system mg/cm <sup>2</sup> mg/cm <sup>2</sup> mg/cm <sup>2</sup>		@ 100°F, @ 210°F, I Number, on Loss,
CUSTOMER	SAMPLE	CORROSION & OXIDATION STABILITY, Corrosion: Steel, mg/cm² Silver, mg/cm² Aluminum, mg/cm² Magnesium, mg/cm² Bronze, AMS4616, mg/cm² Titanium, mg/cm² M-50 Steel, mg/cm²	Appearance: Putting LEChing Corrosion Staining	Oxidation:  Viscosity @ 100°F, Initial  Viscosity @ 210°F, Initial  Viscosity @ 210°F, Initial  Change  Evaporation Loss, %  Sludge, Volume, %

# MIL-L-1808 DEPOSITION TEST

CALCOTING.

DESIGNATION Monsanto Research, Sample: 2000273 (0-79-10)

TEST NO 4513 DATE 6/8/82 OPERATOR Trawick, Pavlicek

Coking Tube 590°F · Oil In 300°F · Oil Flow 300 cc/min · Air Flow 300 cc/min

	T	1	TEM	PERATURES	• F		1			<del></del>
HOURS	TIME	OIL IN	OILOUT	COKIN	C TUBE	CAB AIR	HEATER VOLTS	AIR ROTO. METER	OIL LEVEL	FILTER PRESSURE
		1	3	2	VAP 5P	•	VOLIS		LIVEL	PRESSURE
STAFT	18:00	300	440	590	545	116	190	2.3	0	10
	3C	300	440	590	550	116	190	2.3	6	1.0
1	19:00	300	440	590	555	116	190	2.3	0	1.0
<u> </u>	30	300	445	290	550	116	190	2.3	0_	10
2	20:00	700	440	570	245	116	190	2,3	0	1.0
	35)	300	445	590	540	116	190	2,3	0	10
3	121:00	300	445	590	545	116	190	2.3	0	1.0
	20	300	440	590	550	116	190	2,3	0	1.0
4	7200	310	<b>4</b> 35	550	545	116	190	2.3	0	1.0
<u> </u>	30	300	440	590	550	116	190	2 2	C	1.0
5	5350	300	200	590	550	116	150	्र द	0	1.0
·	30	300	225	590	555	116	195	٦.3	0	10
t.	12466	300	115	590	550	116	190	2.3	0	10
	70	300	240	595	550	116	19.2	2.3	0	10
7	0100	300	240	590	550	117.	190	2.3	0_	10
·	30.	300	440	590	550	1 %	190	و.و	0	1.5
8	10200	300	440	540	550	116	195	22	0	1.0
·	30	300	4.45	590	555	116	190	2 ته	5	10
9	10300	300	220	590	ECO	11/2	190	22	0	1.0
: 	= ~	220	245	C90	555	11/2	140	22	<u></u>	1.0
10	15460	3/6	440	5.0	5.7	16	190	22		10
	30	355	440	590	250	15	195	22	0	1.0
11	0500	300	441	590	550	11.	105	22	G	1.0
	37	300	140	590	550	1.5	16 -	2.2	0	1 :
12	0600	300	240	590	= = =	1.6	190	2.3	5	1 :
	OIL LEVEL				Test Se	ction	r	Tube N		

0.	1	
CLFAN	BLACK COKE LIGHT VARNISH	BLACK
	Viscosity	TAN

Test Section_	C	Tube N	03	
Oil Cons., cc	90	Cooler	Sludge	Clean

Coking Tube, gm .031 Filter Sludge, gm .021

DEPOSIT RATING .33\_\_\_\_\_

Viscosity TAN 174

Before 12.1 .11 174

After 21.4 18.7

Change 9.3 18.59

% Change +76.9

CUSTOMER Monsanto Research Company	DATE July 7, 19	82
SAMPLE 2000268 (0-79-14)		
	Specification	Results
PHYSICAL & CHEMICAL PROPERTIES	0.70	0.74
Neutralization Number (T.A.N.), max	0.30 3.0	0.14
Viscosity @ 210°F, cSt, min Viscosity @ 100°F, cSt		11.7
Flash Point (COC), °F, min	Report 400	395
Evaporation, 6.5 Hrs @ 400°F,	400	393
Wt. Loss, %, max	30.0	17.5
TRACE SEDIMENT		
Precipitation, ml/200 ml, max	0.005	0.000
PARTICULATE CONTAMINATION		
Contamination, mg/ltr, max	10.0	0.0
FOAMING CHARACTERISTICS - Static		
Volume after 30 min aeration, ml, max	100	35
Collapse time, sec, max	60	21.3
FOAMING CHARACTERISTICS - Dynamic 176°F		
Volume @ 1000 cc air, cc, ml, max	100	30
Collapse time, min	60 Max	38.0 sec
@ 1500 cc air, cc ml, max	150	50
Collapse time, min	60 Max	45.9 sec
@ 2000 cc air, ml	Report	60 57 5 nos
Collapse time, min	60 Max	57.5 sec
Volume @ 1000 cc air, cc, ml, max	100	10
Collapse time, min	60 Max	36.3 sec
@ 1500 cc air, cc, ml, max	150	30
Collapse time, min	60 Max	40.1 sec
@ 2000 cc air, ml	Report	30
Collapse time, min	60 Max	41.0 sec
LEAD CORROSION, 1 Hr @ 325°F		0.0
Wt. Change, mg/in <sup>2</sup> , max	6.0	0.9 (0.5)*
SILVER & BRONZE CORROSION, 50 Hrs @ 450°F	7.0	0.0
Silver Wt. Change, mg/in <sup>2</sup> , max	3.0	0.0
Bronze, AMS4616, mg/in2, max	3.0	-0.1

<sup>\*</sup>Corrected results

U CLACTOMED COMPATIBILITY - M.O. H O. MCO.	Specification	Results
H ELASTOMER COMPATIBILITY, 168 Hrs @ 158°F Swell, %, max	12.0 to 35.0	+26.5
FA ELASTOMER COMPATIBILITY, 72 Hrs @ 347°F		
Swell, 6, max	2.0 to 25.0	+13.7
Tensile Strength, % Change, max	50.0	-22.9
Elongation, % Change, max	50.0	-12.5
Hardness, No., Change, max	20.0	-10.0
QVI ELASTOMER COMPATIBILITY, 72 Hrs @ 302°F		
Swell, %, max	2.0 to 30.0	+17.9
Tensile Strength, % Change, max	50.0	-34.0
Elongation, % Change, max	50.0	+ 5.7
Hardness, No., Change, max	20.0	-10.0
FS ELASTOMER COMPATIBILITY, 72 Hrs @ 302°F		
Swell, 6, max	2.0 to 25.0	+ 6.9
Tensile Strength, % Change, max	50.0	-55.4
Elongation, % Change, max	50.0	-28.8
Hardness, No., Change, max	20.0	-10.0
DEPOSITION NUMBER (see attached data sheet)		
Peposit Number, max	1.5	1.14
Viscosity Change, %	Report	42.7
Total Acid Number (TAN)	Report	14.0
Oil Consumption, cc	Report	125
ACCELERATED STORAGE STABILITY @ 230°F		
lead Wt. Loss, mg/in <sup>2</sup>		
48 Hrs, max	25.0	<del>14.5</del> ( 7.0)*
168 Hrs, max	150.0	<del>468.5</del> (225.7)*
VISCOSITY STABILITY - 3 Hrs @ -65°F		
Original, cSt, max	17,000	9,868
Viscosity Change, %, max	6.0	-0.4
3 Hrs, cSt, max	17,000	9,825
VISCOSITY STABILITY, 72 Hrs @ -65°F		
Original, cSt, max	17,000	9,868
Viscosity Change, %, max	6.0	- 7.5
72 Hrs, cSt, max	17,000	9,126
WOREMANSHIP		
Clear, Transparent	Report	Clear

<sup>\*</sup>Corrected results

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II. July / July	
WI	
Monsanto Research Company	(11 01 0) 0,0000
CUSTOMER	,

CUSTOMER	Monsanto Research Company	, A			DATE July 7, 1982	11y 7, 198	2	
SAMPLE:	2000268 (0-79-14)							
CORROSION &	CORROSION & OXIDATION STABILITY, 96 Hrs	Results	<b>ν</b> .}					
Corrosion:	o/cm2	+0.008						
Silver, mg/cm <sup>2</sup>	mg/cm <sup>2</sup>	-0.016						
Aluminum, mg/cm <sup>2</sup>	mg/cm²	-15.764						
Magnesiu Bronze,	Magnesium, mg/cm² Bronze, AMS4616, mg/cm²	0.000						
	Titanıum, mg/cm² M-50 Steel, mg/cm²	-0.004	_					
22. Appearance:		;						
		None						
Etching		None						
Corrosion	nc	age N						
Staining	<b>D</b>				:	;	1	0
Oxidation:		16	24	40	48	64	7/	00
Viscosit	Viscosity @ 100°F, Initial	\$\$ <del>\$</del>	+12.0	+19.7	+23.9	+40.2	+47.9	+60.7
Viscosit	Viscosity @ 210°F, Initial % Change							
Total Ac	Total Acid Number, Initial Change	1.82	2.56	6.64	7.72	11.80	13,54	16.46
Evaporat Sludge,	Evaporation Loss, % Sludge, Volume, %	2.3					-	

96 11.7 +63.2 3.1 -35.4 0.14

# MIL-L-7808 DEPOSITION TEST



TEST LUBRICANT

DESIGNATION Monsanto Research, Sample: 2000268 (0-79-14)

TEST NO 4511 DATE 6/2/82 OPERATOR Revet, Trawick, Pavlicek

Coking Tube 590°F · Oil In 300°F · Oil Flow 300 cc/min · Air Flow 300 cc/min

i	1		TEN	MPERATURE	5. *F		T	T	Υ	T
HOURS.	TIME	OIL IN	OIL OUT	(OF)	VAP SP	CAB AIR	HEATER VOLTS	AIR ROTO- METER	TEAET	FILTER PRESSURE
START	1 -37.7	300	400	590	545	115	195	23	C	1.0
		3-6	425	۲٥,5	565	115	190_	2.3		1.0
1	-1	200	4-5	140	560	115	15 ~	2.3	C	1.0
[ 		300	421		17.	,	-,	23		1.5
2	1	300	420	5:0	565	115	190	2.3	0	1.0
	1.1/	300	420	100	565	100	190	2.3	0	1.0
3	1.218	300	1/20	150	565	115	190	2.3	0	1.0
· ·	Ur	700	420	590	565	115	190	2.3	_	1.0
4	315	7-20	427	-790	535	115	190	2.3	0	1.0
	41	300	420	590	560	115	19c	2.3	0	1.0
5	10.5	320	120	590	760	115	190	2.3	0	10
	115	700	425	590	560	115	190	2.3	0	1.0
6	515	300	425	500	560	115	100	ã. 3		4.0
	<u> </u>	3cc	425	100	730	1	190	23	_0	1.0
7	1315	300	425	1.90	570	115	190	23	0	1.6
	45	370	425	590	550	115	190	23	<i>(</i> )	1.0
8	1715	500	425	590	555	115	190	2.3	0	1,0
	ur	111	2177	5 816	500	115	100	2 =	٠ ع	1.0
9	915	300	470	590	550	115	190	2.3	$\sim$	1.0
	45	200	:150	£90	550	115	190	2.7	0	1.2
10	1915	200	425	590	c	115	190	2.3	0	1.0
	45	300	42=	540	550	115	190	2.3	0	1.2
11	2015	300	420	590	250	115	190	2,3	2)	1.0
	45	360	425	590	-50	115	170	2.3	0	1.0
12	2115	200	1125	5.90		115	190	2.3	<i>C</i> ,	1,0
		•								

V		Vie	comity	TAN
	GLEAN	BLACK	COKE	
	0.2	1		
_	OII.	LEVEL		

Test Section	С	Tube No.	4
Oil Cons., cc_	125	Cooler Sluc	ige Clean

Coking Tube, gm .113
Filter Sludge, gm .014

DEPOSIT RATING 1.14

 Viscosity
 TAN 178

 Before
 11.7

 After
 16.7

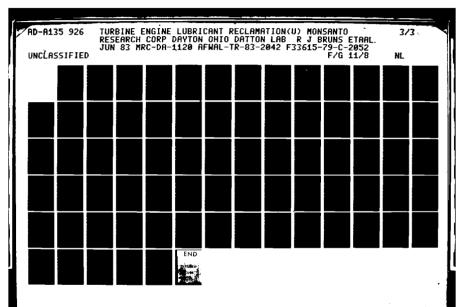
 Change
 5.0

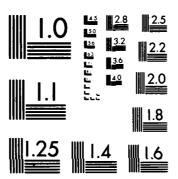
 14.00

# MIL-L-7808H QUALITY CONFORMANCE

CUSTOMER	Monsanto Research Company	DATE 7/8/82	
SAMPLE	2000275 (0-79-12)		
DING STATE		Specification	Results
	G CHEMICAL PROPERTIES Lization Number (T.A.N.), max	0.30	0.11
	ity @ 210°F, cSt, min	3.0	3.2
Viscosi	ity @ 100°F, cSt	Report	13.3
Flash I	Point (COC), °F, min	400.0	430
Evapora	ation Loss, 6.5 Hrs @ 400°F,		
	Loss, %, max	30.0	16.7
TRACE SEDI	IMENT		
Precip	itation, m1/200 ml, max	0.005	0.000
PARTICULS"	TE CONTAMINATION		
Contam:	in tion, mg/ltr, max	10.0	0.0
FOAMING CH	HARACTERISTICS - Static		
Volu	ume after 30 min aeration, ml, max	100.0	60.0
	lapse time, sec, max	60.0	24.6
FOAMING CH	HARACTERISTICS - Dynamic		
	ame @ 1000 cc air, cc, ml, max	100.0	10.0
	lapse time, min	60.0 Max	37.7 sec
	0 1 00 cc air, cc, ml, max	150.0	30.0
Col	lapse time, min	60.0 Max	61.5 sec
	@ 2000 cc air, cc, ml	Report	30.0
Co1: 230°F	lapse time, min	60.0 Max	67.2 sec
	ume @ 1000 cc air, cc, ml, max	100.0	10.0
	lapse time, min	60.0 Max	41.6 sec
COT	@ 1500 cc air, cc, ml, max	150.0	30.0
Co.1	lapse rime, min	50.0 Max	61.5 sec
COT	@ 2000 cc air, cc, ml	Report	30.0
Col	lapse time, min	60.0 Max	56.8 sec
LEAD CORR	OSION, 1 Hr @ 325°F		
Wt. Ch	ange, mg/in2, max	6.0	th4 (0.2)*
SILVER &	BRONZE CORROSION, 30 Hrs @ 450°F		
Silver	Wt. Change, mg/in2, max		- 0.2
	, AMS4616, mg/in <sup>2</sup> , max	3.0	$\zeta^*$ , $\Omega$

<sup>\*</sup>Corrected results





MICROCOPY RESOLUTION TEST CHART
NATIONAL BUREAU OF STANDARDS-1963-A

	Specification	Results
H ELASTOMER COMPATIBILITY, 168 Hrs @ 158°F		
Swell, %, max	12.0 to 35.0	+ 27.8
. A FLASTOMER COMPATIBILITY, 72 Hrs @ 347°F		
Swell, %, max	2.0 to 25.0	+ 14.8
Tensile Strength, % Change, max	50.0	- 27.7
Elongation, % Change, max	50.0	+ 34.5
Hardness, No., Change, max	20.0	5.0
QVI ELASTOMER COMPATIBILITY, 72 Hrs @ 302°F		
Swell, %, max	2.0 to 30.0	+ 9.9
Tensile Strength, % Change, max	50.0	- 45.0
Elongation, % Change, max	50.0	- 20.0
Hardness, No., Change, max	20.0	- 10.0
FS ELASTOMER COMPATIBILITY, 72 Hrs @ 302°F		
Swell, %, max	2.0 to 25.0	+ 6.0
Tensile Strength, % Change, max	50.0	- 52.8
Elongation, % Change, max	50.0	- 47.8
Hardness, No., Change, max	20.0	- 5.0
DEPOSITION NUMBER (see attached data sheet)		
Deposit Number, max	1.5	.37
Viscosity Change, %	Report	+218.9
Total Acid Number (T.A.N.)	Report	20.21
Oil Consumption, cc	Report	75
ACCELERATED STORAGE STABILITY @ 230°F		
Lead Wt. Loss, mg/in <sup>2</sup>		
48 Hrs, max	25.0	1.3 ( 0.6)*
168 Hrs, max	150.0	<del>245.4</del> (118.3)*
VISCOSITY STABILITY, 3 Hrs @ -65°F		
Original, cSt, max	17,000	12,612
Viscosity Change, %, max	6.0	+ 0.2
3 Hrs, cSt, max	17,000	12,643
VISCOSITY STABILITY, 72 Hrs @ -65°F		
Original, cSt, max	17,000	12,612
Viscosity Change, %, max	6.0	- 5.4
72 Hrs, cSt, max	17,000	11,925
WORKMANSHIP		
Clear, Transparent	Report	Clear

<sup>\*</sup>Corrected results

CUSTOMER	Monsanto Research Company				DATE	Jul	July 8, 1982	2	
SAMPLE	2000275 (0-79-12)								
o a moradado	6 2-11 30 VIIII GATS MOTTAGIVE	3010E	Results	νı					
Corrosion:	Corrosion:	396 F							
Steel, mg/cm2	/cm <sup>2</sup>		-0.010						
Silver, mg/cm <sup>2</sup>	g/cm <sup>2</sup>		-0.012						
Aluminum, mg/cm2	mg/cm <sup>2</sup>		0.000						
Magnesium	, mg/cm <sup>2</sup>		-10.798						
Bronze, A	Bronze, AMS4616, mg/cm <sup>2</sup>		-0.026						
Titanium,	mg/cm <sup>2</sup>		-0.008						
M-50 Steel, mg/cm2	1, mg/cm2		-0.006						
18									
νbη			:						
Pitting			None				•		
Etching			None						
Corrosion			Σ Ω						
Staining			None						
Oxidation:	•	16	24	40	48	64	72	88	96
Viscosity	Viscosity @ 100°F, Initial								13.3
		+1.5	+4.5	+8.3	+10.5	+12.8	+15.8	+40.6	+48.9
Viscosity	Viscosity a 210°F, Initial								3.2
Total Acie	Total Acid Number Initial								0.11
	Change	0.67	1.47	2.63	3.03	4.05	5.93	16.61	18.41
Evaporation Loss,	0.	3.4						<del></del> -	
Sludge, Volume, %	olume, %	0.0							

# MIL-L-7808 DEPOSITION TEST



Tube No.

.37

Coking Tube, gm .035

Filter Sludge, gm

DEPOSIT RATING

Cooler Sludge Lt. Sludge

TEST LUBRICANT

Monsanto Research, Sample: 2000275 (0-79-12) DESIGNATION \_\_\_

TEST NO 4514 DATE 6/9/82 OPERATOR Revet, Pavlicek, Trawick

Coking Tube 590°F · Oil In 300°F · Oil Flow 300 cc/min · Air Flow 300 cc/min

		Τ	TEM	PERATURES	• • •		<del></del>		<u> </u>	
HOURS	TIME	OIL IN	OIL OUT		C TUBE	CAB AIR	HEATER	AIR POTO	OIT	FILTER
		1	3	2	VAP SP.	4	VOLTS	METER	LEVEL	PRESSURE
START	<u> </u>	300	445	590	575	115	150	2.5	0	10
	<u> </u>	7 ^ 7	440	500	E60	115	150	23	0	10
1	5500	300	240	555	E + C.	1: 5	150	2 3	0	10
;	7.10	1 = cc	445	590	170	1.5	190	23	c	1.0
2	10000	300	UUK	590	570	115	180	3	C	1.0
	3 /	Foo	445	590	= 70	115	180	2.3	0	1. c
3	1500	300	435	590	570	115	180	2.3	0	1.0
	30	300	445	590	565	115	185	23	0	1.0
4	1150	300	445	590	50	115	185	2.3	0	1,5
	3 C	300	450	500	570	115	185	2.3	0	1.0
5	1=00	300	450	240	580	1.5	185	2.3	0	10
	<b>\$</b>	३ ६ ६	450	590	570	115	185	23	0	1.0
6	1300	310	450	571	576	115	185	2.3	<u>`</u>	1.5
	3.0	3 5 6	450	590	570	115	185	2,3	C	1.0
7	1200	300	450	500	570	1.5	18-	2.3	c	1.6
	<b>2</b> 2	200	450	190	570	115	185	2.3	O,	1.0
8	1500	305	445	590	560	115	185	2.3	0	1.0
	30	300	445	190	560	115	185	2,3	C	1.0
9	1200	300	445	590	560	11.5	185	2.3	0	1.0
	<u> </u>	300	445	590	560	11.5	185	2.3	0	1.0
10	1755	300	445	590	560	115	185	2.3	0	1.0
1	3,0	300	445	590	560	115	185	2,3	0	1.0
11	18,00	300	445	590	545	115	185	2.3	0	1,0
	2,0	200	450	590	565	11.5	185	2.3	6	1.0
12	1600	360	450	590	565	115	185	2.3	0	1.0

OIL	LEVEL		_
CLEAN	BLACK COKE	LIGIT Varnish	BLACK
	v	iscosity	TAN

Before

% Change

12.7

After 40.5 Change 27.8

+218.9

.11 182

Test Section

Oil Cons., cc 75

20.32 20.21

# MIL-L-7808H QUALITY CONFORMANCE

CUSTOMER	Monsanto Research Company	DATE 7/7	/82
SAMPLE	2000267 (0-79-7)		·····
		Specification	Results
	CHEMICAL PROPERTIES		
	ization Number (T.A.N.), max	0.30	0.11
Viscosi	ty @ 210°F, cSt, min	3.0	3.1
Viscosi	ty @ 100°F, cSt	Report	12.1
	oint COC, cSt, °F, min	400	400
Evapora	tion Loss, 6.5 Hrs @ 400°F,		• • •
	Loss, %, max	30.0	14.0
TRACE SEDI	MENT		
	tation, ml/200 ml, max	0.005	0.0
מאסמיורווו אמי	E CONTANTNATION		
	E CONTAMINATION max	10.0	0.0
			•••
FOAMING CHA	ARACTERISTICS - Static		
		100	<b>5</b> 0
	me after 30 min aeration, ml, max	100	50
COLIA	apse time, sec, max	60	12.0
FOAMING CH	ARACTERISTICS - Dynamic		
176°F			
	me @ 1000 cc air, cc, max	100	10
	apse time, min	60 Max	17.9 sec
	@ 1500 cc air, cc, max	150	10
Co11;	apse time, min	60 Max	31.5 sec
<b>*</b> = - ·	@ 2000 cc air, cc	Report	20
Coll:	apse time, min	60 Max	48.6 sec
230°F	apse came, man	oo raa	70.0 300
	me @ 1000 cc air, cc, max	100	10
	apse time, min	60 Max	33.2 sec
<b>▼</b>	@ 1500 cc air, cc, max	150	10
Colla	apse time, min	60 Max	40.6 sec
VV	@ 2000 cc air, cc,	Report	20 20
Co11:	apse time, min	60 Max	74.ó sec
(011	apse time, min	OU MAX	/4.0 566
LEAD CORROS	SION, 1 Hr @ 325°F		
Wt. Char	nge, mg/in², max	6.0	1-4 (+0.3)
CTIVED C DI	RONZE CORROSION, 50 Hrs @ 450°F		
SILVER & DI	MONEE COMMODICIN, SO MIS 6 450 I		
	Wt. Change, mg/in <sup>2</sup> , max	3.0	0.0

	Specification	Results
H ELASTOMER COMPATIBILITY, 168 Hrs @ 158°F		
Swell, %, max	12.0 to 35.0	+27.3
		2, 13
FA ELASTOMER COMPATIBILITY, 72 Hrs @ 347°F		
Swell, %, max	2.0 to 25.0	+12.6
Tensile Strength, % Change, max	50.0	+17.1
Elongation, % Change, max	50.0	+20.3
Hardness, No., Change, max	20.0	5.0
QVI ELASTOMER COMPATIBILITY, 72 Hrs @ 302°F		
Swell, %, max	2.0 to 30.0	+16.0
Tensile Strength, % Change, max	50.0	-35.5
Elongation, % Change, max	50.0	+ 5.7
Hardness, No., Change, max	20.0	15
FS ELASTOMER COMPATIBILITY, 72 Hrs @ 302°F		
Swell, %, max	2.0 to 25.0	+ 5.0
Tensile Strength, % Change, max	50.0	-62.7
Elongation, % Change, max	50.0	-43.9
Hardness, No., Change, max	20.0	-10.0
	20.0	10.0
DEPOSITION NUMBER (see attached data sheet)		
Deposit Number, max	1.5	.32
Viscosity Change, %	Report	+116.5
Total Acid Number	Report	26.97
Oil Consumption, cc	Report	100
ACCELERATED STORAGE STABILITY @ 230°F		
Lead Wt. Loss, mg/in <sup>2</sup>		
48 Hrs, max	25.0	<del>-6.2</del> ( 3.0)*
168 Hrs, max	150.0	<del>194.0</del> (93.5)*
VISCOSITY STABILITY, 3 Hrs @ -65°F		
Original, cSt, max	17,000	10,566
Viscosity Change, %, max	6.0	+ 0.4
3 Hrs, cSt, max	17,000	10,612
o max	27,000	20,022
VISCOSITY STABILITY, 72 Hrs @ -65°F		
Original, cSt, max	17,000	10,566
Viscosity Change, %, max	6.0	-11.7
72 Hrs, cSt, max	17,000	9,326
WORKMANSHIP		
Clear, Transparent	Report	Clear
•		

<sup>\*</sup>Corrected results

	CUSTOMER	Monsanto Research Company			DATE		July 7, 1982			ļ
	SAMPLE	2000267 (0-79-7)								1
			Results							
	CORROSION &	CORROSION & OXIDATION STABILITY, 96 Hrs @ 35	392°F							
	Steel 3	9/cm2	-0.007							
	Silver	6/ c	690.0-							
	Aluminum	. mg/cm <sup>2</sup>	-0.007							
	Magnesiu	m, mg/cm²	-6.774							
	Bronze,	AMS4616, mg/cm <sup>2</sup>	+0.059							
	Titanium	, mg/cm <sup>2</sup>	-0.006						•	
1	M-50 Ste	M-SO Steel, mg/cm <sup>2</sup>	+0.020							
85	A 22 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2									
	appearance. Pitting		None							
	Etching		None							
	Corrosion	_	Mg							
	Staining		Bz							
	Oxidation:		16	24	40	48	64	72	88	96
	Viscosit	Viscosity @ 100°F, Initial	,	(	(				9	12.1
	Viscosit	<pre>% Change Viscosity @ 210°F, Initial % Change</pre>	+7.4	6.6 <del>+</del>	+15.7	+19.0	+23.9	+37.2	+57.0	+63.6 3.1 +38.7
	Total Ac	Total Acid Number, Initial	1,19	1 79	7.87	4.27	8	14.05	14 08 1 . 20 53	0.11
	Evaporat Sludge,	Evaporation Loss, % Sludge, Volume, %	1.3	•		•			07.	10.73

# MIL-L-7808 DEPOSITION TEST

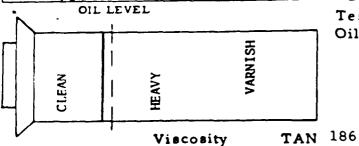


TEST LUBRICANT					PMM WILLIAM IE VW2 18584 - 3151349-3111
1231 Et Internet	Maranaka Basasak	Campla	2000267	(0 70 7)	
DESIGNATION	Monsanto Research,	Sample.	2000207	10-/9-/1	

TEST NO. 4510 DATE 6/1/82 OPERATOR Revet, Trawick, Pavlicek

Coking Tube 590°F · Oil In 300°F · Oil Flow 300 cc/min · Air Flow 300 cc/min

	1		TEN	APERATURES	. *F		]	· ·		Γ
HOURS	TIME	OIL IN	OIL OUT		C TUBE	CAB AIR	HEATER VOLTS	AIR ROTO-	LEVEL	FILTER PRESSURE
-	!	1	,	2	VAP SP.	<u> </u>	ļ <u>.</u>			ļ <u>.</u>
START	14:15	1.15	445	2.00	540	· ;-	1.70 %	6.5	3	1.0
-	45	365	4115	533	550	116	100	2.3	0	1.0
1	سترا سي ا	300	445	590	550	116	120	2.3	6	1.0
	4,5			) /u -	27		۵٠.	5.7	<u>^</u>	10
2	1115	200	4115	120	-770	114	4.5 )	2 3	-	10
	-/-	500	440	590	5.50	1/-	190	2.3	ε	1.0
3	7,5	1300	440	590	545	116	190	2.3	0	1.0
	115	300	445	590	550	116	195	2.7	0	10
4	1516.	700	446	5-63	545	116	190	2.3	0	1.0
	42.27	30C	440	-41		116	197	0,5	0	10
5	15,5	355	11:5	5-20	= :/ =	114	150	2.3	٥	1.2
	: -	300	440	55/	540	116	196	2,7	0	7.6
6	5615	5, 5, 5	440	700	740	11/2	19:	2.3	( .	1.0
		200	111.	591	225	116	190	2,3	<u> </u>	1.0
7		300	445	590	540	116	195	2.3	0	1.0
		30 C	460	590	560	116	195	2,3	<i>c</i> ``	1.5
8	2015	300	4:0	590	560	116	155	2. 7		10
		7	460	343	5+0	116	195	24		1.
9	63/5	315	460		560	11/-	195	<b>ラン</b>		11
	4.5	7 - 1	460	,	560	116	195	フィ		1.1
10	24/5	7	460	•	560	11/2	195	. ,		1.1
		-	A55		560	1	195	٠, ٦		1. ]
11	0115	:	455	e	-60	1 -	195	7		1.1
		7	10		565	1.6	195	7. 1.		1.)
. 12			110		-1.5	1 %	195	2.7	1,	1.



Test Section_	D	Tube No.	1
Oil Cons., cc	100	Cooler Sluc	ige Light

Coking Tube, gm .030
Filter Sludge, gm .018

DEPOSIT RATING .32

Before 12.1 .11
After 26.2 27.08
Change 14.1 26.97



September 23, 1982

Mr. Richard Bruns Monsanto Research Company 1515 Nicholas Road Dayton, OH 45407

SAMPLE: 1997659 (0-79-9), 1997658 (0-79-11), 19976693 (0-79-13), 1997695 (0-79-6)

Richard, following are the test results obtained from your sample as requested with the 703 added per your letter of August 23, 1982.

		Res	sults	
	1997659 (0-79-9)	1997658 (0-79-11)	1997693 (0-79-13)	1997695 0-79-6)
ACCELERATED STORAGE Lead Wt. Loss, mg/in <sup>2</sup>	(95.5)*	(102.4)*	(3.0)*	(30.6)*
48 Hrs	198.2	212.6	6.2	63.5
168 Hrs	293.4	729.5	178.0	283.4
	(141.4)*	(351.5)*	(85.7)*	(136.6)*

Please let us know if we may be of further service or answer any questions you might have regarding these results. Best regards.

Sincerely,

PHILLIP O. FRUIN

Manager, Testing Services

POF:b Enclosures

\*Corrected results

ALCOR Inc. 10130 Jones Maltsberger Road San Antonio, Texas 78284 (512) 349-3771

DATE December 29, 1981		2000227 2000227 2000227 2000227 2000227 A B C D E F F			$(+2.1)^{*}$ $(21.7)^{*}$ $(+8.3)^{*}$ $(+0.4)^{*}$ $(0.5)^{*}$ $(3.0)^{*}$	87.3 -220.8 -36.6 -104.7 -98.7 -99.7 (42.1)* (106.4)* (17.6)* (50.5)* (47.6)* (48.1)*
ch Corporation	D, E, F	Specification			25 mg Мах	150 mg Max
Monsanto Research Corporation	2000227A, B, C,		ABILITY	Lead Weight loss, mg/in <sup>2</sup>	48 Hours	168 Hou: .
CUSTOMER	SAMPLE		STORAGE STABILITY	Lead M	4	16

\*Corrected results

# APPENDIX F

SAMPLE CALCULATIONS

NaOH Addition

Ba(OH)<sub>2</sub>·H<sub>2</sub>O Treatment

### NaOH CALCULATION FOR DISTILLATION

Acid # of 1.3 is defined as milligrams of potassium hydroxide, that is required to neutralize all acidic constituents in 1 g sample or 0.0013 g KOH/g of oil.

$$\frac{\text{0.0013 g KOH}}{\text{g-oii}} \times \frac{40.1 \text{ g-mole NaOH}}{\text{56.1 g-mole KOH}} \times \frac{\text{0.932 g-oil}}{\text{mL}} \times \frac{3,785.3 \text{ mL}}{\text{gal}} \times \frac{\text{gal}}{7.77 \text{ # oil}} \times$$

N(number of lbs of used oil in batch) = 0.422 g of NaOH t dd to batch

# Ba(OH)2·H2O CALCULATION

$$\frac{\text{0.0013 g KOH}}{\text{g-oil}} \times \frac{189 \text{ g-mole Ba}(\text{OH})_2 \cdot \text{H}_2\text{O}}{\text{56.1 g-mole KOH}} \times \frac{\text{0.932 g oil}}{\text{oil}} \times \frac{3,785.3 \text{ mL}}{\text{gal}} \times \frac{\text{gal}}{7.77 \text{ # oil}} \times \frac{\text{gal}}{\text{matherapy of the mole KOH}} \times \frac{\text{O.932 g oil}}{\text{oil}} \times \frac{\text{3.785.3 mL}}{\text{gal}} \times \frac{\text{gal}}{7.77 \text{ # oil}} \times \frac{\text{gal}}{\text{oil}} \times$$

N(number of lbs of used oil in batch) = 1.989 g of  $Ba(OH)_2 \cdot H_2O$  to add to batch but the minimum amount is 0.77% by wt.

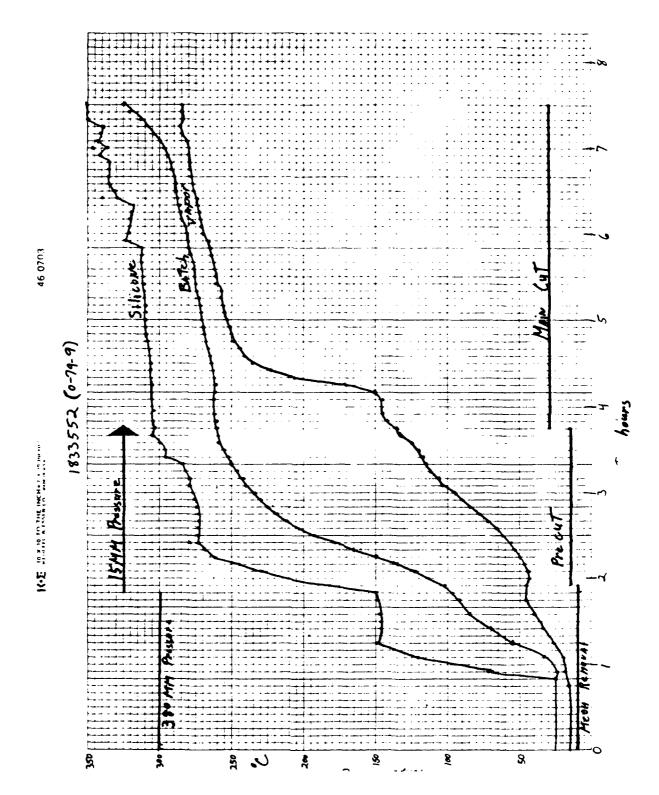
# APPENDIX G

# SELECTED LARGE SCALE DISTILLATION DATA

# LARGE SCALE DISTILLATION DATA

Date			
distilled	NBP	Used oil	Comments
2/17/81	1833517	Virgin basestock	
2/24/81	1833520	0-79-01	
2/26/81	1833523	0-79-02	(No NaOH used in distillation)
3/02/81	1833526	0-79-02	
3/04/81	1833529	0-79-02	
3/06/81	1833534	0-79-03	
3/10/81	1833536	0-79-08	
3/12/81	1833543	0-79-07*	
3/17/81	1833546	0-79-11*	
3/19/81	1833548	0-79-12	
3/30/81	1833550	0-79-14	
4/01/81	1833552	0-79-09*	
4/03/81	1833555	0-79-04	
4/07/81	1833557	0-79-15	
4/09/81	1833560	0-79-06*	
6/18/81	1997632	0-79-13*	
6/15/81	1997633	0-79-10*	
6/23/81	1997634	0-79-15*	Redistilled from above
6/25/81	1997636	0-79-12*	Redistilled from above
6/30/81	1997641	0~79-08*	Redistilled from above
7/02/81	1997644	0-79-14*	Redistilled from above
7/22/82	2000294A	0-82-06	
8/02/82	2000295B	0-82-06	Redistilled from above
7/27/82	2000296	0-82-09	
8/04/82	2000298	0-82-07	No NaOH used in distillation
8/16/82	2276303	0-82-04	

<sup>\*</sup>Distillate totally reclaimed and Mil-L-7808H tested.



ватсн	1831552 (1.74-9)
	41.181
ACID N	10.

### DISTILLATION

INITIALS	DATE		
		1.	Turn on vacuum jet.
		2.	Close valve, bottom of 2F. Pull vacuum.
		3.	Check to see if filter (50 micron) is in, inline filter on drum suction tube.
		4.	Place drum on scale, suck into 2F 155 lbs of used oil (20 gals).
* * * * * * * * * * * * * * * * * * *			oil + container (initial)  oil + container (final)  oil weight
	——————————————————————————————————————	5.	Let oil sit over night in 2F with no agitation, open sample valve for vent.
<del> </del>		6.	Vent vacuum lines.
		7.	Turn off vacuum jet.
60.	<del>- 7</del>	8.	Check batch temperature, should be aroun 20°C. Draw 1/2 gallon oil out of bottom of 2F, if no phase separation, return oil to 2F.
1/-		9.	Start agitator on 2F, make sure batch temperature is below $30^{\circ}\text{C}$ before adding methanol, add gallon of alcoholic NaOH (see supervisor).
RA	<del></del>	10.	Let NaOH oil stir for 1/2 hour without heat.
*/-		11.	Turn on vacuum jet only on main receiver for vacuum check if greater than 10 mm consult supervisor.

BATC	i
DATE	
ACID	NO.

### DISTILLATION - (Cont'd)

INITIALS	DATE		
.65	<del></del>	12.	Vent main receiver.
8.5		13.	Turn on cold water to condenser.
Ga	<del></del>	14.	Open distillate line to small receiver.
<u> </u>	<del></del>	15.	Turn on $N_2$ flow on full, into 2F and adjust vacuum to 15 inches with vent on small receiver.
S.C.		16.	After 1/2 hour hold, start silicone fluid through 2F jacket. Set heater to 150-160°C (300°F on dials).
<u></u> .		17.	Turn on 2F shaft collar coolant pump, and water to its heat exchanger.
<u></u>		16.	Turn on silicone tracer on distillate line.
654 28	<del></del>	19.	Maintain 15 inches vacuum until MeOH is removed (Batch temperature approximately 100°C).
654 R3		<b>2</b> 0.	Pull vacuum slowly until 27 inches, break vacuum by closing vacuum lines and letting N <sub>2</sub> fill to atmospheric pressure.
65m R3		20a.	Turn silicone heaters off, maintain silicone flow.
674 R3		21.	
G) HRB		22.	Pull vacuum on entire system plus 2 receivers @ 15 mm. Use $N_2$ bleed through needle valve on 2F to maintain 15 mm. @ 5 gallon south receiver.
68423		23.	Turn heater on silicone heating fluid to maintain the fluid

BATCH	<del></del>
DATE	<del></del>
ACID NO.	

### DISTILLATION - (Cont'd)

INITIALS DATE		
67m	24.	When vapor temperature @ 135°C, drain double valved distillate
		sample line and switch to main receiver. Maintain 13 1/2 - 15
6511	25.	Turn off silicone tracer line on distillate line.
65n	26.	Once batch temperature is 240°C, start maintaining a difference
		of 40°C between the silicone and batch temperature by increasing
		the silicone oil temperature.
13	27.	Watch for increase in flow (distillation rate) shut off silicone
<u></u>	27.	to 2F if rate becomes too high.
0.		•
13	28.	When batch temperature at 300°C, increase silicone oil tempera-
		ture to 345-355°C.
10	29.	Distill to 325°C batch temperature and 15 mm vacuum in main
		receiver.
SB.	•	
	3ú.	Turn off vacuum to entire system, break vacuum in system with
		N <sub>2</sub> through needle valve on 2F. Turn off silicone heaters and pump.
60		
<i>SP</i> 1	31.	When pot pressure @ 0, open went.
ND	32.	Turn off agitator.
***		
95	33.	Immediately drain (VERY HOT!) still bottoms into a $N_2$ blanket on
		5 gallon car. The can, will be <u>VERY HOT!!</u> Approximately 700°F!
SB	34.	Let 2F cool down for clean-up and recharging.

			BATCH
			DATE
			ACID NO.
			DISTILLATION - (Cont'd)
INITIALS	DATE		
<u> 15 _</u>		<b>3</b> 5.	Turn off cooling water to 2F condenser.
<u>-½</u>		36.	Turn off $N_2$ flow into silicone vent line.
<i>S</i> S		37.	Shut down vacuum jet, first open drain valve on 1st floor manifold, and then shut down.
· 		38.	Drain south 5 gallon receiver and weigh, place liquid into solvent scrap drum, after removing 200 ml sample.
		39.	Drain main receiver into previously cleaned 20 gallon drum and weigh. Mark drum. "Batch Product Reclaim Oil".
		40.	When batch temperature in 2F is at 200°C turn off 2F shaft collar coolant pump, and water to its heat exchanger.
	<del></del>	41.	Turn off N <sub>2</sub> flow into 2F.
		42.	Consult with supervisor for clean-up of 2F.

PATCH [853552 (6-74-4)

DATA SH'ET

RATCH DATE

DATA SHIET

Salacone	40	No. EV	7 e /	242		
	Temp.	Temp.	Main Receiver	Small Receiver	2F	Remarks
U	2.46.0	2,7/1		/6 m m		
+2°C	250	5//		14 mm		
296	252	111		15 mm		
767	255	01/		/4 m m		
299	259	125		15.504		
305	259	751		18.500		
303	760	135		7 2 2		switches Recolveds
304	260	///	15,00	*		
363	262	145	2/	Þ		
**	7	##	‡	80		
305	263	145	11			
305	215	150	/رو			
305	761	121	w w 51			
306	262	207	/smm			
306	263	400	75			
3,5	264	235	14/2			
207	263	142	15			
307	217	642	18			
306	277	242	15%			
309	597	250	15			
309	270	222	15 1/2			
309	271	153	1)			
- + 30 <b>4</b>	27	755	4)			
3.0	112	256	و			
3 1 1	273	156	10.5			

HAT. II DATE

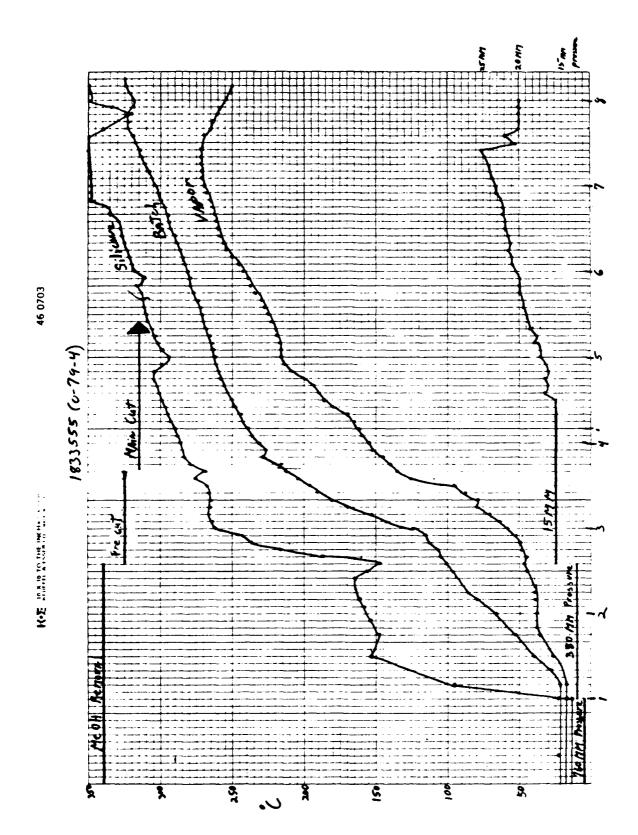
DATA SH'ET

		Remarks																				huber set he 635°t	hut us to 650%				
	Vac. mm	2F																				9	4				
	Vac. men	Small Receiver																									
	Vac. mun	Main Receiver	1500	14.500	· • • • • • • • • • • • • • • • • • • •	18.5	**	*/	<i>ħ/</i>	//	41	5/1	14%	14.5	14.5	۶/	14.5	14.5	//	14.5	*/	*	1/4/	h1	15	7	٠,٠
	Vapor	Temp.	25706	2,196	7.176	282	262	704	265	766	270	175	1/2	212	273	274	277	212	378	219	279	280	180	282	285	286	2.64
	Batch	Temp.	27406	7.566	2760	276	277	278	780	3.60	788	-	587	582	286	187	2.88			295	293	295		300	515	318	3/4
Silicone	Oil Temp.	Outlet	3/11.5	3/1/2	3120	3/5	513	515	3/3	304	321	320	210	319	318		322	325	325	325	325	331	336	543	51	339	355
		Тите	12.20	57:	.30	264	04:	S 7:	. 50	: 55	1300	30:	0/:	-	-			-	-	157:	. 50	.55	100 1	7. 2/	5	>1,6/	- 4

BAT, II DATE

DATA SH: ET

	Remarks														
Vac. mm	2F														
Vac. mm	Small Receiver		Franks												
Vac. mm	Main Receiver	15													
Vapor	Temp.	285													
Batch	Temp.	37	325												
Silicone Oil Temp.	Outlet	357													
		, 12,	$\vdash$												!



ATCH . \$33.555 (U-79-4)	ATCH
ATE 13'8'	ATE _
CID NO.	CID 1

# DISTILLATION

INITIALS	DATE		
		1.	Turn on vacuum jet.
		2.	Close valve, bottom of 2F. Pull vacuum.
		3.	Check to see if filter (50 micron) is in, inline filter on drum suction tube.
		4.	Place drum on scale, suck into 2F 155 lbs of used oil (20 gals).
γ,			oil + container (initial)  oil + container (final)  /39 oil weight
1		5.	Let oil sit over night in 2F with no agitation, open sample valve for vent.
· · · · · ·		6.	Vent vacuum lines.
<u> </u>	<del></del>	7.	Turn off vacuum jet.
<i>F</i> -		8.	Check batch temperature, should be aroun 20°C. Draw 1/2 gallon oil out of bottom of 2F, if no phase separation, return oil to 2F.
RA.		9.	Start agitator on 2F, make sure batch temperature is below 30°C before adding methanol, add gallon of alcoholic NaOH (see supervisor).
an		10.	Let NaOH oil stir for 1/2 hour without heat.
<u>(60)</u>		11.	Turn on vacuum jet only on main receiver for vacuum check if greater than 10 mm consult supervisor.

BATC	·	_
DATE	·	
ACID	NO.	

# DISTILLATION - (Cont'd)

INITIALS	DATE		
<u> </u>		12.	Vent main receiver.
<u> </u>		13.	Turn on cold water to condenser.
		14.	Open distillate line to small receiver.
		15.	Turn on $N_2$ flow on full, into 2F and adjust vacuum to 15 inches with vent on small receiver.
<u>. 75.</u>		16.	After 1/2 hour hold, start silicone fluid through 2F jacket. Set heater to 150-160°C (300°F on dials).
1.5		17.	Turn on 2F shaft collar coolant pump, and water to its heat exchanger.
**************************************		18.	Turn on silicone tracer on distillate line.
.7.	- <del></del>	19.	Maintain 15 inches vacuum until MeOH is removed (Batch temperature approximately 100°C).
:: <u>-</u>			Pull vacuum slowly until 27 inches, break vacuum by closing vacuum lines and letting $N_2$ fill to atmospheric pressure.
<u> 3:</u>		20a. 21.	Turn silicone heaters off, maintain silicone flow.  Drain MeOH from small receiver.
16_		22.	Pull vacuum on entire system plus 2 receivers @ 15 mm. Use $N_2$ bleed through needle valve on 2F to maintain 15 mm @ 5 gallon south receiver.
10		23.	Turn heater on silicone heating fluid to maintain the fluid outlet at 280°C (525°F on dials).

BATCH	
DATE	
ACID NO	·

# DISTILLATION - (Cont'd)

INITIALS	DATE		
<u> </u>		24.	When vapor temperature @ 135°C, drain double valved distillate
			sample line and switch to main receiver. Maintain 13 1/2 - 15
			mm vacuum in main receiver.
5-,		25.	Turn off silicone tracer line on distillate line.
		26.	Once batch temperature is 240°C, start maintaining a difference
			of 40°C between the silicone and batch temperature by increasing
			the silicone oil temperature.
11		27.	Watch for increase in flow (distillation rate) shut off silicone
	<del></del>	2	to 2F if rate becomes too high.
<i>(</i> -3			to 21 12 14ct becomes too 11411.
<u> Gir</u>		28.	When batch temperature at 300°C, increase silicone oil tempera-
			ture to 345-355°C.
		29.	Distill to 325°C batch temperature and 15 mm vacuum in main
			receiver.
		30.	Turn off vacuum to entire system, break vacuum in system with
	<del></del>	٥٠.	N <sub>2</sub> through needle valve on 2F. Turn off silicone heaters and
			pump.
13			
		31.	When pot pressure @ 0, open vent.
00		32.	Turn off agitator.
AB			
* * *		33.	Immediately drain (VERY HOT!) still bottoms into a N; blanket on
			5 gallon can. The can, will be <u>VERY HOT!!</u> Approximately 700°F!
0.0		34.	Let 2F cool down for clean-up and recharging.

BATCH	¹		 
DATE		 	 
ACID	NO.	 	 

# DISTILLATION - (Cont'd)

INITIALS	DATE		
		<b>3</b> 5.	Turn off cooling water to 2F condenser.
<u> </u>		36.	Turn off N <sub>2</sub> flow into milicone vent line.
<u> </u>		37.	Shut down vacuum jet, first open drain valve on 1st floor manifold, and then shut down.
		<b>3</b> 8.	Drain south 5 gallon receiver and weigh, place liquid into solvent scrap drum, after removing 200 ml sample.
		39.	Drain main receiver into previously cleaned 20 gallon drum and weigh. Mark drum. "Batch Product Reclaim Oil".
		<b>4</b> 0.	When batch temperature in 2F is at 200°C turn off 2F shaft collar coolant pump, and water to its heat exchanger.
		41.	Turn off N <sub>2</sub> flow into 2F.
		42.	Consult with supervisor for clean-up of 2F.

PAT 11 18 53555 (0.79.4)
DATE

Carried Management of the Control

DATA SH! ET

	Remarks	cutu/horn orbita to horton	had on sot the 300 pt	Giren Conny of C							Starting to Dine 35 VAC	noor 2.11 cm. > 16	There 35.11 cm. off		time between 17,000 S.A. Com.									through time who	" which some for I no -	Bearing of the
O.S.V.	2F	ATH	1,51	1472		.41	14"	14."	12"	"21		29"								}						
Vac mm	Small Receiver												(5Mn)	His hi	02/51	CH 1/11	15	/>/	15	1/6	15	15,	15%	17	167	
7 P. C.	Main Receiver																									
Varior	Temp.	\	7/	ر(-	A	47	36	۶ <b>و</b>	39	40	÷.	ćh	4.	4.5	62	53	\$ \$	(, 9	1/	28.0	79	57	1.5	136	25	
HAT THE	Temp.	23	2.3	7.7	3.2	oh.	53	67	80	68	47	101	(6.5)	1.7	7.1	21	1:9	143	153	170	131	-2	197	204	ر - د -	
Silicone	Outlet	\	7.5	16		151	841	154	11.9	١٠٢	161	25/	L.K.1	187	217	236	243	712	691		276	Dist	165	27:	9/1	
	Tıme	oril	2000	3, 20	8:10	8.30	4.8	00. 6	9.10	110	4.50	9.3.	2.40	14.6	<del> </del>							1 1 2	3	~		!

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PATA SH! ET

	Silicone	_					
	Oil Temp.	Batch	Vapor	Vac. mm	Vac. mm	Vac. mm	
Time	Outlet	Temp.	Temp.	Main Receiver	Small Receiver	2F	Remarks
10.15	191	7/7	ر طه	16			
1000	462	617	145	/>/			
10.55	2 13	217	153	13.5			
00: //	938	232	155	15.0			
11.05	430	2.3.2	091	165			
11.10	0 6 7	239	79)	10 1/2			
11.15	667	243	170	/3			Sigt phist cleans in arrangh To
11:20	567	144	175	//			
11.34	247	857	180	14/4			
17.72	199	1.00	1.16	13 %			
(1.3)	202	457	190	1, 0)			
11.40	303	957	194	7/			
11.45	305	258	200	16 1/2			
05 1)	305	266	101	9/			
11 55	199	792	214	16 14			N. into be start offer
1.00	441	797	216	17			
1.05	278	263	316	17			
11.10	302	164	C١٠	18			
(1,7	408	797	219	171			turned heaten to sivil
11,10	306	392	642	18%			
11.11	3.09	170	111	18 %1			Town of Books to 5 11 th
35 6	316	1.21	ر ۱۱	19			
	7.7	173	22 8	1:1			
7	1/3	474	730	19,0			
7/	2/4	177	235	01			

BATCH

DATA SHEET

	Silicone						
	OIL TOMP.	Batch	Vapor	Vac. mm	Vac. mm	Vac. mm	
Time	Out let	Temp.	Temp.	Main Receiver	Small Receiver	2F	Remarks
12.50	317	662	237	mum			
75.1	3/1	2 80	139	7.0			Turned boulds to become
13.00	3/8	147	243	2040			
13 05	310	184	246	11			
	277	245	5.5.5	7.1			
12,15	418	637	457	211/2			
13.30	275	397	2,5,5	21/4			
14.21	5.7	296	3 . 7	2.1			
3.0	77.	293	797	11			
1.35	124	7.4.7	197	7			
04.1	33.5	27.	162.	4/12			hunk not to 650°E
+	337	147	197	22/2			
. 50	350	300	266	11 1/2			
5	348	302	397	23/4			
0.0	865	3.7	02.3	23/4			
25	) 4 K	3.6	172	23%			
2 / 2	349	8.24	77.1	74			
	2.5	1/4	172	2 -1/4			
2 ~	5 2 0	1/3	202	27%			
,	. 754	3/2	272	25%			all to two lying out
30		7.4		15%			1 his
		3.19	oir	11			Led. I
	041	222	245	1,7			then for and
		4 2 2	164	, 41			
` Q	325	3.2/	.60	15			though his again

RAT II DATE

DATA SHIET

	Remarks	hut but m														
	Vac. men 2F					F.11,360d										
	Vac. mm Small Receiver					Fia										
	Vac. mm Main Receiver		>/			-(/										
	Vajor Temp.	158	457	157	250											1
	Batch Temp.	1.18	719	321	314	325										
Silicon	Oil Temp. Outlet	334	350	349	350						1					
!	Тіте	34.1	3.00	\$ 65	3.10	3.75						!		 1	1	

### APPENDIX H

## PRODUCT SPECIFICATIONS FOR ADDITIVES AND ADSORBANTS

Dioctyldiphenylamine (DODPA)
Tricresyl Phosphate (TCP)
Quinizarin
Benzotriazole
Ethyl Antioxidant 703
Triphenyl Phosphite
Emery Base Stocks
Phenyl-\alpha Naphthylamine (PANA)
Barium Hydroxide-Monohydrate

Product data sheets were obtained, where possible, for the additives and adsorbents used in the optimized reclamation process. The available data sheets are included here to identify the source and quality of the materials used. The following discussion identifies the source and quality of those materials for which a specification sheet was not available.

Quinizarin was obtained from GAF Corporation. It is identified only as purified 1,4-dihydroxyanthroqinone with no physical properties listed.

Tricresyl phosphate was obtained from FMC Corporation as Kronitex AA.

Triphenyl phosphite was obtained from Eastman Kodak Company. It has a boiling point of 360°C and a density at 25°C of 1.184.

Barium hydroxide monohydrate tech., obtained from Barium & Chemical, Inc., Steubenville, OH 43952.

The filter aid used was Hyflo Super Cel from Johns-Manville Corporation. This material is a Celite diatomite, mainly aluminum silicates, having a specific gravity of 2.30 and the following chemical analysis:  $SiO_2$ , 89.6%;  $Al_2O_3$ , 4.0%;  $Fe_2O_3$ , 1.5%;  $P_2O_5$ , 0.2%;  $TiO_2$ , 0.2%; CaO, 0.5%; MgO, 0.6%,  $Na_2O$  and  $K_2O$ , 3.3%.

PETROLEUM DEPARTMENT

VL81-1

# VANLUBE® 81

#### ASHLESS HIGH-TEMPERATURE ANTIOXIDANT and CORROSION INHIBITOR

#### Typical Properties

Composition:

p,p'-dioctyldiphenylamine

Physical State:

Powder Off-white

Color:

Off-whit

Specific Gravity:

1.01 95C

Melting Point: Distillation Range:

490-500F/0.25-0.75 mm Hg

ńsh:

<0.01%

Heating Loss:

<0.50%

Solubility:

Soluble in silicones, silanes, siloxanes,

diesters, petroleum oils. Insoluble in

water.

VAN! UBE 81 is a specially purified grade of p,p' dioctyldiphenylamine. It was developed commercially after extensive tests showed it to be an effective high-temperature oxidation and corrosion inhibitor in synthetic lubricants based on silane, siloxane, diester, and silicone fluids. In these fluids, VANLUBE 81 at concentrations of 0.5% to 2.0% is an effective oxidation and corrosion inhibitor at temperatures of 400-500F. It is also an effective high-temperature oxidation inhibitor in suitable petroleum base stocks.

In high-temperature lubricating greases, both petroleum and synthetic base, VANLUBE 81 shaws good antioxidant properties in ASTM D-942 oxidation tests and in high-speed spindle tests. Siloxane greases containing 2% VANLUBE 81 have given outstanding results in begring performance tests at 350F.

VANLUBE 81 has good solubility in a variety of synthetic and petroleum base lubricants and is color stable when exposed to light. It can be used as a general-purpose additive in a number of petroleum lubricants which require a stable, ashless, high-temperature oxidation inhibitor.

7410

213



# MATERIAL SAFETY DATA SHEET

APPR PRESSURE THIS PERCENT  APPR DENSITY AIRS EVAPORA  SCLUBLITY IN MATER negligible  APPEARANCE AND DOOR  White powder  SECTION IV FIRE AND EXPLOSION	EMERGENCY TELEPHONE NO 203-853-1400
SECTION III PHYSICAL  D_ P_ P'-Dioctyldiphenylamine  SECTION III PHYSICAL  D_ NG PONT F SPECIFIC  LAPOR PRESSURE THG PERCENT  SCOUNTING FOR STATE SPECIFIC  SPECIFIC SPECIFIC SPECIFIC  SPECIFIC SPECIFIC SPECIFIC  SECTION IV FIRE AND EXPLOSIC	
SECTION III HAZARDOUS IN  MATERIAL  P.P'-Dioctyldiphenylamine  SECTION III PHYSICAL  SPECIFIC  APPOR PRESSURE THIS  SCUBLITY IN MATER  APPEARANCE AND DOOR  White powder  SECTION IV FIRE AND EXPLOSIC	
SECTION II HAZARDOUS IN  MATERIAL  P.P'-Dioctyldiphenylamine  SECTION III PHYSICAL  SPECIFIC  APPOR PRESSURE THIS PERCENT  APPOR DENSITY (AIRI) EVAPORA  SCLUB LITY IN MATER  APPEARANCE AND DOOR  White powder  SECTION IV FIRE AND EXPLOSIC	
SECTION II HAZARDOUS IN  MATERIAL  P. P Dioctyldiphenylamine  SECTION III PHYSICAL  SPECIFIC  SPECIFIC  APOR PRESSURE Hg  WAPOR DENSITY AIRIL  SCOUNT IN MATER  REPEARANCE AND DOR  White powder  SECTION IV FIRE AND EXPLOSIC  FLASH POINT I METHOD USED	
SECTION III PHYSICAL  SPECIFIC  SPECIFIC  APPOR PRESSURE THIS  CAPPOR DENSITY ARRIT.  SCLUBLITY IN MATER  APPEARANCE AND DOOR  White powder  SECTION IV FIRE AND EXPLOSIC	
SECTION III PHYSICAL  SPECIFIC  SPECIFIC  APPOR PRESSURE THIS  CAPPOR DENSITY ARRIT.  SCLUBLITY IN MATER  APPEARANCE AND DOOR  White powder  SECTION IV FIRE AND EXPLOSIC	CREDIENTS
SECTION III PHYSICAL  BOUNG POINT OF SPECIFIC  LABOR PRESSURE THIS PERCENT BY VOLUM  SCUID OVER INTEREST POINT OF SECTION IV FIRE AND EXPLOSION  SECTION IV FIRE AND EXPLOSION  FLASH POINT OMETHOD USED	TLV UN'S
SECTION III PHYSICAL  SPECIFIC  SPECIFIC  APPENDENTIAL  SPECIFIC  PERCENT  BY VOLUS  SPECIFIC  PERCENT  RY VOLUS  SPECIFIC  PERCENT  RY VOLUS  SPECIFIC  PERCENT  RY VOLUS  SPECIFIC  SPECIFIC  SPECIFIC  PERCENT  SPECIFIC  SPECI	TLV UNIS
SPECIFIC  APOR PRESSURE THE PERCENT  APOR DENSITY AIRIS  SCLUBLITY IN MATER  APPEARANCE AND DOOR  White powder  SECTION IV FIRE AND EXPLOSIC  FLASH POINT I METHOD USED	Acute oral LD
SPECIFIC  APOR PRESSURE THE PERCENT  APOR DENSITY AIRIS  SCLUBLITY IN MATER  APPEARANCE AND DOOR  White powder  SECTION IV FIRE AND EXPLOSIC  FLASH POINT I METHOD USED	>5,000 mg kg rats
SPECIFIC  APOR PRESSURE THE PERCENT  APOR DENSITY AIRIS  SCLUBLITY IN MATER  APPEARANCE AND DOOR  White powder  SECTION IV FIRE AND EXPLOSIC  FLASH POINT I METHOD USED	
SPECIFIC  APOR PRESSURE THE PERCENT  APOR DENSITY AIRIS  SCLUBLITY IN MATER  APPEARANCE AND DOOR  White powder  SECTION IV FIRE AND EXPLOSIC  FLASH POINT I METHOD USED	
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SPECIFIC  APOR PRESSURE THE PERCENT  APOR DENSITY AIRIS  SCLUBLITY IN MATER  APPEARANCE AND DOOR  White powder  SECTION IV FIRE AND EXPLOSIC  FLASH POINT I METHOD USED	DATA
PERCENT PERCEN	GRAVITY : H,Oat
SECTION IV FIRE AND EXPLOSIC	· · · · · · · · · · · · · · · · · · ·
SCLUB LITY IN MATER negligible  APPEARANCE AND DOOR White powder  SECTION IV FIRE AND EXPLOSION FLASH POINT I METHOD USED	VOLATILE
negligible  white powder  SECTION IV FIRE AND EXPLOSIC	TION RATE
SECTION IV FIRE AND EXPLOSIC	Density Mg/m <sup>3</sup>
FLASH POINT I METHOD USED FLAM	Density Mg/m 1.01
FLASH POINT I METHOD USED FLAM	N HAZARD DATA
EXT NOU SHING MED A	MABLE LIM TS
CO3. foam, dry chemical	
UNUSUAL FIRE AND EXPLOSION HAZARDS	

information presented herein has been combiled from sources considered to be dependable and is accurate and relable to the best into unknowledge and be ethalf is not guaranteed to be so. Nothing herein is to be construed as recommending any prescree or any ordination or any palable to the ethalf of any award requal or it is the use is responsiblely to determine for himself resultant in a part of actions of a social so it safety makes and as one of a social soc



703

# "ETHYL" ANTIOXIDANT 703

2. 6-Di-tert butyl-or-dimethylamino-p-cresol

#### TYPE

with the second temperature stable, phenotic antioxidant.

#### TYPICAL PROPERTIES

Form Color Molecular weight Crystalline powder Light yellow 263.4 Melting point Boiling point Flash point (COC) 94°C 201°F 179°C 354°F at 40 mm >200°F

#### **APPLICATIONS**

Ox day or a hibbor in ratural and symmetric rubbers, polyolefin plastics, resins, adhesives, petroleum oils and waxes

SQLUBILITY (Wt. C) at 20°C.

Toluene Ethyl Alcohol 22 V 28 I

Water 10℃ NaOH < 0.0007 < 0.002

#### TOXICITY

The acute ora, I.D. for "Ethyl" Antioxidant 703 is 1030 mg kg of body weight.

ETHYL CORPORATION THAT

Ordering

and

Shipping

Information

Refer to product as:

# "ETHYL" Antioxidant 703

#### Container sizes:

100 B) net (24 gal) nobjecturnable drom  $(25) {\rm He}({\rm ret}) = 8 \, g_{\rm pl}$  , non-reformable dram 5 in net 2 gal non-returnable dram-

#### Weight Information

Drum Size	100 lb	25 lb	5 16
Net contents, gal (nominal)	24	8	2
Net contents lb	100	25	5
Tare 15 approx	9	5	15
Diniensions d'ameter in	16	115	8 5
Dintensions height in	30 75	21 25	9 75
Volume cuft	4 59	1 63	C 42

#### Rail Shipments:

Minimum carload lots 025 drums 24 gallon size

#### Truck Shipments:

Minimum trioblead with 271 drums 24 gradus ze-

#### Mail or wire orders:

Ethyl Corporate n

Industrial Chemicals Division Bateri Reage La 70821

#### Telephone orders

Baten Rouge, La 504-344-5222

#### Shipping point:

Orangeburg South Carolina

#### IN CANADA

Orders should be placed with Ethyl Corporation Limited 48 St. Clair Avenue West. Toronto.

The information presented here is be reverted be accorded and relating to exerted withough advance of ending to the organism of the control o

ETHYL CORPORATION / ETHYL TOASE 451 FLOR CA INDUSTRIAL CHEMICALS DIVISION / BATCH ROLDE LA TOBUT

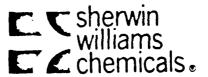


CHICAGO + CLEVELAND + HOUSTON + LOS ANGELES + NEW YORK + TULSA

# **BENZOTRIAZOLE - Photo Grade**

Code: BT-PG Order Entry No.: X19 HI 5589

Synonym: 1,2,3-Benzotriazole



technical bulletin 142

NH NH

 $C_{\varepsilon}H_{S}N_{3}$ 

M.W. 119.12

# PROFERTIES

Appearance

White crystalline needles.

	Specifications*	Typical Analysis
Assay	98.0% min	99.5
Appearance of 1% Solution	Essentially clear	Essentially clear
Pesidue After Ignition	0.5° max.	0.08
Volatile Matter at 70°C	0.5% max.	0.08
Identity Melting Point	<b>9</b> 8-99°C	<b>9</b> 8-99°C

\*Meets requirements of American National Standard Institute specification PH4.204-1972. Methods of analysis are given in the specification.

USES

Photographic chemical—developer, anti-fogging agent and restrainer in gelatin emulsions. Also in other areas where light sensitive substances are useful. (See References).

TOXICITY

The oral LD $_{10}$  in white rats is 560 mg per kg. Chemicals with values of this order are generally considered moderately toxic.

Tests made on intact and abraded skin of rabbits showed no skin

irritation.

SAFE HANDLING

Benzotriazole and its solutions are not serious industrial mazards provided that workers are properly instructed and adequately supervised in handling procedures. Where adequate ventilation is not available, approved respiratory and eye protection are required

in dust laden areas.

**AVAILABILITY** 

Readily available from stock. Package sizes, prices and other details of sale are stated in our latest Chemicals Price List.

# REFERENCES

Battaglia, Photogr. Sci. Eng. 1970, 14 (4), 275; CA 73, 50708.

Brit. 1,173,426 (1969); CA 73, 40477.

Fr. 1,542,505 (1968); CA 71, 66055.

Ger. Offen. 2,028,214 (1971); (To Western Electric); CA 74, 118423.

Sahyun, Photogr. Sci. Eng. 1970, 14(3), 192; CA 73, 20423. Ibid, 1971 15(1), 48; CA 74, 48046.

Sheberstov and Borokova, Zh. Nauch. Prikl. Fotogr. Kinematogr., 1969, 14(4), 292; CA 71, 96868.

U.S. 3,547,635 (1970) (To Eastman Kodak); CA 74, 149194.

Zyuskin and Braichevskaya, Zh. Nauch. Prikl. Fotogr. Kinematogr. 1968, 13(5), 326; CA 70, 16012.

Zyuskin and Glinskaya, Zh. Nauch. Prikl. Fotogr. Kinematogr. 1969, 14(6), 470 (Russ); CA 72, 61365.



P.O.Box 6520, Cleveland, O. 44101

Space partners and the property of the control of t

TABLE I
SPECIFICATIONS OF EMERY LUBRICANT BASES

Specification	Emery 2910	Emery 2911	Emery 2957	Emery 2958	Emery 2960	•	Emery 2932
Acid Value, Max.	0.2	0.2	0.2	0.2	0.02	0.08	0.10
Hydroxyl Value, Max.	3.0	2.0	2.0	2.0	3.0	3.0	3.0
lodine Value, Max		2.0	2.0	2.0	1.0		• •
Flash Point, °F, Min.	400	300	400	400	430	<b>4</b> 80	460
Fire Point, *F, Min.		350	450	450		• •	• -
Cloud Point, *F, Max.	<b>-4</b> 0	<del>-4</del> 0	<b>-4</b> 0	<b>-4</b> 0	<b>-4</b> 0		<del>-4</del> 0
Pour Point, *F, Max.	-70	<del>-</del> 90	-50	<del>-</del> 75	-75	<b>-6</b> 5	<b>-</b> 75
Viscosity, Cs.							
210°F, Min.	2.60	1.60	3.20	2.85	4.3	4.65-4.85	4.0
100°F, Max.			• •	• •		22.0-24.0	
-40°F, Max.			1800				5000
-65°F, Max.	6500	800		7000	• •		
*SOD Lead Corrosion, Mg/sq. in.							
1 Hour, Max.	<u>+</u> 4	••	<u>+</u> 4	<u>±</u> 4	<u>+</u> 4	No negative** values	No negative** values
*Oxidation Stability, 347°F · Hours, Min.	100		100	100		••	

<sup>\*</sup>Contains 0.5% purified phenothiazine 0.1% Ethyl anti-oxidant 703

<sup>\*\*</sup>Contains 0.5% purified phenothiazine

#### UNIROYAL CHEMICAL

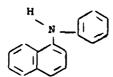
Division of UNIROYAL Inc Naugatuck Connecticut 06770

# Naugatuck Chemicals



#### PRELIMINARY DATA SHEET

NAUGARD® PANA for Sym. 14.7. - M



Phenyl-Alpha-Naphthylamine

Naugard P.W. is an aromatic amine antioxidant for synthetic lubricants, lubricating greases, industrial oils, and railroad diesel engine oils. It offers excellent antioxidant activity at an economical cost. Naugard PANA is particularly useful in high temperature applications. It provides effective protection in synthetic lubricants at temperatures that at times reach 600-700°F (315-371°C)

Naugard PANA is also an important antioxidant in soap thickened greases as well as those made with non-soap thickeners. In addition to its many applications in petroleum, Naugard PANALIS used as a rubber antioxidant. It is the primary antioxidant for neoprene rubber. Recommended concentrations for this product vary from 0.2-1.0% depending on the application.

#### Typical Properties

N-phenyl-1-naphthylamine Chemical Name

Physical Form Tan to purple crushed solid

Specific Gravity ⊌ 25°C

1.23

131°F (55°C) Melting Point

0.1% Ash

Flash Point

417°F (214°C) (Penske Marten)

Soluble in acetone, alcohol, benzene, carbon Solubility tetrachloride and chloroform. Insoluble in water.

The recommendations for the use of our products are hased on tests believed is beine' able. However we do not guarantee the results to be obtained by others under different and the results to himper this brochure is intended as a recommendation to use our products so as to infringe on any patent



### Storage Stability

Naugard PA... has excellent storage stability. Long exposure to air and light will cause the product to darken but will notaffect antioxidant activity.

## Toxicity and Handling Precautions

PANA has been used for many years without any known serious health problems. As it has a reported oral LD<sub>50</sub> (rats) of 1625 mg/kg it should be treated as a "toxic" substance. Human experience has indicated some skin irritation potential.

Trace amounts of alpha-naphthylamine (400-800 ppm) and beta-naphyl amine (0-80 ppm) have been detected in Naugard PANA. These substances have been designated as carcinogens by OSHA and when present at minimum levels of 10,000 ppm for alpha-naphthylamine and 1,000 ppm for beta-naphthylamine are subject to OSHA Regulations Numbers 1910.1004 and 1910.1009.

In the handling of PANA we advise taking appropriate precautions to avoid ingestion, contact with the skin, and breathing of dust or vapors.

## UNIROYAL CHEMICAL

Division of UNIRCYAL Inc. Neugatura: Connecticus 06770 Naugatuck<sup>®</sup> Chemicals



### TENTATIVE SALES SPECIFICATION

### NAUGARD PANA

(Phenyl Alpha Naphthylamine)

## FOR Petroleum & Synthetic Turbine Oils

TEST	MINIMUM	MAXIMUM	TEST METHOD
Assay, %	99	-	GLC-31QC
Aslı, 🗴	-	0.05	G-68-G
Toluene Ins., %	-	0.03	G-68-G
Moisture, %	-	0.1	G-24
Appearance		to purple d cast in d	crushed solid rums.

JAS/GJH:1mh Revised 7-28-77

The recommendations for the use of our products are based on tests believed to be reliable. However, we do not guarantee the results to be obtained by others under different conditions. Nothing in this brochure is intended as a recommendation to use our products so as to infringe on any patent.



# APPENDIX I

# PROCESS HISTORY FOR EACH USE OIL LOT

0-79-1 - After distillation treated with Ca(OH)<sub>2</sub>, oil developed a foam test volume over  $300^+$  mL. Terminated further work up.

in the fight and the fight area and a fight at the same accompanion and

0-79-2 - Distilled one sample without NaOH as a control, had a fairly high foam test volume.

Distilled a 2nd sample with NaOH which had a very high foam test volume. We redistilled the sample and the sample still had a high foam test volume, so terminated further work up. A expanded GC of the oil does not indicate similarity to 0-82- used oils which also had foaming problems after distillation.

- 0-79-3 After distillation, treated with Ca(OH)<sub>2</sub>, oil developed a high test volume. Terminated further processing.
- 0-79-4 Oil had left thick charred residue in still. Treat oil with  $Ca(OH)_2$ , oil developed a high test volume. Terminated further processing.
- 0-79-5 Oil had a high acid number, assumed that a residue similar to 0-79-4 would develop upon distillation, so did not attempt distillation.
- 0-79-6 The distillate was treated with Ba(OH)<sub>2</sub>·H<sub>2</sub>O overnight at 43°C, then treated with attapulgus clay. The sample was reformulated and sent for MIL-L-7808H testing.
- 0-79-7 The distillate was treated with Ba(OH)<sub>2</sub>·H<sub>2</sub>O overnight at 43°C (treatment temperature was to be 50°C). Precipitate (ppt) developed in oil upon sitting for a month, refiltered. Nine months after filtering, additional ppt in oil. Refiltered the oil, then heat treated to 60°C overnight and then reformulated. The sample was then MIL-L-7808H tested.

- 0-79-8 The distillate was treated with Ca(OH)<sub>2</sub>, oil developed a high foam test volume. Redistilled the distillate and treated with Ba(OH)<sub>2</sub>·H<sub>2</sub>O @ 50°C overnight. The oil was then reformulated and sent out for MIL-L-7808H testing.
- 0-79-9 The distillate was treated with Ba(OH)<sub>2</sub>·H<sub>2</sub>O overnight at 43°C. The sample was formulated and sent for MIL-L-7808H testing.
- 0-79-10 The distillate was treated with Ba(OH)<sub>2</sub>·H<sub>2</sub>O overnight at 45°C. The sample contained a small amount of ppt, which was filtered out by final filtration after reformulation. Sample sent out for MIL-L-7808H testing.
- 0-79-11 The distillate was treated with Ba(OH)<sub>2</sub>·H<sub>2</sub>O overnight at 45°C. The sample was reformulated and sent for MIL-L-7808H testing.
- 0-79-12 The distillate was treated with Ca(OH)<sub>2</sub>, oil developed a high foam test volume. Redistilled the distillate and treated with Ba(OH)<sub>2</sub>·H<sub>2</sub>O @ 45-50°C overnight. The oil was then reformulated and sent out for MIL-L-7808H testing. The sample had previously contained small amount of ppt, which was filtered out by final filtration after reformulation.
- 0-79-13 The distillate was treated with Ba(OH)<sub>2</sub> overnight at 45°C, then treated with attapulgus clay. The sample was reformulated and sent for MIL-L-7808H testing.
- 0-79-14 The distillate was treated with Ca(OH)<sub>2</sub>, oil developed a high foam test volume. Redistilled the distillate and treated with Ba(OH)<sub>2</sub>·H<sub>2</sub>O @ 42°C (treatment temperature was to be 50°C). Precipitate had developed in oil upon sitting for a month, refiltered. Nine months after the refiltering, additional ppt in oil. Refiltered the oil, then heat treated to 60°C overnight and then reformulated. The sample was then MIL-L-7808H tested.

- 0-79-15 The distillate was treated with Ca(OH)<sub>2</sub>, oil developed a high foam volume. Redistilled the distillate and treated with Ba(OH)<sub>2</sub>·H<sub>2</sub>O overnight at 50°C. The sample was reformulated and sent for MIL-L-7808H testing.
- $\underline{0-82-4}$  The distillate failed the foam test, further processing terminated.
- 0-82-5 The oil had a strong kerosene small, stronger then 0-82-4, did not process this sample.
- <u>0-82-6</u> Dis llate had a high foam test volume, redistillation of the distillate did not improve the test results.
- <u>0-82-7</u> Distillate failed the foam test, further processing terminated.
- 0-82-8 The oil had a strong kerosene small, stronger then 0-82-4, did not process this sample.
- <u>0-82-9</u> Distillate failed the foam test, further processing terminated.
- $\underline{0-82-10}$  Lab distillation for foam testing of distillate. Did pass foam test, but the foam volume was higher than would like for further processing.
- $\underline{0-82-11}$  Lab distillation for foam testing of distillate, did not pass foam test.
- $\underline{0-82-12}$  Lab distillation for foam testing of distillate, did not pass foam test.

DISTILLATION DATA FOR EACH RECEIVED USED OIL TABLE 1-1.

「日本の方式を対する」というとなるとなる。
「日本の方式を対する」というとは、

Distillation time hour	7. 4	0.5 4	\$2.5	8 5	6//6	9		7.5	7.5	0 80	5 5	7.5	8.5	0 8	£ 80	7.0	5.75	7.0	6 75	0 8	6.75	9.5	•	0.6	0 6	⊃ <b>6</b>		ۍ.	0 <b>1</b>	2.0	2.0
Density	0.93	0.93	0.93	0 93	0 94	6 93	0.93	0.92	0 92	0.92	0.92	0.93	0.92	0.93	₹6 O	0.92			0.92										1	ı	1
Distillate foam test ml	20	75.	300	245	7.5	960	•	01	10	20	10	2	10	01	15	15	01	15	15	10	10	150	,	300	300	300		300	75	175	300
NaOH used in distillation	Ye.,	No	Yes	Yes	Yes	Yes		Yes	Yes	Yes	Yes	Yes	Yes	Yes	Yes	Yes	Yes	Yes	Yes	Yes	Yes	Yes	,	Yes	Yes	NC NC	•	Yes	No	S.	No
Main cut collections V1 to B1	135 325	155-325	115-325	135-325	135-325	135, 325		13%-325	135-325	135-325	135-375	135-325	135-325	135-325	135-325	135-325	135-325	135-325	135-325	135-325	135-325	200-325"	•	135-325,	200-325,	200-325 <sup>h</sup>		200-325 <sup>h</sup>	200-375	200-375	200 375
Final acid #	1 1/	1 5.9	0 8 3	0.37	1.16	6.24		1.03	0.85	0.78	0 52	0.70	1.36	0.73	10.1	0.50	26 0	68 0	0.44	0.75	0.38	1.68	,	2 5	16.0	8.1	1	1 5.5		,	
Starting acid #	0.48	0.45	0.45	0.83	0.63	15.52	23.8	3.14	3.69	2.17	00.00	£6.0	1.30	1.36	0.42	0.22	0.74	2.51	00.00	3.0	0.0	0.21	0.21	0.17	2.3	•	0.18	0 68	•	1	
Loss	,	6 1	1.5			6.1	•	6.0	4 1	3.5	9 0	4.1	5.1	1 2	4.1	•	5.5	3.9	5.6	5.8	1 8			0.9		3.0	٠	•	6.0	3.2	0 8
Still bottons,	0 6	7.3	6.4	5.5	10 O	23 4	1	5 11	11.3	10.8	2 0	3.5	5.1	<b>9</b> .9	5.8	2.6	1.1	8.7	4.4	7.7	5 6	32 0	•	36 0	19.0	27.0	,	5.81	ر. ت	'n	26.0
Main cut, 8 (recovery)	90.2	ሩ 68	90 3	91.5	18.8	5.59	•	82.5	82 0	833	92.5	0 16	68.5	0.06	88.5	96.1	88.5	67.9	89.2	84.5	93.0	0.19		53 0	0.89	54.0	•	80	74 0	5 98 5 5	32 0
Stall toppings/ pre-cut.	1.0	1.3	1.8	3.0	99	5.0	,	5.1	5.6	2.4	6.1	1.6	1 3	2.0	1.6	1.6	6.1	4.5	3 8	2.0	5 6	12.0		0.3	13 0	16.0		22 0	15.0	<b>~</b> .s	39-6
Used ort	0-79-01	70-66-0	0.79-02	0-79-02	0-79-03	0-79-04	0-79-05	90-66-0	0-79-07	0-79-08	0-79-08	60-62-0	0-79-10°	0-79-11	0-79-12	0-79-17";	0-79-13	0-79-14	0-79-14	0-79-15 d f	0-79-15	0-82-04,	0-85-05	0-85-06	90-28-0	0-82-07	0-85-0c	60-78-0	0-82-10	0-87-11	(i -82-12)
Ţ	18335.0	18335.73	1800/6	1833,54	1833534	18 (1.155	•	18. ,60	1813543				1997633	1533546	1633548	1997636	1997632	1833550	1497644	1833557		2276303		200029-1A	2000/2958	20002	:	70.707.66	2011.519	2, 16,518	7.763.0

 $<sup>^{</sup>m d}$ ann cut consists of distillates collected from Vapor temperature  $^{
m 0}$ C to batch temperature  $^{
m 0}$ C.

bar processed

Wednetilled distillate from above. Constitled only in lab

insofficient Notional

fourth of the control process of the BILL's Masker to china

# APPENDIX J

IR, GC, AND HPLC CHROMATOGRAMS OF USED OIL LOTS RECEIVED LATER IN PROGRAM TO DEVELOP ADDITIONAL SCREENING METHODS

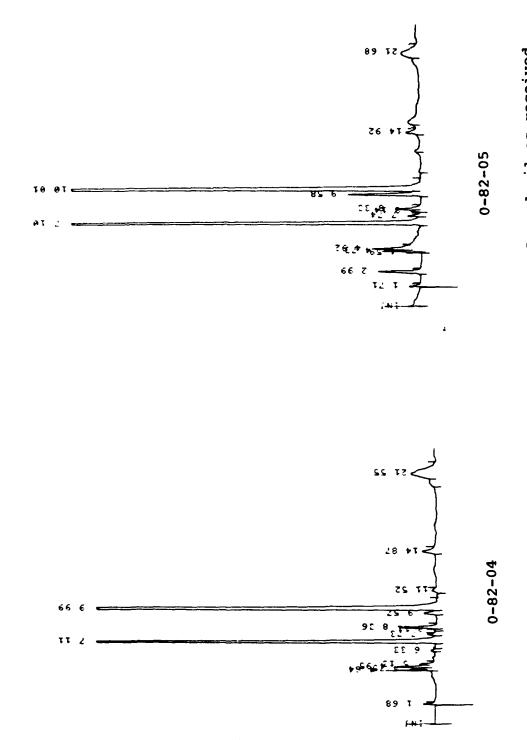


Figure J-1. High performance liquid chromatograms of used oil as received.

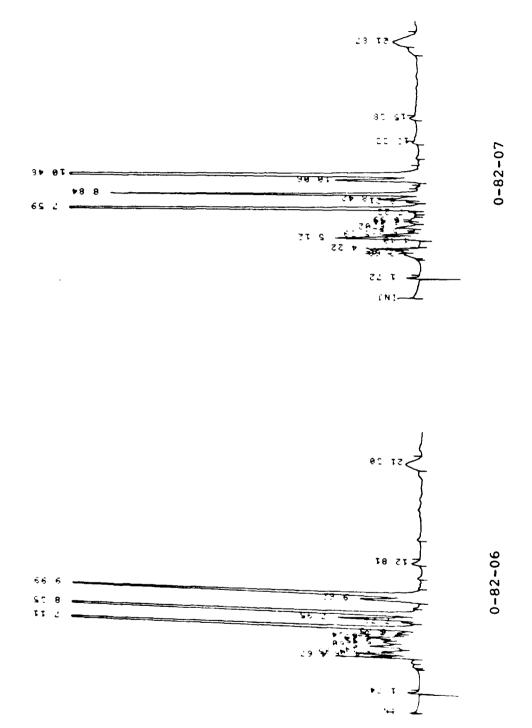
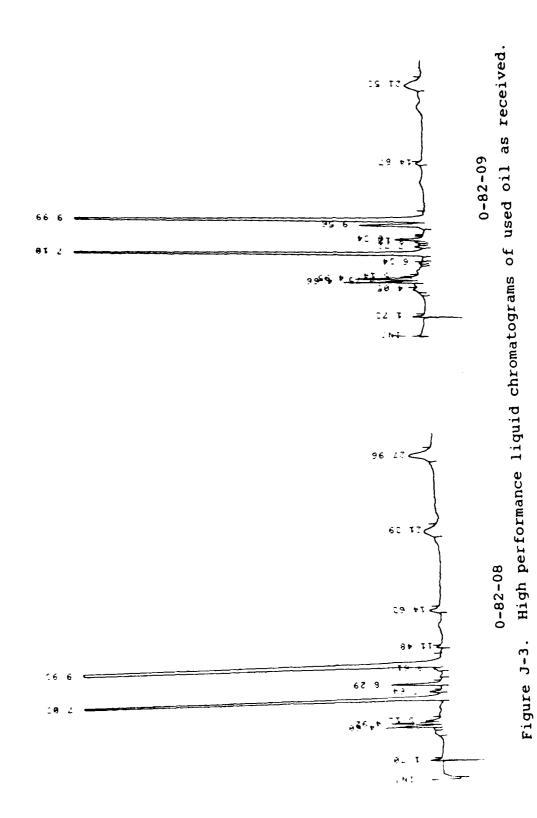


Figure J-2. High performance liquid chromatograms of used oil as received.



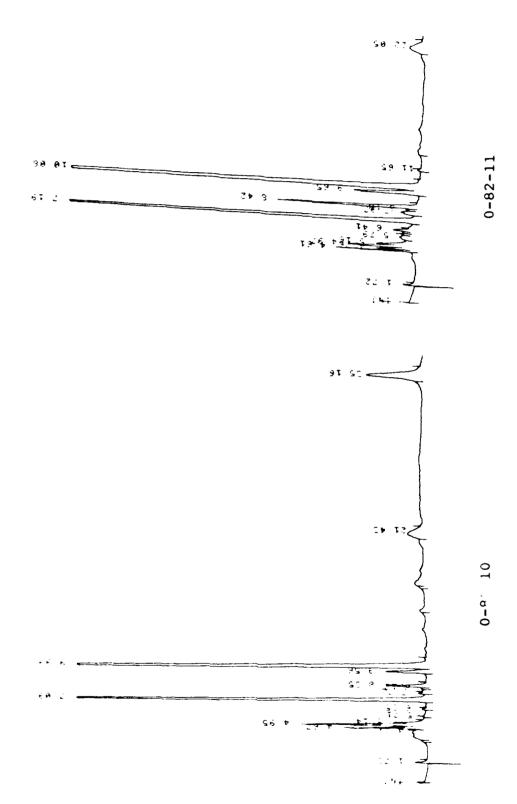
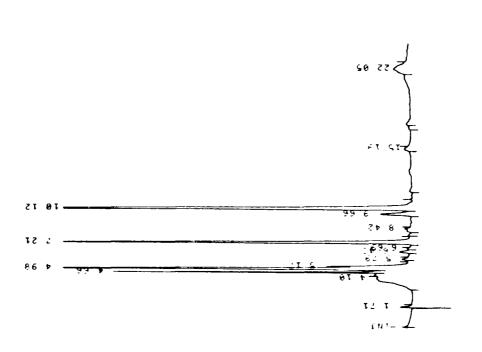


Figure J-4. High performance liquid chromatograms of used oil as received.



The second of th

High performance liquid chromatograms of used oil as received. Figure J-5.

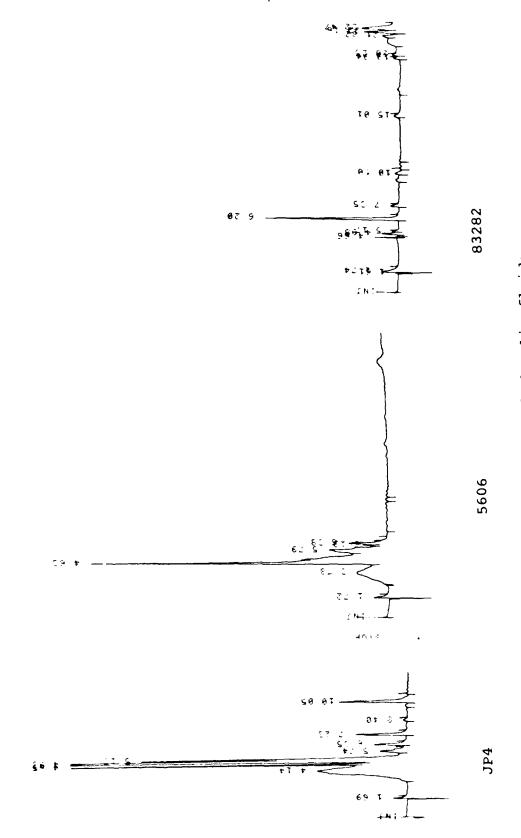


Figure J-6. HPLC of JP4 and 2 hydraulic fluids.

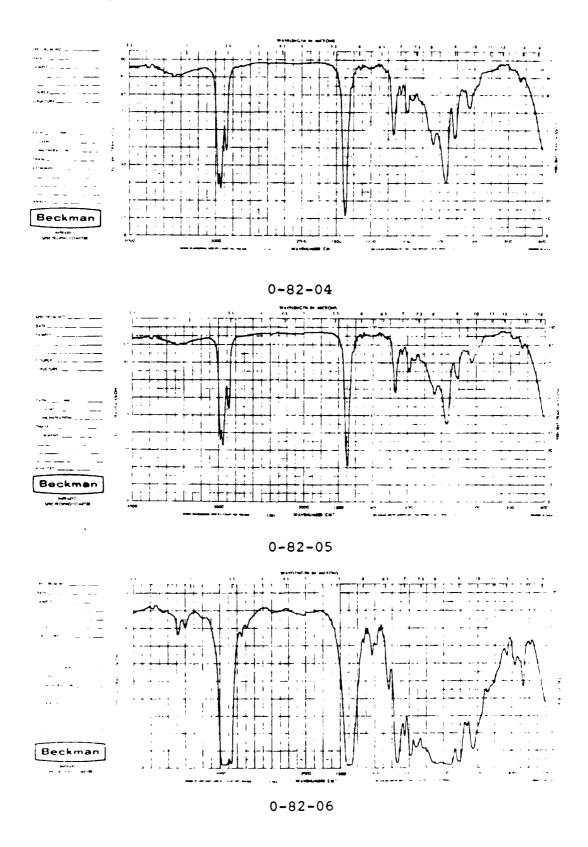


Figure J-7. Infrared spectrum of used oils.

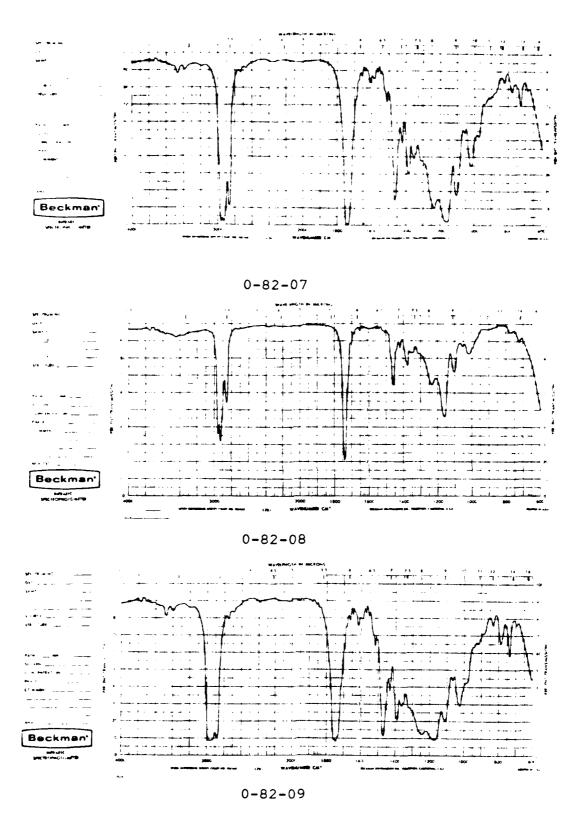


Figure J-8. Infrared spectrum of used oils.

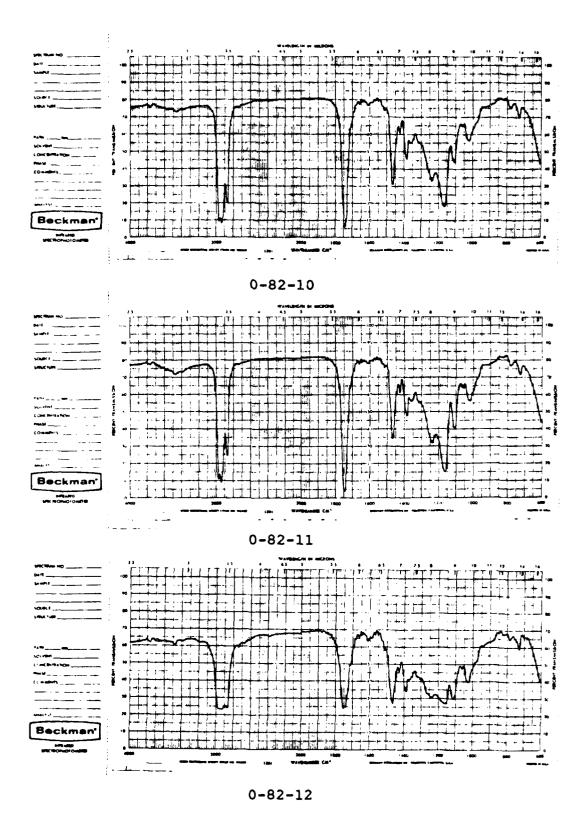


Figure J-9. Infrared spectrum of used oils.

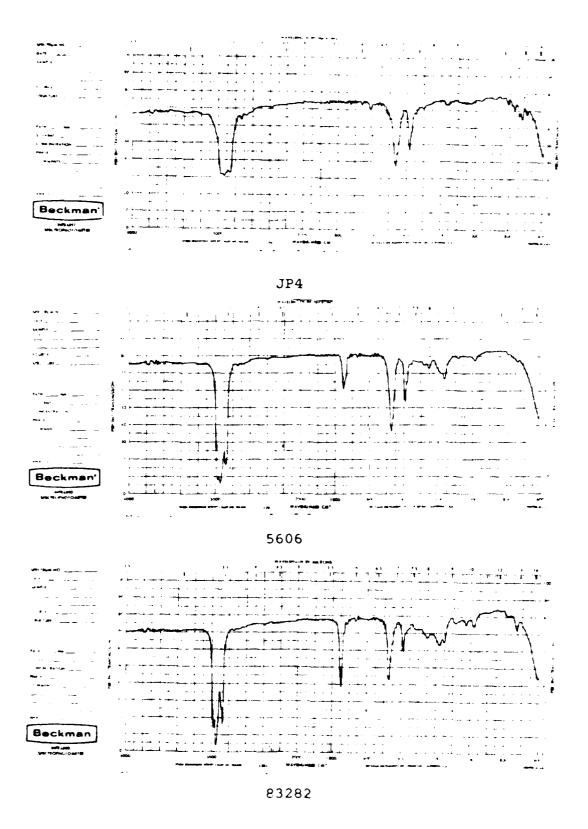
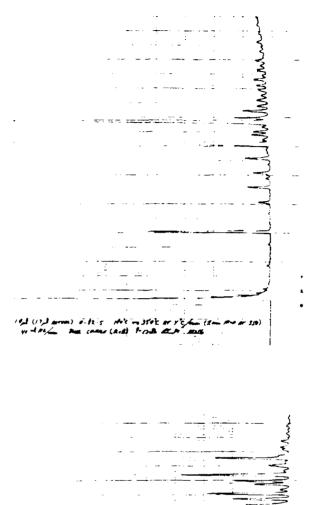
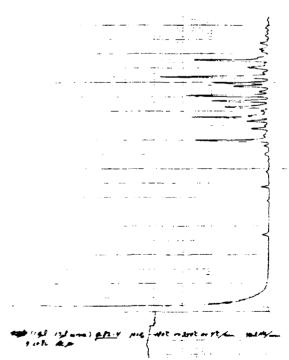
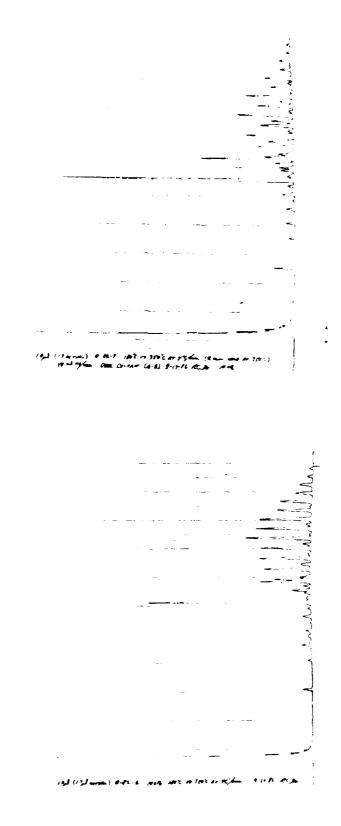


Figure J-10. Infrared spectrum of JP4 and 2 hydraulic fluids.

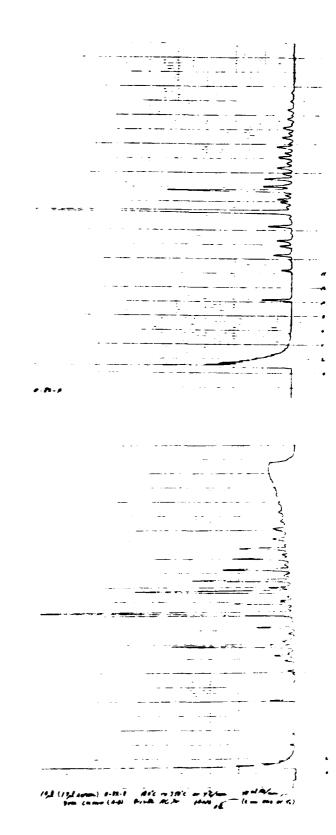




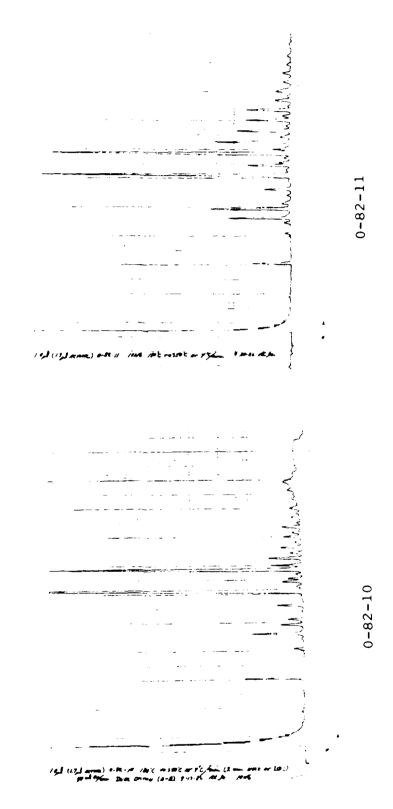
Gas chromatograms of used oils as received; GC conditions 180-350°C. Figure J-11.



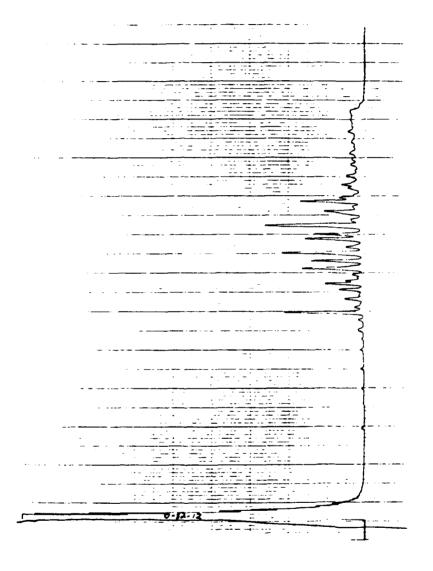
Gas chromatograms of used oils as received; GC conditions 180-350°C. Figure J-12.



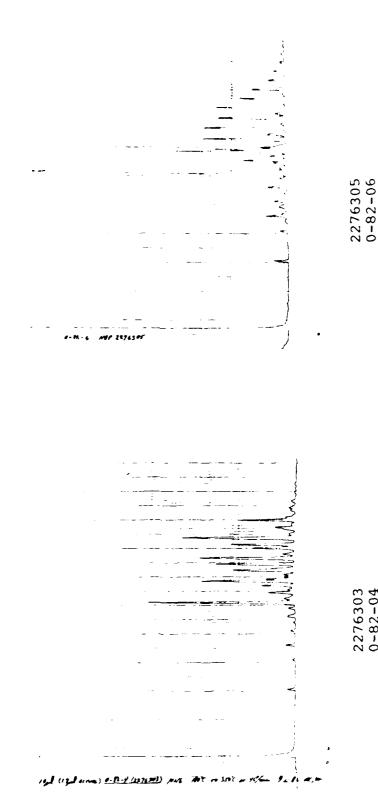
Gas chromatograms of used oils as received; GC conditions 180-350°C.



Gas chromatograms of used oils as received; GC conditions  $180 \text{--} 350 \, ^{\circ}\text{C}.$ 

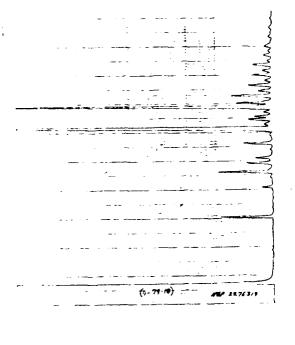


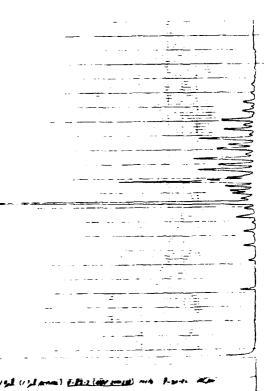
Gas chromatograms of used oils as received; GC conditions 180-350°C. Figure J-15.



Gas chromatograms of distillates, of selected used oils; GC condition  $180-350\,^{\circ}\mathrm{C}$ . Figure J-16.

2276303 0-82-04

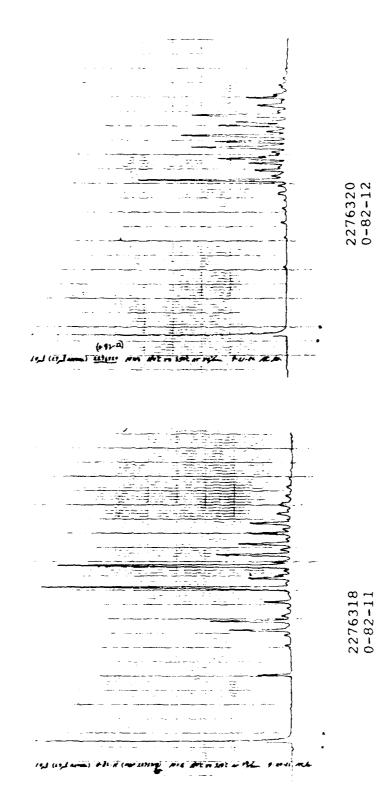




Gas chromatograms of distillates, of selected used oils; GC condition 180-350°C. Figure J-17.

2276319 0-79-10

> 2000298 0-82-07



Gas chromatograms of distillates, of selected used oils; GC condition  $180\text{--}350\,^{\circ}\text{C}\,.$ Figure J-18.

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2276303 (0-82-04) distillate 100-200°C @ 4°C/min 210-350°C @ 32°/min received 100-210°C @ 4°/min 0-82-04

Gas chromatograms of various samples of used oils. Figure J-19.

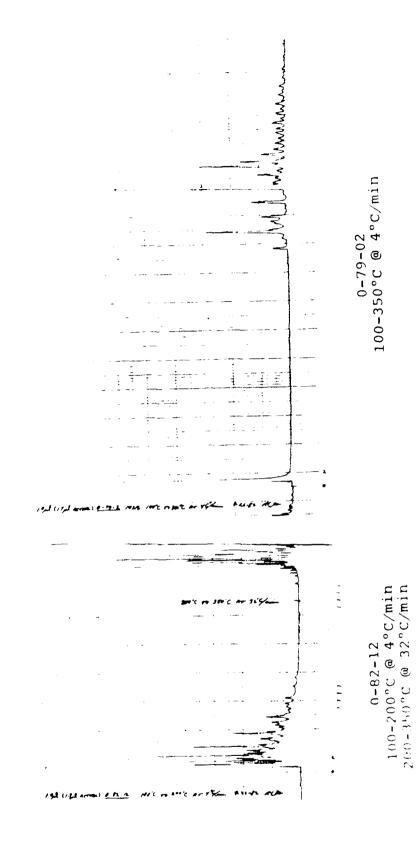
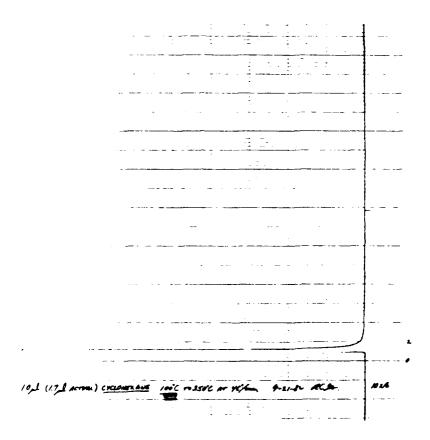


Figure J-20. Gas chromatogram of used oils.



Cyclohexane 100-350°C @ 4°C/min

Figure J-21. Gas chromatogram of cyclohexane.

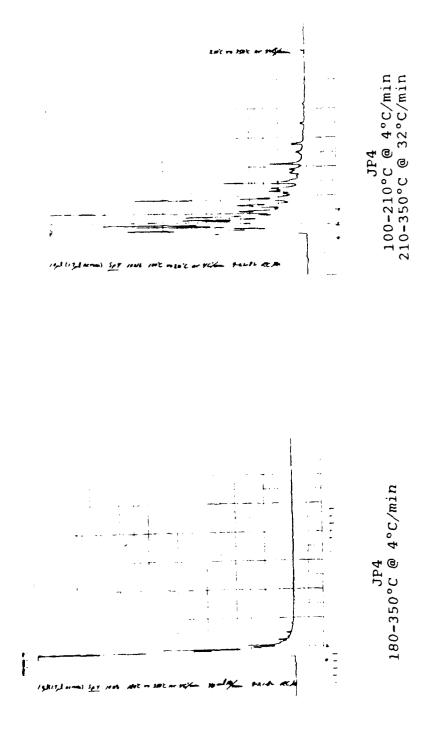


Figure J-22. Gas chromatogram of jet fuel JP4.

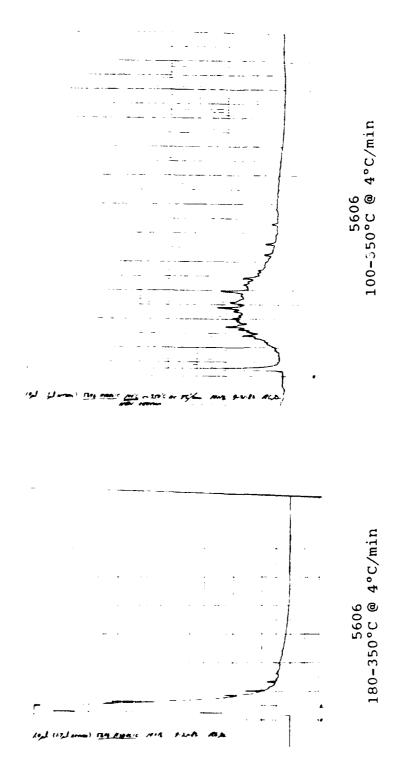


Figure J-23. Gas chromatogram of hydraulic fluid 5606.

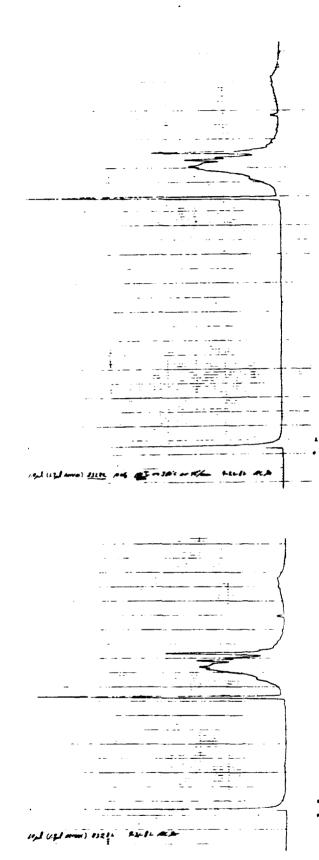


Figure J-24. Gas chromatogram of hydraulic fluid 83282.

83282 100-350°C @ 4°C/min

83282 180-350°C @ 4°C/min

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